

DATE : 11-02-2018

HINTS & SOLUTIONS

PART-A (PHYSICS)

1. Light of.....
Sol. Because below threshold frequency there is no photoelectron emitted. Hence no photo current.
2. When photons.....
Sol. Self explanatory.
3. The maximum.....
Sol. $K.E_{\max} = \frac{hc}{\lambda} - \phi$
 $\frac{1}{2}mv^2 = \frac{hc}{\lambda} - \phi$
 $v = \left(\frac{2hc - \lambda\phi}{m\lambda} \right)^{1/2}$
6. A photon
Sol. Frequency of light does not change with medium.
7. An X-ray.....
Sol. Some of the energy of photon will be absorbed by the electron. Hence, energy of the photon will reduce correspondingly wavelength will increase and frequency decreases.
8. Yellow light.....
Sol. stopping potential (V_s) = 0.25 V
 $eV_s = hv + W$
 Where W is work function W
 $\Rightarrow W = \frac{hc}{\lambda} - eV_s$
 $= \frac{12400(A^\circ eV)}{5770A^\circ} - 0.25eV$
 $= 2.75 - 0.25$
 $W = 1.90 eV$
 $W = \frac{hc}{\lambda_s}$
 λ_s - threshold wave length
 $\Rightarrow \lambda_s = hc / w$
 $= \frac{12400A^\circ eV}{1.40(eV)} = 6526A^\circ$
 $= 653 \text{ nm}$
9. Which of.....
Sol. de-Broglie wave length $\lambda = \frac{h}{p}$
 $\Rightarrow P = \frac{h}{\lambda}$ or $P \propto \frac{1}{\lambda}$

10. In an α -decay.....
Sol. We have $K_\alpha = \frac{m_y}{m_y + m_\alpha} .Q$
 $\Rightarrow K_\alpha = \frac{A-4}{A} .Q \Rightarrow 48 = \frac{A-4}{A} .50 \Rightarrow A = 100$
11. The half-life.....
Sol. In one half life, half of the nuclei will decay as $T_{av.} > T_{1/2}$, more than half of the nuclei will decay in one average life time.
 $\frac{\ln 2}{\lambda_x} = \frac{1}{\lambda_y} \Rightarrow \lambda_y > \lambda_x$
 Rate of decay
 $\left| \frac{dN}{dt} \right| = \lambda . N.$
Aliter. $\therefore \left| \frac{dN}{dt} \right|_y > \left| \frac{dN}{dt} \right|_x$
 or $\frac{0.693}{\lambda_x} = \frac{1}{\lambda_y}$
 $\therefore \lambda_x = 0.693 \lambda_y$
 $\lambda_x < \lambda_y$ or $= \lambda N$
12. At time $t = 0$
Sol. $A = A_0 e^{-\lambda t}$
 $\ln A = \ln A_0 - \lambda t$
 $\Rightarrow y = c - mx$ equation of straight line.
 $\ln A$ versus t is a linearly decreasing graph with slope depending to λ . As λ does not change, slope remains same.
13. ^{22}Ne nucleus.....
Sol. $^{22}_{10}\text{Ne} \rightarrow ^{14}_6\text{X} + 2\alpha$
14. A nucleus ${}_n\text{X}^m$
Sol. In α -particle emission atomic mass decreases by 4 unit and atomic number decreases by 2 unit. In β -particle emission, atomic mass remains unchanged and atomic number increases by 1 unit.
 The reaction can be shown as
 ${}_n\text{X}^m \xrightarrow{\alpha} {}_{n-2}\text{Y}^{m-4}$
 ${}_{n-2}\text{Y}^{m-4} \xrightarrow{2\beta} {}_n\text{X}^{m-4}$
 Thus, the resulting nucleus is the isotope of parent nucleus and is ${}_n\text{X}^{m-4}$.
15. Complete the.....
Sol. In a nuclear reaction, atomic mass and charge number remain conserved,
 For a nuclear reaction to be completed, the mass number and charge number on both sides should be same.
 If we complete the equation by choice (1), then the complete reaction is
 Total atomic number on LHS = $92 + 0 = 92$
 ${}_{92}\text{U}^{235} + {}_0\text{n}^1 \rightarrow {}_{38}\text{Sr}^{90} + {}_{54}\text{Xe}^{143} + 3{}_0\text{n}^1$

Total atomic number on RHS
 = 38 + 54 + 0 = 92
 Total atomic number on RHS = 235 + 1 = 236
 Total atomic number on RHS
 = 90 + 143 + 3 × 1 = 236
 Thus, choice (1) is correct.

18. If a mica.....

Sol. When a mica sheet is introduced in the path of one of interfering beams, the whole interference pattern is displaced by an amount $\frac{D}{d} (\mu - 1) t$ in the direction of introduction of sheet.

19. In a YDSE.....

Sol. As $\lambda \ll d$; we can we $\beta = \frac{\lambda D}{d}$
 we get $\beta = \frac{500 \times 10^{-9} \times 1}{10^{-3}} = 0.5 \text{ mm}$.

As β is not very small; hence it might so happen that till 1000th maxima, we no longer can apply $y' = 1000 \times \beta$.
 Lets see if we can apply:
 At 1000th maxima. Path difference is 1000 λ .

$$\Rightarrow 1000 \lambda = d \sin \theta = \frac{d \times y}{\sqrt{D^2 + y^2}}$$

$$\Rightarrow (5 \times 10^{-4})^2 = \frac{(10^{-3} \text{ m})^2 \times y^2}{D^2 + y^2}$$

$$\Rightarrow 0.25 D^2 = y^2 (1 - 0.25)$$

$$\Rightarrow y = \left(\frac{0.25}{0.75} \right)^{\frac{1}{2}} \times D$$

$$y = \frac{D}{\sqrt{3}} = 0.577 \text{ m}$$

As 0.577 m. and 0.5 m. are quite distant, so we could not use $y' = 1000 \beta$ for such a high maxima.

21. The threshold.....

Sol. Threshold freq. = γ_0
 $K.E_{\text{max}} = h\gamma - w$
 $= h(2\gamma_0) - h\gamma_0$
 $= h\gamma_0 = \frac{1}{2} m v_1^2 \Rightarrow v_1 \alpha \sqrt{\gamma_0}$

$$\text{Now } K.E_{\text{max}} = h5\gamma_0 - h\gamma_0 = 4h\gamma_0 = \frac{1}{2} m v_2^2$$

$$v_2 \alpha \sqrt{v_1} \Rightarrow v_2 \alpha \sqrt{4\gamma_0} \Rightarrow v_2 = 2 \times v_1 = 8 \times 10^3 \text{ m/s}$$

22. The attractive.....

Sol. $U = eV = eV_0 \ln \left(\frac{r}{r_0} \right) \quad \therefore |F| = \left| -\frac{dU}{dr} \right| = \frac{eV_0}{r}$

This force will provide the necessary centripetal force.
 Hence

$$\frac{mv^2}{r} = \frac{eV_0}{r} \quad \text{or} \quad v = \sqrt{\frac{eV_0}{m}} \dots(i)$$

$$mvr = \frac{dh}{2\pi} \dots(ii)$$

Dividing equation (ii) by (i) we have

$$mr = \left(\frac{nh}{2\pi} \right) \sqrt{\frac{m}{eV_0}} \quad \text{or} \quad r_n \propto n$$

23. The wavelength.....

Sol. For $K_{\alpha} \Rightarrow \sqrt{v} \propto (z-1) \Rightarrow \frac{1}{\sqrt{\lambda}} \propto (z-1)$

$$\text{or } \lambda \propto \frac{1}{(z-1)^2} \dots(i)$$

$$4\lambda \propto \frac{1}{(z'-1)^2} \dots(ii)$$

$$\Rightarrow \frac{1}{4} = \frac{(z'-1)^2}{(z-1)^2} \Rightarrow \frac{z'-1}{z-1} = \frac{1}{2}$$

$$\Rightarrow 2z' - 2 = z - 1 \Rightarrow 2z' - 2 = 11 - 1 = 10$$

$$\Rightarrow z' = 6$$

24. If Bohr's

Sol. $(r_n) \left(\frac{m^2}{z} \right) (0.53 \text{ \AA}) = (n \times 0.53) \text{ \AA}$

$$\therefore \frac{m^2}{z} = n$$

$m = 5$ for 100 Fm²⁵⁷ (the outermost shell) and $z = 100$

$$\therefore n = \frac{(5)^2}{100} = \frac{1}{4}$$

25. Bombardment of.....

Sol. For ${}_Z^AX^A$, $Z = 0 + 5 - 2 = 3$ and $A = 1 + 10 - 4 = 7$

26. If 10% of a.....

Sol. After 5 days $N = 90\%$
 After 10 days $N = 90 - 9 = 81\%$

$$\text{After 15 days } 81 - \frac{10}{100} \times 81 \approx 73\%$$

28. For a parallel.....

Sol. Width of central maxima = $\frac{2\lambda D}{a}$

29. An unpolarised

Sol. When unpolarised light passes through a polaroid, intensity of emergent light,

$$i = I_0 (\cos^2 \theta)_{\text{mean}}$$

$$= I_0 \times \frac{1}{2} = \frac{I_0}{2}$$

30. Find the half.....

Sol. Here $\sin \theta = \frac{\lambda}{a}$

Where θ is half angular width of the central maximum.

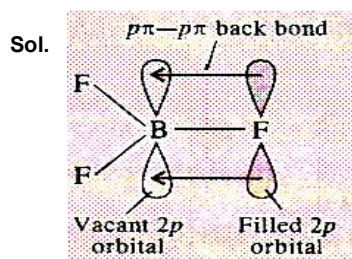
$$A = 12 \times 10^{-5} \text{ cm}, \lambda = 6000 \text{ \AA} = 6 \times 10^{-5} \text{ cm}$$

$$\therefore \sin \theta = \frac{\lambda}{a} = \frac{6 \times 10^{-5}}{12 \times 10^{-5}} = 0.50$$

or $\theta = 30^\circ$

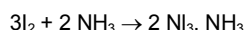
PART-B (CHEMISTRY)

31. The main factor



32. A dark violet solid (X)

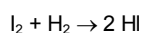
Sol. X is I₂ and Y is HI. The various reactions are



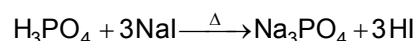
(X) (Explosive)



(violet coloured gas) (X)

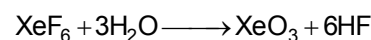
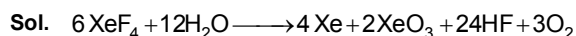


(X) (Y)



(salt of HI) (Y)

33. XeF₄ and XeF₆



Hence, (1) is correct

34. The presence of

Sol. F-centres made a crystal coloured and paramagnetic.

35. A binary solid (A⁺ B⁻)

Sol. Let B⁻ ions constituting the lattice = 1

∴ No. of tetrahedral holes = 2

Now, 25% of tetrahedral holes are occupied by A⁺.

∴ No. of A⁺ ions = 25% of 2 = 1/2

A : B

1/2 : 1

1 : 2

∴ formula is AB₂.

36. In a fcc arrangement

Sol. Out of the eight atoms present at the corners of unit cell, two atoms are missing. Thus, contribution of atoms per unit cell = 6/8. Since six atoms are present at the face centres and an atom at face centre is shared by two cubes, thus, contribution of atoms per unit cell = 6/2 = 3.

Formula of the compound is A_{6/8}B₃ or A₆B₂₄ or AB₄.

37. A compound XY

Sol. For bcc structure of the compounds

Nearest neighbor distance,

$$d = \frac{\sqrt{3}a}{2} = \frac{\sqrt{3} \times 480}{2} \text{ pm}$$

Now $d = r_+ + r_-$

$$\text{Thus, } r_+ + r_- = \frac{\sqrt{3} \times 480}{2} \text{ pm}$$

$$r_+ + 225 \text{ pm} = 415.68 \text{ pm} \text{ or } r_+ = 415.68 - 225 = 190.68 \text{ pm}$$

38. In the cyclic process

Sol. Process direction in the given diagram is clockwise so process direction in P-V diagram is anti-clockwise.

Net work done by system = Area of the circle

$$= \pi \times \frac{(P_2 - P_1)}{2} \cdot \frac{(V_2 - V_1)}{2}$$

39. Calculate average

Sol. (2) Average $C_{v,m} = \frac{n_1 C_{v,m1} + n_2 C_{v,m2}}{n_1 + n_2}$

$$= \frac{2 \times \frac{3}{2}R + 2 \times \frac{5}{2}R}{2 + 2} = 2R$$

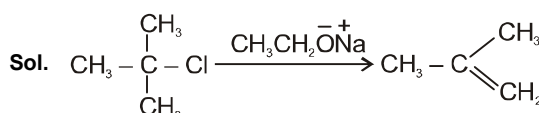
42. 1 mole of NH₃ gas

Sol. $T_1 V_1^{\gamma-1} = T_2 V_2^{\gamma-1} \Rightarrow 300 \times V^{1/3} = T_2 (8V)^{1/3} \Rightarrow T_2 = 150 \text{ K}$

$$W = nC_v(T_2 - T_1) = 1 \times 3R(150 - 300) = 3 \times 2(-150)$$

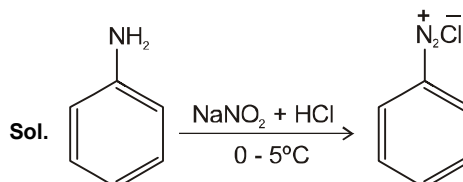
$$= -900 \text{ cal}$$

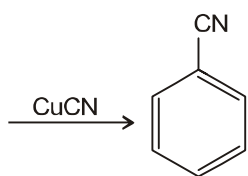
43. The major product



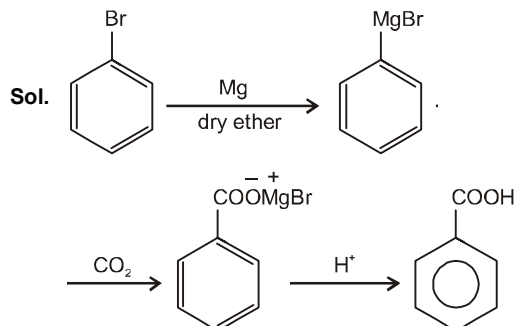
E - 2 reaction

44. Reaction of aniline

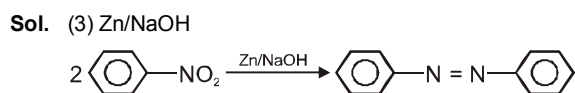




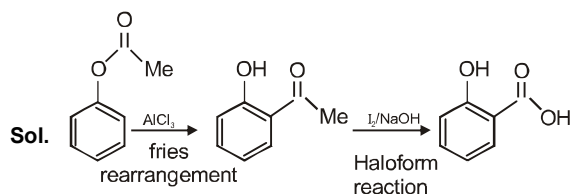
45. In the reaction.....



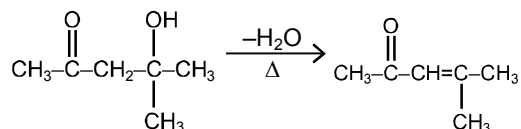
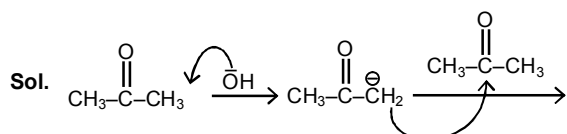
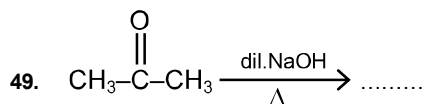
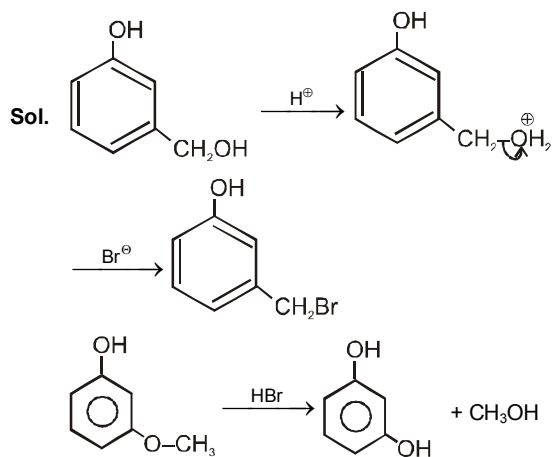
46. Azobenzene will



47. In the following transformation.....



48. Two isomeric compounds



The above reaction is Aldol condensation reaction.

51. Which of the following

Sol. ClF3 does not exist because F does not have vacant d-orbitals and no electron excitation is possible. Thus, F can not form more than one bond.

52. The correct sequence

Sol. $\text{HIO}_4, \text{ICl}_5, \text{I}_2, \text{HI}$

53. Greater reactivity

Sol. Factual question.

54. Which is **incorrect**.....

Sol. Orthorhombic crystal has $a \neq b \neq c, \alpha = \gamma = 90^\circ$. The given dimensions are for monoclinic crystal.

55. CsCl has bcc arrangement

Sol. CsCl has bcc structure.

Interionic distance (4) =

$$\frac{\sqrt{3}}{2} a = \frac{\sqrt{3}}{2} \times 400 = 346.4 \text{ pm}$$

56. A gas expands against

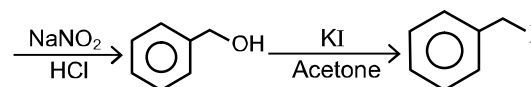
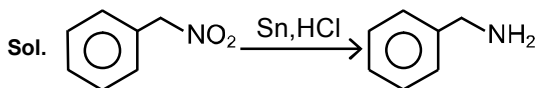
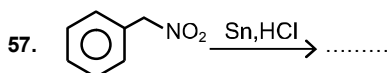
Sol. $\int dw = \int -P.dV$

$$\Rightarrow w = - \int_{V_1}^{V_2} 20 \frac{dV}{V} = -20 \ln \frac{V_2}{V_1}$$

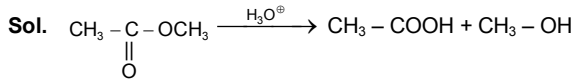
$$w = -46.06 \text{ Latm} = -4665.8 \text{ J}$$

$$\Delta U = q + w \Rightarrow 400 = q - 4665.8$$

$$q = 5065.8 \text{ J}$$



58. Acetaldehyde can be



60. The correct order

Sol. Rate of $\text{S}_{\text{N}}2$ reaction : $1^\circ > 2^\circ > 3^\circ$, as β -branching increases steric crowding increases in transition state so it makes less stable T.S.

PART-C (MATHEMATICS)

61. If $(\vec{a} \times \vec{b})^2$

Sol. $(\vec{a} \times \vec{b})^2 + (\vec{a} \cdot \vec{b})^2 = 144$

$$= |\vec{a}|^2 |\vec{b}|^2 (\sin^2 \theta + \cos^2 \theta) = 144$$

$$|\vec{a}| |\vec{b}| = 12$$

$$|\vec{b}| = \frac{12}{4} = 3$$

62. If ω is an.....

Sol. $\omega^{17} + \omega^7 = \omega^2 + \omega = -1$

$$\cos\left(-\pi + \frac{\pi}{3}\right) = \cos\left(-\frac{2\pi}{3}\right) = -\frac{1}{2}$$

63. The number.....

Sol. Vector perpendicular to \vec{a} and \vec{b} of unit length

$$= \frac{\pm(\vec{a} \times \vec{b})}{|\vec{a} \times \vec{b}|}$$

$$\text{i.e.} = 2$$

65. For parabola.....

Ans. Given equation

$$x^2 + 2y - 3x + 5 = 0 \Rightarrow x^2 - 3x + 5 = -2y$$

$$x^2 - 3x = \frac{9}{4} + 5 - \frac{9}{4} = -2y$$

$$\Rightarrow \left(x - \frac{3}{2}\right)^2 = -2y - \frac{11}{4}$$

$$\left(x - \frac{3}{2}\right)^2 = -2\left(y + \frac{11}{8}\right)$$

$$\Rightarrow x^2 = -4AY$$

$$\text{Hence vertex} \equiv \left(\frac{3}{2}, -\frac{11}{8}\right)$$

$$\text{Axis } x - \frac{3}{2} = 0 \Rightarrow x = \frac{3}{2}$$

$$\text{directrix } y = -\frac{7}{8} \Rightarrow \text{latus rectum} = 2$$

66. The line.....

Sol. We have the equation of the plane

$$\pi : x - y + z - 1 = 0 \text{ and } (x_1, y_1, z_1) = (0, 0, 0), (x_2, y_2, z_2) = (1, -2, -5)$$

Ratio

$$= \frac{\lambda}{1} = -\frac{(ax_1 + by_1 + cz_1 + d_1)}{(ax_2 + by_2 + cz_2 + d_1)} = \frac{(-1)}{1 - (-2) - 5 - 1} = -\frac{1}{3}$$

$$\Rightarrow \text{Ratio} = \frac{1}{3} \text{ (externally) (option C.)}$$

67. The order.....

$$\text{Sol. Here } \left[1 + \left(\frac{dy}{dx}\right)^2\right]^5 = x^2 \left(\frac{d^2y}{dx^3}\right)^2$$

$$\therefore \text{order} = 3, \text{degree} = 2$$

68. The conjugate.....

$$\text{Sol. complex number} = \frac{1}{-i - 1}$$

70. Number of.....

Sol. Given vectors are co-planar if

$$\begin{vmatrix} \lambda & 1 & 2 \\ 1 & \lambda & -1 \\ 2 & -1 & \lambda \end{vmatrix} = 0$$

$$\Rightarrow \lambda^3 - 6\lambda - 4 = 0$$

$$\Rightarrow (\lambda + 2)(\lambda^2 - 2\lambda - 2) = 0$$

will give 3 real value of λ .

71. A point

Sol. Given that

$$y^2 + z^2 + x^2 + x^2 + y^2 = 36$$

$$\Rightarrow 2(x^2 + y^2 + z^2) = 36$$

$$\Rightarrow x^2 + y^2 + z^2 = 18$$

$$\Rightarrow \sqrt{x^2 + y^2 + z^2} = 3\sqrt{2}$$

72. Equation.....

Sol. d.c.s of x-axis = (1, 0, 0)

$$\text{so } \frac{x}{1} = \frac{y}{0} = \frac{z}{0}$$

73. The slope.....

Sol. $\frac{dy}{dx} = \frac{1}{2y}$

$\Rightarrow y^2 = x + c$

$\therefore (4, 3)$ satisfies

$\Rightarrow 9 = 4 + c$

$\Rightarrow c = 5$

$\therefore y^2 = x + 5$

74. The locus.....

Sol. $|z - 1| = |z + i|$

$\Rightarrow |(x - 1) + iy| = |x + i(y + 1)|$

$\Rightarrow (x - 1)^2 + y^2 = x^2 + (y + 1)^2$

$\Rightarrow x + y = 0$, which is a line through the origin.

75. The equation.....

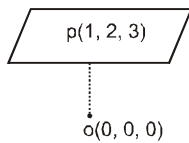
Sol. The point of intersection of two perpendicular tangents the parabola lies on directrix of the parabola.

Since, equation of directrix is $0 = x + 2$

So point is $(-2, 0)$

76. Equation of.....

Sol.



Required equation = $1(x - 1) + 2(x - 2) + 3(x - 3) = 0$

$\Rightarrow x + 2y + 3z = 14$

77. The solution

Sol. $(x^2 + y^2) dy = xy dx$; $\frac{dx}{dy} = \frac{x}{y} + \frac{y}{x}$

Let $x = vy$

\therefore we get $v + y \frac{dv}{dy} = v + \frac{1}{v}$

$\Rightarrow y \frac{dv}{dy} = \frac{1}{v} \Rightarrow \int v dv = \int \frac{dy}{y}$

$\Rightarrow \frac{v^2}{2} = \ln y + c$

$\Rightarrow \frac{x^2}{2y^2} = \ln y + c$

at $x = 1$; $y = 1$

$\Rightarrow c = \frac{1}{2}$

$\Rightarrow \frac{x^2}{2y^2} = \ln y + \frac{1}{2}$

at $x = x_0$; $y = e$

$\frac{x_0^2}{2e^2} = 1 + \frac{1}{2}$

$\Rightarrow x_0^2 = 3e^2$

$\Rightarrow x_0 = \pm \sqrt{3} e$

78. If $z = \frac{7-i}{3-4i}$

Sol. $z = \frac{7-i}{3-4i} = \frac{(7-i)(3+4i)}{9+16} = \frac{25+25i}{25}$

$= 1 + i = \sqrt{2} (\cos \pi/4 + i \sin \pi/4)$

$z^{14} = 2^7 \left\{ \cos \left(\frac{14\pi}{4} \right) + i \sin \left(\frac{14\pi}{4} \right) \right\}$

$= 2^7 [\cos (7\pi/2) + i \sin(7\pi/2)] = -2^7 i$

79. The distance

Sol. The equation of a line passing through P (2, 3, 4) and parallel to the line

$\frac{x+3}{3} = \frac{y-2}{6} = \frac{z}{2}$ is $\frac{x-2}{3} = \frac{y-3}{6} = \frac{z-4}{2}$

Any point on this line is given by

Any point on this line is given by

$\frac{x-2}{3} = \frac{y-3}{6} = \frac{z-4}{2} = \lambda$

$\Rightarrow x = 3\lambda + 2, y = 6\lambda + 3, z = 2\lambda + 4$

If the point Q $(3\lambda + 2, 6\lambda + 3, 2\lambda + 4)$ lies on the plane $3x + 2y + 2z + 5 = 0$. Then,

$3(3\lambda + 2) + 2(6\lambda + 3) + 2(2\lambda + 4) + 5 = 0$

$\Rightarrow 25\lambda + 25 = 0$

$\Rightarrow \lambda = -1$

So the coordinates of Q are $(-1, -3, 2)$

\therefore Required distance = $PQ = \sqrt{(2+1)^2 + (3+3)^2 + (4-2)^2}$

$= \sqrt{9+36+4} = 7$.

80. If $|\vec{a}| = 7$

Sol. $|\vec{a} + \vec{b}|^2 + |\vec{a} - \vec{b}|^2 = 2(|\vec{a}|^2 + |\vec{b}|^2) = 2(49 + 121) = 340$

$|\vec{a} - \vec{b}|^2 = 340 - 300 = 40$

81. The equation.....

Sol. Direction ratios of the normal are $\langle -1, -5, 6 \rangle$

\therefore equation of a plane is $-1(x - 2) - 5(y + 3) + 6(z - 1) = 0$

i.e. $x + 5y - 6z + 19 = 0$.

82. The inclination.....

Sol. $2x + 2 \frac{dy}{dx} = 8$

at $x = 5$, $\frac{dy}{dx} = -1$

84. If $|z - i|$

Sol. Given $|z - i| < 1$

Now, $|z + 12 - 6i| = |(z - i) + (12 - 5i)| \leq |z - i| + |12 - 5i|$

$[\because |z_1 + z_2| \leq |z_1| + |z_2|]$

$< 1 + 13 = 14$

Hence $|z + 12 - 6i| < 14$

85. If $y = 2x$

Sol. $y = 2x - 3, y^2 = 4a \left(x - \frac{1}{3}\right)$

$(2x - 3)^2 = 4a \left(x - \frac{1}{3}\right)$

$4x^2 - 12x + 9 = 4ax - \frac{4a}{3}$

$4x^2 - (12 + 4a)x + \left(9 + \frac{4a}{3}\right) = 0$

$\therefore D = 0$

$(12 + 4a)^2 - 4 \cdot 4 \cdot \left(9 + \frac{4a}{3}\right) = 0$

$\therefore a = \frac{-14}{3}$

86. If $z_1 = 1 + i$

Sol. $\arg \left(\frac{z_1}{z_2}\right) = \arg(1 + i) - \arg(\sqrt{3} + i)$

$= \frac{\pi}{4} - \frac{\pi}{6} = \frac{\pi}{12} \Rightarrow \text{Arg} \left(\frac{z_1}{z_2}\right)^{50} = 50 \left(\frac{\pi}{12}\right) = 4\pi + \frac{\pi}{6}$

So principal argument is $\left(\frac{\pi}{6}\right)$

87. Let Z is a.....

Sol. Z will be circumcentre of triangle having vertices (0, 0), (0, 6) and (8, 0) in argand plan.

88. The equation.....

Sol. Direction ratio of normal to the plane

$\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 2 & 3 & 1 \\ 3 & 1 & 2 \end{vmatrix} = -7\hat{i} + 7\hat{j} - 7\hat{k}$

So equation of plane is

$-7(x - 1) + 7(y - 2) - 7(z - 3) = 0$

$-7x + 7y - 7z + 14 = 0$

$-7x + 7y - 7z + 14 = 0$

$x - y + z = 2$

DATE : 11-02-2018**ANSWER KEY****CODE-0****PART-A (PHYSICS)**

- | | | | | | | | | | | | | | |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 1. | (4) | 2. | (2) | 3. | (2) | 4. | (3) | 5. | (4) | 6. | (4) | 7. | (4) |
| 8. | (2) | 9. | (4) | 10. | (2) | 11. | (3) | 12. | (2) | 13. | (2) | 14. | (1) |
| 15. | (1) | 16. | (1) | 17. | (4) | 18. | (2) | 19. | (2) | 20. | (2) | 21. | (3) |
| 22. | (2) | 23. | (3) | 24. | (2) | 25. | (3) | 26. | (2) | 27. | (3) | 28. | (4) |
| 29. | (3) | 30. | (4) | | | | | | | | | | |

PART-B (CHEMISTRY)

- | | | | | | | | | | | | | | |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 31. | (3) | 32. | (4) | 33. | (1) | 34. | (2) | 35. | (2) | 36. | (4) | 37. | (1) |
| 38. | (3) | 39. | (2) | 40. | (3) | 41. | (3) | 42. | (1) | 43. | (2) | 44. | (3) |
| 45. | (3) | 46. | (3) | 47. | (2) | 48. | (1) | 49. | (3) | 50. | (2) | 51. | (2) |
| 52. | (4) | 53. | (1) | 54. | (2) | 55. | (1) | 56. | (3) | 57. | (4) | 58. | (1) |
| 59. | (3) | 60. | (1) | | | | | | | | | | |

PART-C (MATHEMATICS)

- | | | | | | | | | | | | | | |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 61. | (3) | 62. | (2) | 63. | (2) | 64. | (1) | 65. | (3) | 66. | (3) | 67. | (1) |
| 68. | (3) | 69. | (4) | 70. | (3) | 71. | (3) | 72. | (3) | 73. | (3) | 74. | (3) |
| 75. | (4) | 76. | (3) | 77. | (1) | 78. | (3) | 79. | (3) | 80. | (3) | 81. | (1) |
| 82. | (3) | 83. | (5) | 84. | (1) | 85. | (4) | 86. | (1) | 87. | (3) | 88. | (5) |
| 89. | (4) | 90. | (2) | | | | | | | | | | |