

MATHEMATICS

1. Let L_1 be a line.....
Sol. The chords are of equal length, then the distances of the centre from the lines are equal.

Let L_1 be $y - mx = 0$. Centre is $\left(\frac{1}{2}, -\frac{3}{2}\right)$

$$\frac{\left| \frac{3}{2} - m \right|}{\sqrt{m^2 + 1}} = \frac{\left| \frac{1}{2} - \frac{3}{2} - 1 \right|}{\sqrt{2}} \Rightarrow 7m^2 - 6m - 1 = 0$$

$$\Rightarrow m = 1, -\frac{1}{7}$$

2. Consider the circles

Sol. $S_1 : x^2 + y^2 = 4$
 and $S_2 = x^2 + y^2 - 2x - 4y + 4 = 0$
 centre: $(0, 0)$; radius = 2
 centre: $(1, 2)$; radius = 1

(A) d = distance between centres = $\sqrt{5}$

$$r_1 + r_2 = 3 \Rightarrow |r_1 - r_2| = 1$$

$$\therefore |r_1 - r_2| < d < r_1 + r_2$$

\therefore these 2 circles are intersecting.

\therefore number of common tangents is 2.

(B) $P(h, k)$ power of point P is same w.r.t. these two circles.

$$\therefore \sqrt{h^2 + k^2 - 4} = \sqrt{h^2 + k^2 - 2h - 4k + 4}$$

$$-4 = -2h - 4k + 4$$

$$2h + 4k - 8 = 0$$

$$x + 2y - 4 = 0$$

(C) y intercept of S_1 is $2\sqrt{4} = 4$

y intercept of S_2 is $2\sqrt{4-4} = 0$

\therefore sum of y -intercept = 4

(D) $2(0+0) = -4+4$

\therefore circle is orthogonal.

3. Locus of the

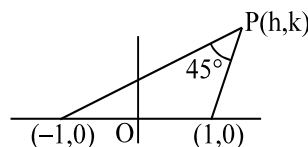
Sol. $m_1 = \frac{k}{h-1}$; $m_2 = \frac{k}{h+1}$;

$$\therefore \tan 45^\circ = \left| \frac{\frac{k}{h-1} - \frac{k}{h+1}}{1 + \frac{k^2}{h^2-1}} \right| = \left| \frac{k(h+1-h+1)}{h^2-1+k^2} \right|$$

$$h^2 + k^2 - 1 = \pm 2k$$

$$\text{locus is } x^2 + y^2 \pm 2y - 1 = 0$$

$$x^2 + (y \pm 1)^2 = (\sqrt{2})^2$$



$(0, 1)$ or $(0, -1)$ and radius = $\sqrt{2}$

4. The sum of the
Sol. The total number of numbers that can be formed with the digits 3, 4, 5, 6 taken all at a time
 = $P(4, 4) = 4! = 24$
 Each of the digits 3, 4, 5, 6 occurs in 3!
 = $3 \times 2 = 6$ times in the unit's place of all the numbers
 = $(3 + 4 + 5 + 6) \times 6 = 18 \times 6 = 108$

5. If letters of word

Sol. (A)

A					
---	--	--	--	--	--

 $\frac{5!}{2!} = 60$

K					
---	--	--	--	--	--

 $\frac{5!}{2!2!} = 30$

P	A	A			
---	---	---	--	--	--

 $\frac{3!}{2!} = 3$

P	A	K			
---	---	---	--	--	--

 $\frac{3!}{2!} = 3$

P	A	R	A		
---	---	---	---	--	--

 $2! = 2$

P	A	R	K	A	R
---	---	---	---	---	---

 $1 = 1$

so rank is 99

(B)

A	A				
---	---	--	--	--	--

 $\frac{4!}{2!} = 12$

A	K				
---	---	--	--	--	--

 $\frac{4!}{2!} = 12$

A	P				
---	---	--	--	--	--

 $\frac{4!}{2!} = 12$

A	R	A			
---	---	---	--	--	--

 $3! = 6$

A	R	K			
---	---	---	--	--	--

 $3! = 6$

Sum = 48

49th word is

A	R	P	A	K	R
---	---	---	---	---	---

50th word is

A	R	P	A	R	K
---	---	---	---	---	---

(C)

P					
---	--	--	--	--	--

 $\frac{5!}{2!2!} = 30$

Rank of RAAKPR is $60 + 30 + 30 + 1 = 121$



6. You have n
Sol. The number of all possible pairs of objects that could be obtained from 'n' objects is ${}^n C_2 = \frac{n(n-1)}{2}$

Total weight of $\frac{n(n-1)}{2}$

pairs is $\frac{n(n-1)}{2} \times 2 \times w = 120 \dots(1)$

The number of all possible triplets are ${}^n C_3$
 Total weight of ${}^n C_3$ triplets is

$\frac{n(n-1)(n-2)}{6} \times 3 \times w = 480 \dots(2)$

on solving $n = 10$ and $w = \frac{4}{3}$

11. Equation of circle

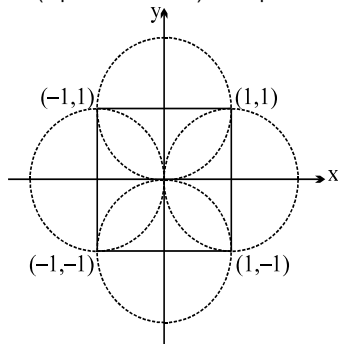
Sol. Let radius = r

Then $C_1 C_2 = r_1 \pm r_2$

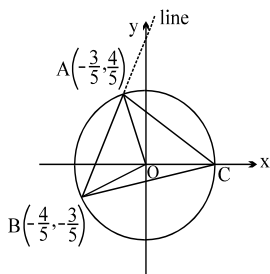
$5 = r \pm 1$
 $r = 4$ or 6

28. Four unit circles.....

Sol. Ar. (square of sides 2) = 4 sq. units



31. The circle with



Sol.

Solving $y = 7x + 5$ and the circle $x^2 + y^2 = 1$

$A\left(-\frac{3}{5}, \frac{4}{5}\right)$ and $B\left(-\frac{4}{5}, -\frac{3}{5}\right)$

$m_{OA} = -\frac{4}{3}$; $m_{OB} = \frac{3}{4}$

Hence $\angle AOB = 90^\circ$

$\Rightarrow \angle ACB = \frac{\pi}{4} = \tan^{-1}(1)$

32. A variable circle

Sol. Power : $P = a^2 + b^2 - 2(t^2 - 3t + 1)a - 2(t^2 + 2t)b + t$

$= -(2a + 2b)t^2 + (6a - 4b + 1)t + a^2 + b^2 - 2a$

This power is independent of the parameter t if and only if $2a + 2b = 0 \Rightarrow a = -b$ and $6a - 4b + 1 = 0$

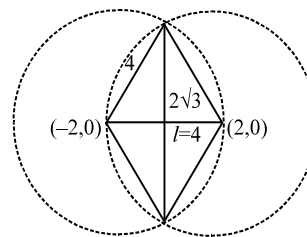
$\Rightarrow a = -\frac{1}{10}$ and $b = \frac{1}{10}$

33. A rhombus is

Sol. circles with centre (2, 0) and (-2, 0) each with radius 4
 $\Rightarrow y$ -axis is their common chord.

The inscribed rhombus has its diagonals equal to 4 and $4\sqrt{3}$

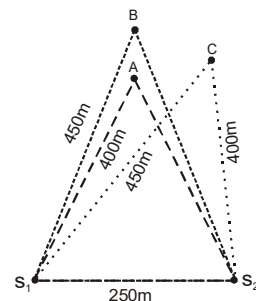
$\therefore A = \frac{d_1 d_2}{2} = 8\sqrt{3}$



PHYSICS

1. Two radio station

Sol.



At points A and B, path difference between the waves coming from two radio stations is zero. Hence there will be constructive interference at A and B.

For point C, path difference between the waves is 50 metre i.e.

$\frac{\lambda}{2}$ so destructive interference takes places at point C.

3. At displacement

Sol. Self-explanatory

4. Two wave

Sol. Resultant Displacement

$y = y_1 + y_2$

for y to be zero

$(2x - 3t)^2 = (2x + 3t - 6)^2$

on solving $(x - \frac{3}{2})(t - 1) = 0$

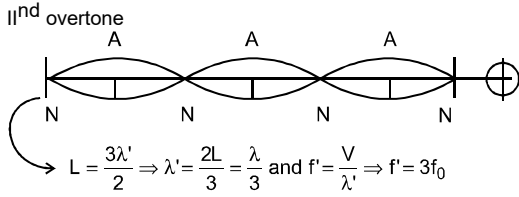
Therefore,

at $x = \frac{3}{2}$, resultant displacement is zero for all values of t.

5. A string of

Sol. Fundamental mode $\longrightarrow L = \frac{\lambda}{2} \Rightarrow \lambda = 2L$

$$f_0 = \frac{V}{\lambda} = \frac{V}{2L}$$



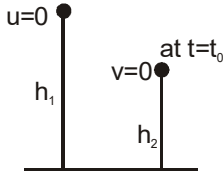
So it contain 4 node and -3 antinode frequency is 3 time at fundamental frequency wavelength is $\frac{1}{3}$ time at fundamental wavelength.

6. A small ball

Sol. From the figure-1 work done by gravity from $t = 0$ to $t = t_0$ is

$$W = mg(h_1 - h_2)$$

at $t=0$



fixed horizontal surface
Since initial and final velocity of ball is zero its average acceleration will be zero.
Since net work done is zero from time interval $t = 0$ to $t = t_0$.
Hence work done by forces except gravity is $mg(h_2 - h_1)$.

7. A uniform

Sol. $T = 2\pi \sqrt{\frac{I}{mgy_{cm}}}$

$$I = \frac{MR^2}{2} - \left(\frac{1}{2} \frac{M}{4} \left(\frac{R}{2} \right)^2 + \frac{M}{4} \left(\frac{R}{2} \right)^2 \right) = \frac{13MR^2}{32}$$

$$\Rightarrow y_{cm} = \frac{R}{6}$$

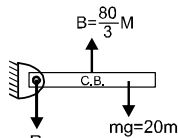
$$m = \frac{3M}{4} c$$

$$T = 2\pi \sqrt{\frac{13R}{4g}}$$

8. A rod of mass

Sol. $B = \rho_{\ell} V_{sub} g = 10^3 \times \left(\frac{8}{3} \times 10^{-3} \right) \times 1 \times 10$

$$B = \frac{80}{3} \text{ N}$$



Balancing torque about the hinge point,

$$\left(\frac{80}{3} \right) \times \frac{1}{2} = (20) (x)$$

(A,B) $x = \frac{2}{3} \text{ m}$

Applying force balance,

$$R + 20 = \frac{80}{3}$$

(C) $R = \frac{20}{3} \text{ M}$

(D) If we displace the rod, still torque will remain balanced.

9. A rod of length

Sol. Centre of mass = $\frac{\int_0^L \lambda_0 x^2 dx}{\int_0^L \lambda_0 x dx} = \frac{\lambda_0 \frac{L^3}{3}}{\lambda_0 \frac{L^2}{2}} = \frac{2L}{3}$

$$\text{Mass} = \int_0^L \lambda_0 x dx = \frac{\lambda_0 L^2}{2}$$

$$T_1 + T_2 = \frac{\lambda_0 L^2}{2} g$$

Balancing torque about A,

$$T_2 \times L = \frac{\lambda_0 L^2}{2} \times \frac{2L}{3} g$$

$$T_2 = \frac{\lambda_0 L^2 g}{3} \quad T_1 = \frac{\lambda_0 L^2 g}{6}$$

$$L = 0.3 \text{ m} \quad \lambda_0 = 100 \text{ kg/m}^2$$

$$T_2 = 30 \text{ N}, \quad T_1 = 15 \text{ N}$$

$$V_1 = \sqrt{\frac{T_1}{\mu_1}} \quad \text{and} \quad V_2 = \sqrt{\frac{T_2}{\mu_2}}$$

$$\frac{V_1}{V_2} = \sqrt{\frac{T_1 \mu_2}{T_2 \mu_1}} = \frac{3}{2}$$

$$\frac{n_1}{2\ell} \sqrt{\frac{15}{2}} = \frac{n_2}{2\ell} \sqrt{\frac{30}{9}} \quad (\ell \text{ is the length of the wire})$$

$$\frac{n_1}{n_2} = \frac{2}{3}$$

10. A particle is

Sol. The mean position of the particle will be at $x = 2A$ and its displacement amplitude will be A .

11. A source emit

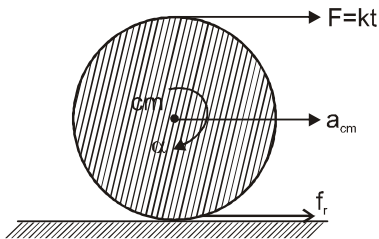
Sol. $\lambda' = \frac{V - V_s}{f} = \frac{332 - 32}{1000} = 0.3 \text{ m}$

$$f' = f \frac{(V + V_0)}{V - V_s} = 1000 \times \frac{332 + 64}{332 - 32} = 1320 \text{ Hz}$$

$$\lambda'' = \frac{V - V_0}{f'} = 0.2 \text{ m.}$$

12. A uniform solid

Sol.



$$F + f_r = (m) (a_{cm})$$

.....(i)

$$(\tau_{net})_{cm} = FR - f_r R = \left(\frac{MR^2}{2}\right) \alpha \quad \text{.....(ii)}$$

$$a_{cm} = \alpha R \quad \text{.....(iii)}$$

Solving get, $a_{cm} = \frac{4F}{3m}$ and $f_r = \frac{F}{3} = \frac{kt}{3}$

So, graph between f_r and t will be straight line

(B) & (C) $a_{cm} = \frac{4}{3m} (kt) = \frac{dv_{cm}}{dt}$

$$\int_{v_{cm}=0}^{v_{cm}=v_{cm}} dv_{cm} = \frac{4k}{3m} \int_{t=0}^{t=t} t dt$$

$$\Rightarrow v_{cm} = \frac{2k}{3m} t^2$$

So, velocity of the highest point is $2v_{cm} = \frac{4k}{3m} t^2$

(D) To prevent sliding,

$$f_r < \mu_s M$$

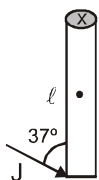
$$\frac{kt}{3} < (\mu_s) (mg)$$

$$t < \frac{3\mu_s mg}{k}$$

13. A thin rod of

Sol. $J \sin 37^\circ \ell = \frac{m \ell^2}{3} \omega = L$

$$K.E. = \frac{1}{2} I \omega^2$$



$$V = \frac{\ell}{2} \omega$$

14. Graph shows

Sol. Area under the curve is equal to number of molecules of the gas sample. Hence

$$N = \frac{1}{2} \cdot a \cdot V_0 \Rightarrow aV_0 = 2N$$

$$V_{avg} = \int_0^\infty v N(v) dv = \frac{1}{N} \int_0^{V_0} C \left(\frac{a}{V_0} \cdot v\right) dv = \frac{2}{3} V_0 \Rightarrow$$

$$\frac{V_{avg}}{V_0} = \frac{2}{3}$$

$$V_{rms}^2 = \frac{1}{N} \int_0^\infty v^2 N(v) dv$$

$$= \frac{1}{N} \int_0^{V_0} v^2 \left(\frac{a}{V_0} \cdot v\right) dv = \frac{V_0^2}{2} \Rightarrow \frac{V_{rms}}{V_0} = \frac{1}{\sqrt{2}}$$

Area under the curve from $0.5 V_0$ to V_0 is $\frac{3}{4}$ of total area.

15. The relation

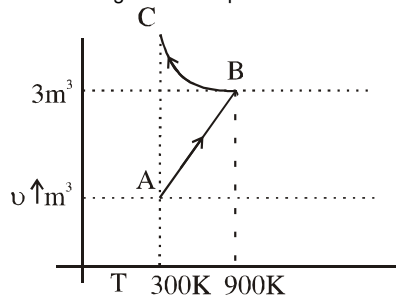
Sol. As $\Delta Q = W$
 $\Rightarrow \Delta U = 0 \therefore T_A = T_B$

16. The ratio

Sol. Since $T_C = 2T_A$
 $\therefore P_C = 2P_A$

18. Work done

Sol. The V-T diagram for the process would be :



17) $\frac{V_1}{T_1} = \frac{V_2}{T_2}$ (for process AB)

$$\frac{1}{300} = \frac{3}{T_2} \Rightarrow T_2 = 900K$$

$$dU = nC_v dT = 2 \left[\frac{5R}{2} \right] [600] = 10^4 J$$

18) $dW = W_{ab} + W_{bc} = PdV + \frac{P_1 V_1 - P_2 V_2}{\gamma - 1}$

$$= nRdT + \frac{nRdT}{\gamma - 1} = 35kJ$$

19. Choose the

Sol. Using equation of continuity $A_1 v_1 = A_2 v_2$

$$(12 \text{ cm}^2) v_A = (6 \text{ cm}^2) (8.0 \text{ m/s})$$

$$v_A = 4.0 \text{ m/s}$$

20. Choose the

Sol. Applying Bernoulli's principle between point A and C that are at same horizontal level

$$\frac{1}{2}\rho V_A^2 + p_A = \frac{1}{2}\rho V_C^2 + p_{atm}$$

$$\Rightarrow p_A = (1.01 \times 10^5 \text{ N/m}^2) + \frac{1}{2} \times 13,600 (8^2 - 4^2)$$

$$= 4.27 \times 10^5 \text{ N/m}^2$$

21. A sample of

Ans. (A) – Q ; (B) – P ; (C) – Q ; (D) – S

Sol. Number of moles = n (assume)

$$C_V = \frac{3}{2} R \quad \text{and} \quad C_P = \frac{5}{2} R \quad (\text{Monoatomic})$$

$$T_A = 300 \text{ K}$$

$$T_B / T_A = 2/1$$

$$\text{Hence} \quad T_B = 600 \text{ K}$$

$$Q_{BC} = W_{BC} = nRT_B \ln \left(\frac{V_C}{V_B} \right) = (n) (R) (600) \ln \left(\frac{4V_0}{2V_0} \right) =$$

$$600 n R \ln 2$$

$$\therefore Q_{DA} = W_{DA} = nRT_D \ln \left(\frac{V_A}{V_D} \right) = (n) (R) (300) \ln$$

$$\left(\frac{V_0}{4V_0} \right) = 300 n R \ln \left(\frac{1}{4} \right)$$

$$Q_{DA} = -600 n R \ln 2$$

$$\therefore \left| \frac{\Delta Q_{B \rightarrow C}}{\Delta Q_{D \rightarrow A}} \right| = 1$$

Pressure is constant from A to B.

Pressure decreases from B to C and again from C to D.

Pressure increases from D to A.

$$\text{So} \quad \frac{P_{\max}}{P_{\min}} = \frac{P_A}{P_D} = \left(\frac{P_A V_A}{P_D V_D} \right) \left(\frac{V_D}{V_A} \right)$$

$$= \left(\frac{nRT_A}{nRT_D} \right) \left(\frac{4}{1} \right) = 4$$

Ans : (s)

22. An ideal gas

Sol. Root mean square speed of molecules

$$= \sqrt{\frac{3RT}{M}} = 1.732 \sqrt{\frac{RT}{M}}$$

Most probable speed of molecules

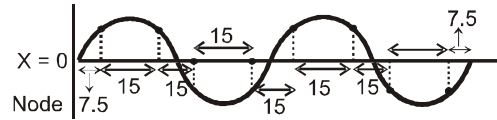
$$= \sqrt{\frac{2RT}{M}} = 1.44 \sqrt{\frac{RT}{M}}$$

Average velocity of a molecule is zero

Speed of any individual molecule may be anything.

23. A string 120 cm

Sol. 3rd overtone \Rightarrow which covers 120 cm length of the string



24. In the resonance
Sol. for first resonance

$$\ell_1 + \varepsilon = \frac{v}{4f_0}$$

for second resonance

$$\ell_2 + \varepsilon = \frac{3v}{4f_0}$$

for the third resonance

$$\ell_3 + \varepsilon = \frac{5v}{4f_0}$$

Solving get $\ell_3 = 2\ell_2 - \ell_1$

25. A sound

Sol. $I = \frac{p^2}{\rho v}$

$$\frac{I_2}{I_1} = \frac{p_2^2}{p_1^2} \times \frac{\rho_1 v_1}{\rho_2 v_2}$$

$$= \frac{9}{16} \times \frac{1.5}{3} \times \frac{400}{1200} = \frac{9}{16 \times 3 \times 2} = \frac{3}{32}$$

26. A non-uniform

Sol. $\mu = Kx = \frac{dM}{dx}$

$$\int_0^M dM = \int_0^{\ell} Kx \, dx \quad \text{and} \quad K = \frac{2M}{\ell^2}$$

$$v = \sqrt{\frac{F}{\mu}} = \sqrt{\frac{F}{Kx}} = \frac{dx}{dt}$$

$$\Rightarrow \int_0^{\ell} \sqrt{x} \, dx = \sqrt{\frac{F}{K}} \int_0^t dt$$

$$\therefore t = \sqrt{\frac{4\ell^3}{9} \cdot \frac{K}{F}} = \sqrt{\frac{4\ell^3}{9F} \cdot \frac{2m}{\ell^2}}$$

$$= \sqrt{\frac{8M\ell}{9F}} = \sqrt{\frac{8 \times 45 \times 1.5}{9 \times 15}} = 2.$$

27. A sonometer

Sol. $\ell_1 + \ell_2 + \ell_3 = 114$

$$n_1 = \frac{1}{2\ell_1} \sqrt{\frac{T}{m}}, \quad n_2 = \frac{1}{2\ell_2} \sqrt{\frac{T}{m}},$$

$$n_3 = \frac{1}{2\ell_3} \sqrt{\frac{T}{m}}$$

$$n_2 = \frac{n_1}{3} \quad \& \quad n_3 = \frac{n_1}{4}$$

$$\ell_1 = 72 \text{ cm}, \quad \ell_2 = 24 \text{ cm}, \quad \ell_3 = 18 \text{ cm},$$

$$\ell_1 = 12 \times 6 \text{ cm}$$

$$x = 6$$



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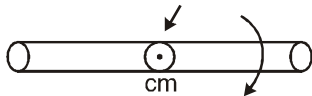
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28. A thin uniform

Sol. As $L = \frac{2}{3} t^3 + 2t^2$

$$\frac{dL}{dt} = 2t^2 + 4t = J$$

$$J = 2 \left(\frac{9}{4} \right) + 4 \left(\frac{3}{2} \right)$$



$$J = \frac{21}{2} \text{ Nm}$$

an $\tau = I \alpha$

$$\frac{21}{2} = \frac{(0.25)(1.2)^2}{12} \alpha$$

$$\alpha = 350 \text{ rad/sec}^2$$

$$= 50 \times 7$$

$\therefore x = 7$

29. A uniform

Sol. From conservation of angular momentum.

$$mu \frac{L}{2} + mu \frac{L}{2} = \left[2m \frac{L^2}{12} + m \left(\frac{L}{2} \right)^2 + \left(\frac{L}{2} \right)^2 \right] \omega$$

$$muL = \left[\frac{mL^2}{6} + \frac{mL^2}{4} + \frac{mL^2}{4} \right] \omega = \frac{2}{3} mL^2 \omega$$

$$\text{or } \omega = \frac{3u}{2L} = \frac{3 \times 6}{2 \times 1} = 9 \text{ rad/s}$$

30. A rough track

Sol. Minimum velocity required at D $\frac{mv^2}{R-r} = mg$

$$\Rightarrow v = \sqrt{g(R-r)}$$

Energy conservation between A and D

$$mg(h-2R+r) = \frac{1}{2} mv^2 + \frac{1}{2} \frac{mr^2}{2} \times \frac{v^2}{r^2}$$

$$g(h-2R+r) = \frac{3}{4} g(R-r)$$

$$R = \frac{4h+7r}{11} = \frac{52+14}{11} = 6 \text{ cm.}$$

31. Find the moment

Sol. $I = \frac{ma^2}{12} + m \left(\frac{a}{\sqrt{2}} \right)^2 = \frac{7ma^2}{12} = 7.$

32. How many

Sol. The velocity of sound propagation in gas is

$$v = \sqrt{\frac{\gamma P}{\rho}} \Rightarrow 330 = \sqrt{\frac{\gamma \times 10^5}{1400/1089}}$$

$$\Rightarrow \gamma = 1.4$$

The relation between γ and degrees of freedom 'F' is

$$\gamma = 1 + \frac{2}{F} \quad 1.4 = 1 + \frac{2}{F}$$

$$\Rightarrow F = 5$$

[Ans: 5]

33. Figure shows

Sol. For rectangular hyperbola

$XY = \text{constant}$

$U_p = \text{constant}$

$n C_V T_p = \text{constant}$

$$n \frac{5}{2} R T \frac{M}{V} = \text{constant}$$

$$\frac{5}{2} P \frac{VM}{V} = \text{constant}$$

$\Rightarrow P = \text{constant}$ (isobaric process)

$$\Delta U_{B \rightarrow A} = 37 - 2 = 35 \text{ J} = \frac{5}{2} nR\Delta T$$

$$14 \text{ J} = nR\Delta T \quad \therefore W = 14 \text{ J}$$

34. 2 moles of

Sol. $\Delta U = (1/2)f_1 n_1 R\Delta T + (1/2)f_2 n_2 R\Delta T = (1/2)(f_1 n_1 + f_2 n_2) R\Delta T =$

$$(1/2)(5 \times 2 + 3 \times 5) 2 \Delta T = 25 \Delta T$$

Since $W = 0, \Delta U = Q = 25 \text{ Cal}$

35. A wheel of

Sol. Net external torque on the system is zero, Therefore, angular momentum will remain conserved. Thus,

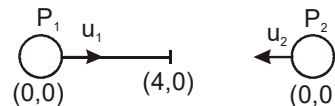
$$I_1 \omega_1 = I_2 \omega_2 \quad \text{or} \quad \omega_2 = \frac{I_1 \omega_1}{I_2}$$

Here, $I_1 = I, \omega_1 = \omega_0, I_2 = I + mR^2$

$$\therefore \omega_2 = \frac{I \omega_0}{I + mR^2}$$

36. Two particles

Sol.



$$u_1 = \frac{4}{t_0} \text{ and } u_2 = \frac{6}{t_0}$$

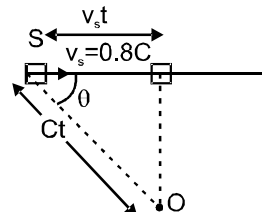
At $t = t_0$, x-coordinate of c.m. is 4. Hence after further t_0 time

$$X_{cm} = 4 + v_{cm} \cdot t_0$$

$$= 4 + \left(\frac{4}{t_0} - \frac{6}{t_0} \right) \cdot t_0 = 3$$

37. A source of sound

Sol.



$$\cos \theta = \frac{v_s t}{Ct} = 0.8$$

$$f = f \left[\frac{C}{C - v_s \cos \theta} \right]$$

$$= 1.8 \left[\frac{C}{C - 0.8C \cos \theta} \right]$$

$$= 1.8 \frac{1}{1 - 0.8^2} = 1.8 \frac{1}{1 - 0.64} = \frac{1.8}{0.36}$$

$$= \frac{180}{36} = 5 \text{ kHz.}$$

38. A block of

Sol. For maximum speed,

$$\frac{dv}{dt} = 0 \Rightarrow a = 0$$

If at maximum speed, elongation is x, then

$$F = kx \Rightarrow x = \frac{F}{k}$$

Now, by WET,

$$-\frac{1}{2}kx^2 + Fx = \frac{1}{2}mv^2 \quad \therefore$$

$$V = \frac{F}{\sqrt{mk}} = 6 \text{ m/s}$$

39. A uniform

Sol. Let ρ be the density of the material, ρ_0 be the density of water. When the sphere has just started sinking, the weight of the sphere = weight of water displaced

$$\Rightarrow \frac{4}{3} \pi (R^3 - r^3) \rho g = \frac{4}{3} \pi R^3 \rho_0 g$$

$$\Rightarrow (R^3 - r^3) \rho = R^3 \rho_0 \Rightarrow \frac{R^3 - r^3}{R^3} = \frac{\rho_0}{\rho}$$

$$\Rightarrow \frac{r}{R} = \frac{(7)^{1/3}}{2}$$

40. Two sound

Sol. $f_1 = \frac{V}{\lambda}$

$$f_1 = \frac{330}{5} = 66$$

$$f_2 = \frac{330}{5.5} = 60$$

$$\Delta f = f_1 - f_2 = 6$$

CHEMISTRY

5. Which of the following

Sol. Boiling of egg leads to denaturation of proteins which has random coil structure.

6. For the given reaction.....

Sol. $\frac{\Delta H_2 - \Delta H_1}{T_2 - T_1} = (\Delta C_p)_{\text{reaction}} = 2 \times 20 - 20 = 20 \text{ J/K}$

$\therefore (\Delta C_p)_{\text{reaction}}$ is positive on $T \uparrow \Delta H \uparrow$

$$\text{at } 300\text{K } \Delta G = \Delta H - T\Delta S$$



$$= 30 - 300 \times 150 \times 10^{-3}$$

$$= -15 \text{ KJ/mole}$$

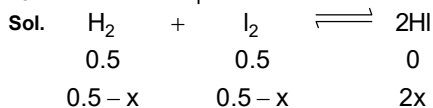
13. A piece of red litmus paper

Sol. BaO_2 and Na_2O_2 are peroxide. Aqueous solution of both give H_2O_2 which litmus paper turns white.

14. Which of the following

Sol. (C)  is more polar than  ($\mu = 0$)
(D) Bond angle in CF_4 & CCl_4 are identical.

16. The value of equilibrium



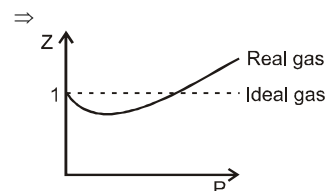
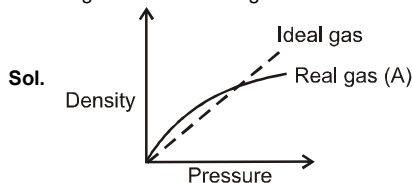
$$K_c = \frac{\left(\frac{2x}{7}\right)^2}{\left(\frac{0.5-x}{7}\right)^2} = \frac{4x^2}{(0.5-x)^2} = 49$$

$$7 = \frac{2x}{0.5-x} \Rightarrow x = \frac{3.5}{9}$$

$$\text{total pressure} = nRT/V = 8.21 \text{ atm}$$

$$\text{concentration of HI} = 2x/V = \frac{1}{9} \text{ M}$$

17. A gas shows following



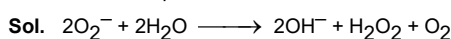
18. For a real gas

Sol. A real gas behaves ideally at Boyle's temperature.

$$Z = \frac{PV_M}{RT} = 1$$

$$P = \frac{RT}{V_M} = \frac{R \frac{a}{Rb}}{V_M} = \frac{a}{V_M \cdot b}$$

20. Which compound will



21. In Column-I there are

Sol. $Z = \frac{PV}{nRT}$ $i \Rightarrow$ ideal, $R \Rightarrow$ real



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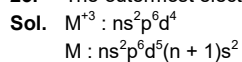
$$Z = \frac{P}{\left(\frac{nRT}{V}\right)} = \frac{P_R}{P_i} \quad (\text{at same } n, T, V)$$

$$Z = \frac{V}{\left(\frac{nRT}{P}\right)} = \frac{V_R}{V_i} \quad (\text{at same } P, T, n)$$

$$Z = \frac{\left(\frac{PV}{RT}\right)}{n} = \frac{n_i}{n_R} \quad (\text{at same } P, V, T)$$

$$Z = \frac{\left(\frac{PV}{nR}\right)}{T} = \frac{T_i}{T_R} \quad (\text{at same } P, V, n)$$

23. The outermost electronic

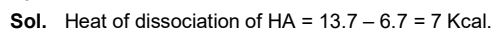


24. A mixture of 32 gm of O_2

Sol. $\left(\frac{r_{D_2}}{r_{O_2}}\right) = \frac{n_{D_2}}{n_{O_2}} \sqrt{\frac{M_{O_2}}{M_{D_2}}}$; $n_{O_2} = 1$; $n_{D_2} = 4$

$$\Rightarrow 4 \sqrt{\frac{32}{4}} = 8\sqrt{2}$$

25. If heat of reaction for.....



26. 1 mole of an ideal gas

Sol. Process is isochoric.

27. 1 mole of an ideal gas

Sol. In isothermal reversible process

$$\Delta S = \frac{q_{rev}}{T}$$

$$q = -w = 2.303 RT \log \frac{V_2}{V_1}$$

$$= 2740.6 \text{ J/mole}$$

$$\Delta S_{sys} = \frac{q_{rev}}{T} = \frac{2740.6}{300} = 9.135 \text{ J/K mole}$$

$$\Delta S_{surr} = -\Delta S_{system} = -9.135 \text{ JK}^{-1} \text{ mole}^{-1}$$

$$\Delta S_{universe} = 0$$

28. Consider the following

Sol. (i), (ii), (iv), (v)

$$\Delta S_{sys} \text{ is } +ve$$

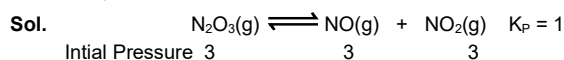
$$\Delta S_{surr} = 0, \Delta H = 0, \Delta U = 0$$

$$\Delta G = \Delta H - \Delta(TS)$$

$$= 0 - T\Delta S$$

$$= -T\Delta S$$

29. N_2O_3 dissociates into



at eq. $3 + P$ $3 - P$ $3 - P$

$$K_P = \frac{(3 - P)^2}{(3 + P)} = 1$$

Solving this $P = 1$ atm

30. How many of the following

Sol. only (3) is correct

31. In how many of the

Sol. (i) $Z = \frac{PM}{dRT} = \frac{10 \times 32}{20 \times \frac{1}{12} \times 300} = \frac{16}{25}$

$\therefore O_2$ shows negative deviation

(ii) $Z = \frac{6}{11.2} \Rightarrow Z < 1 \quad \therefore N_2$ shows negative deviation

(iii) A shows negative deviation at $T = T_C$ and $P < P_C$

(iv) $Z > 1$

(v) $Z > 1$

(vi) $Z < 1$

(vii) $P = \text{low}, T = T_B$

$\therefore Z = 1$ or $PV = nRT$

32. How many species

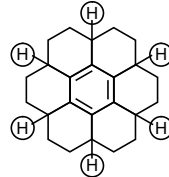
Sol. (i), (ii), (iv), (vii), (viii)

35. $K_2CO_3, Rb_2CO_3, CaCO_3, \dots$

Sol. $K_2CO_3, Na_2CO_3, Rb_2CO_3, CaCO_3, SrCO_3$ & $BaCO_3$ are thermally stable than $MgCO_3$.

37. Total number of

Sol.



$(H) \rightarrow$ Hyperconjugative hydrogen.

38. How many of the

Sol. (a), (b), (c), (d) & (g) are correct.

39. How many of the

Sol. (1) $CsH > KH > NaH > LiH$

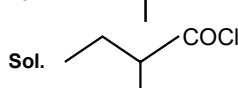
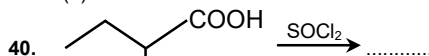
\rightarrow Reducing power

(7) $Li > Na > K > Rb$

\rightarrow Metallic bond strength

(9) $Cs < K < Ca < Be$

\rightarrow Ionisation enthalpy



DATE : 04-02-2018
ANSWER KEY
CODE-0
MATHEMATICS

1. (BC) 2. (ABD) 3. (CD) 4. (ABC) 5. (ABD) 6. (BC) 7. (A)
 8. (D) 9. (A) 10. (D) 11. (AC) 12. (A) 13. (AC) 14. (C)
 15. (D) 16. (A) 17. (A) 18. (B) 19. (A) 20. (B)
 21. (A) → R; (B) → S; (C) → P; (D) → S 22. (A) → S; (B) → P; (C) → S; (D) → Q 23. (7)
 24. (8) 25. (6) 26. (2) 27. (0) 28. (4) 29. (5) 30. (2)
 31. (5) 32. (0) 33. (8) 34. (7) 35. (6) 36. (2) 37. (5)
 38. (4) 39. (0) 40. (0)

PHYSICS

1. (A) 2. (AD) 3. (ABCD) 4. (D) 5. (D) 6. (ACD) 7. (AC)
 8. (AC) 9. (ACD) 10. (BCD) 11. (ABD) 12. (ACD) 13. (ABCD) 14. (ABCD)
 15. (C) 16. (B) 17. (A) 18. (C) 19. (A) 20. (AC)
 21. (A) – Q ; (B) – P ; (C) – Q ; (D) – S
 22. (A) – P, Q ; (B) – Q, R, S ; (C) – P, Q, R, S ; (D) – P, Q, R, S 23. (3) 24. (3)
 25. (3) 26. (2) 27. (6) 28. (7) 29. (9) 30. (6) 31. (7)
 32. (5) 33. (7) 34. (1) 35. (1) 36. (4) 37. (5) 38. (6)
 39. (7) 40. (6)

CHEMISTRY

1. (ABD) 2. (AD) 3. (ABCD) 4. (D) 5. (BD) 6. (AC) 7. (ABC)
 8. (ABD) 9. (A) 10. (BC) 11. (D) 12. (B) 13. (BC) 14. (CD)
 15. (ABCD) 16. (AD) 17. (ABC) 18. (C) 19. (AB) 20. (B)
 21. (A) – Q, S; (B) – P, R; (C) – Q, S; (D) – P, R 22. (A) – P, Q ; (B) – P, Q, R; (C) – P, Q, R; (D) – P, Q, S
 23. (5) 24. (8) 25. (7) 26. (0) 27. (0) 28. (4) 29. (2)
 30. (6) 31. (4) 32. (5) 33. (4) 34. (2) 35. (6) 36. (3)
 37. (6) 38. (5) 39. (6) 40. (5)