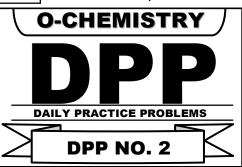
TARGET: NEET (UG) 2024

Course: SARANSH (Youtube Live CRASH COURSE)



**DPP No.: 2** 

1. When excess of ethyl alcohol heated at 140°C with concentrate sulphuric acid the compound that distill is:

(1) C<sub>2</sub>H<sub>5</sub>—O—C<sub>2</sub>H<sub>5</sub>

(2) Ethyl hydrogen sulphate

(3)  $CH_2 = CH_2$ 

(4) di ethyl sulphate

2. Which of the following step is involved in the acid catalysed dehydration of alcohols?

(1) Expulsion of a OH- ion

(2) A free radical intermediate formation

- (3) A carbocation intermediate formation (4) A carbanion intermediate formation
- **3.** Dehydration of alcohol is an example of :

(1) addition reaction

(2) substitution reaction

(3) elimination reaction

(4) rearrangement

- **4.** Thionyl chloride method is preferred over phosphorus pentachloride method for the preparation of alkyl chloride because-
  - (1) The reaction goes to completion.
  - (2) The by-products being gases escape into the atmosphere leaving behind almost pure alkyl chloride.
  - (3) Thionyl chloride is cheap while phosphorus pentachloride is costly.
  - (4) None of the above.
- 5. Primary, secondary and tertiary alcohols are distinguished by

(1) Oxidation method

(2) Lucas test

(3) Victor mayer's test

(4) All of the above

**6.** For the reaction,  $C_2H_5OH + HX \xrightarrow{ZnX_2} C_2H_5X$ , the order of reactivity is :

(1) HI > HCI > HBr

(2) HI > HBr > HCl

(3) HCl > HBr > HI

(4) HBr > HI > HCI

7. Among the following, the one which reacts most readily with ethanol is

(1) p-nitro benzyl bromide

(2) p-chloro benzyl bromide

(3) p-methoxy benzyl bromide

(4) p-methyl benzyl bromide



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8. In which of the following reaction the product obtained is t-butylmethyl ether?

(1) CH<sub>3</sub>OH + HO —CH<sub>2</sub>—CH<sub>3</sub> 
$$\xrightarrow{\text{conc.H}_2SO_4}$$
 (2) CH<sub>3</sub>—C—Br+ CH<sub>3</sub>OH  $\xrightarrow{\text{NaOH}}$  CH<sub>3</sub>

9. NaOH,  $\Delta$  Major product is :

Which is correct option for the above reaction.

- (1) Reaction 2<sup>nd</sup> follows unimolecular mechanism mainly.
- (2) Reactant & product Y have same configuration
- (3) In the reaction  $2^{nd}$  Walden inversion takes place at  $\alpha$  carbon.
- (4) All are correct.

## **ANSWER KEY**

- **1.** (1) **2.** (3) **3.** (3) **4.** (2) **5.** (4) **6.** (2) **7.** (3)
- 8. (3) 9. (2) 10. (3)