SCHOLASTIC APTITUDE TEST PAPER

- A particle completes two revolutions in 50 seconds in a circular path of radius 7m. Distance covered and displacement in two revolutions will be, respectively.
 (1) 44m, 14 m
 (2) 88 m, 44m
 (3) 44m, zero
 (4) 88 m, zero
- 2. Distance time graph of two cars A and B is shown. The ratio of speeds of A and B.



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12.		kg is moving with velocity	15 ms ⁻¹ on a horizonta	I plane work done by the
	force of gravity will be	()		
	(1) zero	(2) 1500 J	(3) 15000 J	(4) 150000 J.
13.		wing infrasonic sound is pro		
	(1) Porpoises	(2) dolphins	(3) elephants	(4) bats
14.	A sound wave has frequency of 4 kHz and wavelength 25 cm. Then distance travelled by sound i			
	2 sec. (in km) will be		(3) 3	(4) 4
	(1) 1	(2) 2	(3) 3	(4) 4
15.	When an image forr objects will be.	med by a concave mirror i	s virtual, erect and enlar	ged, than position of the
	(1) beyond centre of	curvature C		
	(2) in between centre(3) at focus F	e of curvature C and focus I		
	(4) in between pole F	P and focus F.		
16.	If half of a convex ler	e is blackoned, then which	of the following statemen	ts is correct?
10.	If half of a convex lens is blackened, then which of the following statements is correct? (1) Image will not be formed			
	(2) Image formed will be half of size of the object and intensity will be unchanged			
	(3) Image will be formed fully but intensity becomes half(4) Image will be formed fully and intensity will be unchanged			
17.	The far point of a my	opic person is at 100 cm ir	front of the eve. The nat	ure and power of the long
17.	required to correct th		Thom of the eye. The hat	ure and power or the lens
	(1) Concave lens, + 7	1D (2) concave lens, - 1D	(3) convex lens, + 1D	(4) convex lens, - 1D
18.	Which of the followin	g materials is an alloy?		
	(1) Ebonite	(2) Manganese	(3) Manganin	(4) Nickel
19.		vity of the material of condu		ubled and area of cross –
	section is tripled, the (1)	n its electrical resistivity wil (2) 2	l be. (3) 3	(4) 4
20.		equal resistance are conr		
	equivalent resistances in both cases are R_s and R_P respectively, then the value of $\frac{R_s}{R_P}$ will be.			
	(1) 1 : 1	(2) 3 : 1	(3) 9 : 1	(4) 1 : 9
21.	In the following figure	e, the motion of an electror	i is shown in a figure in ur	niform magnetic field. The

direction of force acting an electron will be.



- (1) along positive X axis
 (2) along negative X axis
 (3) Perpendicular to plane of paper, outwards
 (4) perpendicular to plane of paper, inwards.

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22.	Bio- gas contains meth (1) up to 25%	nane gas. (2) up to 50%	(3) up to 12%	(4) up to 75%
23.	The semi- conductor n (1) tungsten	naterial used for making so (2) silicon	lar cell is. (3) copper	(4) aluminium
24.	Which of the following low energy particles is bombarded on the nucleus of heavy atom in nuclear			
	fission? (1) Proton	(2) Deuteron	(3) Alpha particle	(4) Neutron
25.	 Which of the following statements is correct? (1) Inside the bar magnet, the direction of magnetic field lines is from its north pole to south pole. (2) Outside the bar magnet, the direction of magnetic field lines is from south pole to north pole. (3) Magnetic field lines are closed curves (4) Magnetic field lines are open curves. 			
26.		owing is the correct order	of increasing attraction be	etween the particles of
	matter? (1) Water < Hydrogen (3) Salt < Water < Hyd		(2) Hydrogen < Salt < Wa (4) Hydrogen < Water < 3	
27.	At which temperature (1) 265 K	will physical state of water l (2) 298 K	be liquid? (3) 378 K	(4) 398 K
28.	Which one of the follow (1) Butter	ving is an example of emul (2) Fog	sion? (3) Milk of magnesia	(4) Milk
29.	Which is the suitable r in their boiling points? (1) Sublimat <mark>i</mark> on	nethod of separation of two (2) Fractional distillation	o miscible liquids having les (3) Chromatography	ss than 25ºC difference (4) Evaporation
30.	Which one of the follow (1) Rusting of iron	ving is not a chemical chan (2) Digestion of food	ige? (3) Burning of paper	(4) Melting of candle
31.	The correct formula of (1) $Al_2(SO_4)_3$	the compound aluminium s (2) $AI_3(SO_3)_2$	sulphite is. (3) $Al_2(SO_3)_3$	(4) $AI_3(SO_4)_2$
32.	What is the number of (1) 2.40×10^{22}	molecules in 4.25 g of amr (2) 2.40×10^{23}	nonia ? (3) 1.51×10 ²³	(4) 1.51×10 ²²
33.	Which one of the follow (1) Chlorine	ving molecules does not ha (2) Nitrogen	ave atomicity two? (3) Oxygen	(4) Phosphorus
34.	The number of electro (1) 4	ns in the M- Shell of sulphu (2) 5	r atom is. (3) 6	(4) 7
35.	If the ratio of two isoto (1) 16.00 u	pes ¹⁴ ₇ X and ¹⁵ ₇ X of an eler (2) 17.75 u	nent X is 4 : 1, its average (3) 12.84 u	atomic mass will be. (4) 14.20 u.
36.	Which of the following (1) Protium $\begin{bmatrix} 1\\1\\ \end{bmatrix}$	does not possess neutron? (2) Deuterium(² ₁ H)	? (3) Tritium $\begin{pmatrix} 3\\ 1 \end{pmatrix}$	(4) Helium $\begin{pmatrix} 4\\2 \end{pmatrix}$ He
37.	Which substance is ge $ZnO+C \rightarrow Zn+CO$ (1) ZnO	tting reduced in the followin	ng reaction? (3) Zn	(4) CO



■STATE TALENT SEARCH EXAMINATION-2016 | 27-11-2016 Resonance 38. $AgNO_3(aq) + NaCl(aq) \rightarrow AgCl(s) + NaNO_3(aq)$ The reaction given above is which type of the following? (2) Double displacement (3) Dissociation (1) Redox (4) Combination. 39. Which acid is found in ant sting? (1) Acetic acid (2) Oxalic acid (3) Citric acid (4) Methanoic acid The pH values of four solutions A, B, C and D are 12,4,7 and 2 respectively. Which of the following 40. statements is false for these solutions? (1) Solution A is basic (2) Solution B and D are acidic (3) Solution D has minimum concentration of H⁺ ions (4) The concentrations of H^+ and OH^- ions are equal in solution C. Which gas reacts with lime water and turns it milky? 41. (4) Carbon dioxide. (1) Oxygen (2) Chlorine (3) Hydrogen Which of the following metals can be cut easily with a knife? 42. (2) Tin (4) Sodium (1) Iron (3) Zinc Which is the most reactive metal among the following? 43. (1) Calcium (4) Silver (2) Zinc (3) Copper Which one of the following compounds has -C functional group ? 44. (1) Propanol (2) Propanone (3) Propanal (4) Propene 45. What is obtained when ethanol is heated with excess of conc. H₂SO₄ at 443 K temperature? (1) Ethane (2) Ethene (3) Ethyne (4) Mathane 46. Which one of following hydrocarbons will not display addition reaction? (1) $C_2 H_4$ (2) $C_2 H_2$ (3) $C_{3}H_{6}$ (4) $C_{3}H_{8}$ Which one of the following properties does not belong to ionic compounds? 47. (1) Hard and brittle (2) High melting and boiling points (3) Usually water soluble (4) Good conductor of electricity in solid state. 48. If X, Y, and Z are Dobereiner triads and atomic masses of X and Z are 15 and 75, respectively, what will be the atomic mass of Y? (3) 90 (1) 45 (2) 60(4) 30 Which one of the following atoms is biggest in size? 49. (1) B (2) Be (3) O (4) N 50. The element which stands in third period and second group of modern periodic table is. (2) Beryllium (1) Sodium (3) Magnesium (4) Aluminium. 51. The scientist who further expanded cell theory was. (1) Leeuwenhoek (2) Schleiden (4) Purkinje (3) Virchow The pair of organelles which are able to make their own protein is. 52. (1) Mitochondria, Lysosome (2) Ribosome, Lysosome (3) Plastid, Golgi body (4) Plastid, Mitochondria



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53.	The chemical present on cell wall of bark in plants (1) Cutin (2) Suberin	s is. (3) Lignin	(4) Pectin
54.	Correct pair of essential micronutrients for plant g (1) iron, magnesium (2) phosphorus, sulphur	rowth is. (3) potassium, chlorine	(4) zinc, copper
55.	The organism which is more sensitive to the level (1) Algae (2) Fungi	l of sulphur dioxide in the a (3) Lichen	ir is. (4) Bacteria
56.	The hormone responsible for wilting of leaves is.(1) Auxin(2) Cytokinin	(3) Gibberellin	(4) Abscisic acid.
57.	During anaerobic respiration, the chemical built u (1) Pyruvic acid (2) Lactic acid	p in our muscles is. (3) Oxaloacetic acid	(4) Citric acid
58.	The growth of pollen tube towards ovule shows.(1) Hydrotropism(2) Phototropism	(3) Geotropism	(4) Chemotropism
59.	If produce of a terrestrial food chain synthesises energy that reaches to tertiary consumers will be. (1) 1000 k. cal (2) 100 k. cal		nen the quantity of food (4) 1k .cal.
60.	The name of local system of canal irrigation in Hi (1) Tal (2) Kulh	machal Pradesh is? (3) Nadi	(4) Pyne
61.	In animals, hormone secretion control mechanism (1) secretion mechansim (3) hyposecretion mechanism	n is called as. (2) feedback mechanism (4) hypersecretion mecha	
62.	An example of phylum echinodermata is. (1) Leech (2) Housefly	(3) Liver fluke	(4) Starfish
63.	Which of the following diseases diseases is cause (1) Anaemia (3) Leukaemia	ed due to the deficiency of (2) Haemophilia (4) Thrombocytopenia	RBC?
64.	Which of the following is not a sexually transmitter(1) Syphilis(2) Gonorrhoea	ed disease? (3) AIDS	(4) Tuberculosis.
65.	Cell organelle, which has own DNA and ribosome (1) Mitochondria (2) Lysosome	e is. (3) Golgi apparatus	(4) Vacuoles
66.	An example of cold – blooded animal is. (1) Echidna (2) Kangaroo	(3) Pigeon	(4) Snake
67.	Which organ controls involuntary reactions?(1) Cerebellum(2) Cerebrum	(3) Medulla	(4) Forebrain
68.	Spinal chord is protected by the structure. (1) Vertebral column (2) Skull	(3) Sternum	(4) Ribs
69.	Protein digesting enzyme secreted by gastric glar (1) Trypsin (2) Pepsin	nd is. (3) Erepsin	(4) Lipase
70.	The biological process involved in the removal of (1) Respiration(2) Defecation	harmful metabolic waste fr (3) Excretion	rom the body is called. (4) Egestion





79. If the first, second and last terms of an A.P. are a, b and 2a respectively, then its sum is (1) $\frac{ab}{2(b-a)}$ (2) $\frac{ab}{(b-a)}$ (3) $\frac{3ab}{2(b-a)}$ (4) $\frac{2ab}{(b-a)}$

80. The value of $\cos 1^{\circ} . \cos 2^{\circ} \cos 3^{\circ} ... \cos 100^{\circ}$ is. (1) -1 (2) 0 (3) 1 (4) None of these



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81.		• then $\frac{\tan A \cdot \tan B + \tan A \cdot \cot B}{\sin A \cdot \sec B}$	$-\frac{\sin^2 B}{\cos^2 A}$ is equal to.	(4)
	(1) cot ² A	(2) $\cot^2 B$	(3) -tan ² A	(4) $-\cot^2 A$
82.	If $\cos\theta + \cos(1) - 1$	$s^2 \theta = 1$, then $sin^2 \theta + sin^4 \theta$ is eq (2) 0	ual to. (3) 1	(4) None of these
83.		f elevation of the top of a tower s		
	-	stance d towards the foot of the	e tower the angle of eleva	ation is found to be β . The
	height of the		h	d
	(1) $\frac{d}{\cot \alpha + \cot \alpha}$	$\frac{d}{\cot\beta}$ (2) $\frac{d}{\cot\alpha - \cot\beta}$	(3) $\frac{d}{\tan\beta - \tan\alpha}$	(4) $\frac{d}{\tan\beta + \tan\alpha}$
84.		e between the given points (cos	, , , ,	
	(1) √3	(2) $\sqrt{2}$	(3) 2	(4) 1
85.		of a sphere is increased by 10	0%, then how much per c	ent does the volume of the
	sphere increation (1) 33.00%	ase? (2) 33.10%	(3) 33.11%	(4) 33.12%
86.	In a ∆ABC ,	$\angle A = 90^{\circ}, AB = 5 \text{ cm and } AC =$	12 cm. If AD \perp BC, then A	AD is equal to.
	(1) $\frac{13}{2}$ cm	(2) $\frac{60}{13}$ cm	(3) $\frac{13}{60}$ cm	(4) $\frac{2}{13}$ cm
87.		figure if AD, AE and BC are ta	angents to the circle at D	, E and F respectively then
	which statem	nent is true?	D	
			<u>c</u>	
		A	E	

(1) AD = AB + BC + CA (2) 2AD = AB + BC + CA (3) 3AD = AB + BC + CA (4)4AD=AB+BC+ CA

Е

B

- If a wire is bent in the shape of a square, then the area of this square is 81 cm². If the wire is bent 88. into a semicircular shape, then the area of the semicircle is equal to. (4) 154 cm² (1) 22 cm² (2) 44 cm^2 (3) 77 cm²
- 89. A bucket, in the form of a frustum of a cone holds 28.49 litres of water. The radii of the top and bottom are 28 cm and 21 cm respectively. Then the height of the bucket is. (4) 10 cm (1) 15 cm (2) 18 cm (3) 20 cm
- 90. Pythagoras was the student of. (1) Thales (2) Euclid

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(3) Thales and Euclid both(4) None of these



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