SCHOLASTIC APTITUDE TEST (SAT)_HINTS & SOLUTIONS

15. In M₂O₃ valency of metal M⁺³ So the formula of metal nitride will be M_{\bullet}^{+3}

$$\int M_3 N_3 \Rightarrow MN$$

Ans. (2) is correct

16. Alloys are the homogeneous mixture of metals or metals & non metals Brass is an alloy of copper and zinc

Brass \Rightarrow Cu Zn

70% 30%
Alloys are solid-solid solution
Ans. (3) is correct

17. Given data mass of glucose = 1.80 gmMolecular mass of glucose ($C_6H_{12}O_6$) = 180 gm

> no. of moles = $\frac{1.80}{180}$ = 10^{-2} moles Number of molecules of glucose = $10^{-2} \times 6.023 \times 10^{23}$ = 6.023×10^{21} molecules of C₆H₁₂O₆ Total number of oxygen atoms present in glucose = $6 \times 6.023 \times 10^{21}$

Number of moles of water $=\frac{36g}{18g}=2$ moles

Number of H₂O molecules = $2 \times 6.023 \times 10^{23}$ H₂O molecules so no. of oxygen atoms in water = 12.046×10^{23} Total number of oxygen atoms present in solution i.e.(water + glucose) $12.46 \times 10^{23} + 36.138 \times 10^{21}$ $\Rightarrow 12.046 \times 10^{23} + 0.36 \times 10^{23}$ = 12.406×10^{23} oxygen atoms **Ans. (1) is correct**

18. Given data ${}^{35}Cl \rightarrow 25\% \rightarrow x_1$ ${}^{37}Cl \rightarrow 75\% \rightarrow x_2$

 $\Rightarrow \text{ average atomic mass} = \frac{p_1 x_1 + p_2 x_2}{p_1 + p_2}$ $= \frac{35 \times 25 + 37 \times 75}{100}$ $\Rightarrow 36.5 \text{ u}$ Ans. (4) is correct

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- A,E,G are alkali metal they always form ionic compound with 16 & 17 group member
 J is alkaline earth metals & they form ionic compound with 16 & 17 group member
 16th group elements (R & Q) form covalent compounds
 Ans. (1) is correct
- 24. When acid reacts with metal carbonate or bicarbonates gives CO_2 gas which is colourless and odourless gas & turn lime water milky. Metal carbonate + $CH_3COOH \rightarrow Salt + CO_2$

 $\begin{array}{c} CH_{3}COOH + CH_{3}OH & \xrightarrow{Conc.H_{2}SO_{4}} & CH_{3}COOCH_{3} + H_{2}O \\ (X) & ester \\ & (Methyl \ ethanoate, \ sweet \ smell) \end{array}$

Ans. (2) is correct $(C_2H_4O_2 i.e. CH_3COOH)$

(x)

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- 26. According to reactivity series Zn is more reactive than Ag so it can displace Ag. Ans. (2) is correct.
- 27. Due to latent heat, temperature. remains constant during state conversion, In case of water temperature. remains constant at 0°C (m.p.) due to latent heat of fusion and also remains constant at 100°C due to latent heat of vaporisation Ans (3) is correct
- 28. Velocity

During OA velocity is uniform
 ∴ acceleration is constant and positive
 During AB velocity is constant
 ∴ acceleration is zero
 During BC velocity decreasing uniformly acceleration is constant again, and negative.

29. u = - 2m

$$V = -0.4m$$

$$\frac{1}{f} = \frac{1}{v} - \frac{1}{u}$$

$$P = \frac{1}{f} = -\frac{1}{0.4} - \frac{1}{(-2)}$$

$$= -\frac{1}{0.4} + \frac{1}{(-2)}$$

$$= -\frac{10}{4} + \frac{1}{2}$$

$$= \frac{-10 + 2}{4} = \frac{-8}{4} = -2$$

$$P = -2 D$$
That means conceive loss of

That means concave lens of power 2D.

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V = u - gtO = 5 - 10t5 = 10t

$$\frac{5}{10} = 0.5 = t$$

So the time taken to reach from A to B is 1.5 - 0.5 = 1 sec.

$$H = ut + \frac{1}{2}gt^2$$

$$H = \frac{1}{2} \times 10 \times 1$$

H = 5 m (Height between A & B) Height Covered when the ball rebound

S₁ = ut -
$$\frac{1}{2}$$
gt²
S₁ = 5 × 0.5 - $\frac{1}{2}$ × 10 × (0.5)² = 1.25 m
∴ Net displacement = 5 - 1.25 = 3.75 m

31. Since Force is equal to rate of change of momentum

$$F = \frac{\Delta M v}{\Delta t} = v \left(\frac{\Delta M}{\Delta t} \right)$$

$$F = v \left(\frac{Volume \times density}{\Delta t} \right)$$

$$F = v \left(\frac{A \times \ell \times \rho}{\Delta t} \right)$$

$$F = v \left(\frac{A \times v \times \Delta t \times \rho}{\Delta t} \right)$$

$$F = v^{2} \rho A \qquad (Hint : Area = \pi r^{2} = \pi \left(\frac{D}{2} \right)^{2})$$

$$F = \frac{\pi v^{2} \rho d^{2}}{4}$$

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35. Since sound is traveling from water to air.

Therefore frequency remains constant it depends on source of sound. Now $v = f\lambda$

If v will decrease then λ will also decrease as f is constant.

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36.	Initial volume = Final volume)	
	$A_1 L_1 = A_2 L_2$		
	$\pi \mathbf{r_1}^2 \ell_1 = \pi \mathbf{r_2}^2 \times \ell_2$		
	$\frac{r_{1}^{2}}{r_{2}^{2}} = \frac{\ell_{2}}{\ell_{1}}$		
	$\frac{R_1}{R_2} = \frac{\rho\ell_1}{A_1} \times \frac{A_2}{\rho\ell_2}$		
	$\frac{R_{1}}{R_{2}} = \frac{\ell_{1}}{\ell_{2}} \times \frac{\pi r_{2}^{2}}{\pi r_{1}^{2}}$		
	$\frac{R_1}{R_2} = \frac{r_2^2}{r_1^2} \times \frac{r_2^2}{r_1^2}$		
	$\frac{R_1}{R_2} = \left(\frac{r_2}{r_1}\right)^4$		
	Now, $D_2 = \frac{D_1}{5}$		
	$\mathbf{r}_2 = \frac{\mathbf{r}_1}{5} .$		
	so $\frac{R_1}{R_2} = \left(\frac{r_1}{5}\right)^4$		
	$\frac{R_1}{R_2} = \frac{1}{625}$		
	$625R_1 = R_2$		

37. According to law of conservation of momentum.

mu = (M + m) v $v = \frac{mu}{M + m}$

Loss of kinetic energy = $\frac{1}{2}mu^2 - \frac{1}{2}(m + M)v^2$. = $\frac{1}{2}mu^2 - \frac{1}{2}(m + M)\left(\frac{mu}{M+m}\right)^2$ = $\frac{1}{2}mu^2\left[1 - \frac{m}{M+m}\right]$ = $\frac{1}{2}mu^2 \times \frac{M}{M+m}$ = $\frac{1}{2}\frac{Mmu^2}{(M+m)}$.



magnetic field around PQ more than RS. So force on the arms PQ and RS will be unequal and opposite.

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Resonance

Cube of a natural number is of the form 9 m, 9m + 1, 9m + 8 41. So, possible remainder are 0, 1, 8 Sum = 0 + 1 + 8 = 9.Option (4). 42. Given P(1) = 3P(3) = 5Let r(x) = ax + bSo P(x) = (x - 1) (x - 3) q(x) + ax + b $P(1) = a + b \Rightarrow a + b = 3$ $\mathsf{P(3)} = 3\mathsf{a} + \mathsf{b} \Longrightarrow 3\mathsf{a} + \mathsf{b} = 5$ r(x) = ax + b = x + 22a = 2 r(-2) = (-2) + 2a = 1 b = 2 r(-2) = 0Option (3). 43. px + 3y - (p - 3) = 012x + py - p = 0For infinitely many solution = -3 $= - \frac{p-3}{2}$ р 12 p р From I and II I and II $P^2 = 36$ p = 0 $P = \pm 6$ p = 6 So p = 6 Option (1). $\alpha^2 - b\alpha + 6 = 0$ 44. $\alpha^2 - 6\alpha + c = 0$ $\overline{(6-b)\alpha+6-c=0}$ $\alpha = \frac{c-6}{6-b}$ $\alpha \cdot 3a = 6 \Rightarrow a = \frac{2}{\alpha}$ α + 4a = 6 $\alpha + \frac{4 \times 2}{\alpha} = 6.$ $\alpha + \frac{8}{\alpha} = 6$ $\alpha^2 + 8 - 6\alpha = 0$ $\alpha^2 - 6\alpha + 8 = 0$ $(\alpha - 4) (\alpha - 2) = 0$ $\alpha = 4, 2.$

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53.	Let the side are x, 2x, 3x $(x, y) = 2^{3}$
	$V = (X) (2X) (3X) = 6X^{-1}$ ATQ $6x^{3} < 400$
	$x^{3} < 66^{\frac{2}{3}}$
	$\lambda \ge 0.0\frac{1}{3}$
	\therefore x = 64. maximum number of cubes = 6x ³ = 6 × 64 = 384.
54.	1, 2, 3, 18 19, 20, 21.
	4, 5, 6, 14, 15, 16, 17
	7, 8, 9, 10, 11, 12, 13
	18+14+10 42
	Mean = $\frac{3}{3} = \frac{3}{3} = 14.$
55	N ₄ = 2272
00.	$N_2 = 875$
	when N_1 and N_2 divide by N give remainder r.
	$\therefore \qquad (N_1 - r) - (N_2 - r) \text{ is also divisible by N.}$
	(2272 - r) - (875 - r)
	N 1207
	$\frac{1397}{N \rightarrow 11,127}$
	∴ N = 127
	Sum of digit = $1 + 2 + 7 = 10$.
	Ę
50	Α Θ Β
56.	90 - 0 J
	X
	$\sin(90, \theta) = \frac{BE}{2}$
	$\cos\theta = \frac{\Delta E}{1} = \frac{4}{5}$
	$BE = \frac{4}{2}$
	5
	$AB = \sqrt{1 - (\frac{4}{2})^2}$
	γ (5)
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<i>/ \</i>	Educating for botton to marrie	

Luuca	
	$= \sqrt{\frac{9}{25}} = \frac{3}{5}$ $E\left(\frac{3}{5}, -\frac{4}{5}\right)$
57.	$P(x) = x^{2} + 5kx + k^{2} + 5$ $P(-2) = 0$ $4 - 10k + k^{2} + 5 = 0$ $k^{2} - 10k + 9 = 0$ $k = 9, 1$ (1)
	But $P(-3) \neq 0$ $9 - 15k + k^2 + 5 \neq 0$ $k^2 - 15k + 14 \neq 0$ $(k - 14) (k - 1) \neq 0$ $k \neq 14, 1$ (2) from (1) and (2). k = 9.
58.	$cos^{4}\theta + sin^{2}\theta = m$ $(1 - sin^{2}\theta)^{2} + sin^{2}\theta = m$ $sin^{4}\theta - sin^{2}\theta + 1 = m$ $\left(sin^{2}\theta - \frac{1}{2}\right)^{2} + \frac{3}{4} = m$
	$m_{min} \text{ when } \sin^2 \theta - \frac{1}{2} = 0$ $\therefore \qquad m_{min} = \frac{3}{4}$ $m_{max} \text{ when } \sin^2 \theta = 0$
	$\therefore \qquad m_{max} = \left(0 - \frac{1}{2}\right)^2 + \frac{3}{4} \\ = \frac{1}{4} + \frac{3}{4} = 1.$
59.	$2A + 3B + 1C = 26 \qquad \dots \dots (1)$ $3A + 2B + 2C = 35 \qquad \dots \dots (2)$ $(1) \times 3 + (2) \times 2$ We get $12A + 13B + 7C = 26 \times 3 + 35 \times 2$ = 78 + 70. = 148.

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