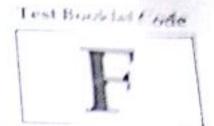
This booklet contribe 24 printed pages.

# PARER - 1: CHEMISTRY, MATHEMATICS & PHYSICS

Do not open this Test Booklet until you are anked to do so.

Read carefully the Instructions on the Back Cover of this Test Booklet.



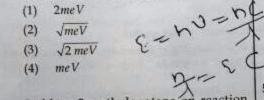
Important Instructions :

- Immediately fill in the particulars on this page of the Test Booklet with only Blue / Bluek Bull Point Pen provided by the Board.
- The Answer Sheet is kept inside this Test Booklet. When you are directed to open few fest Booklet, take out the Answer Sheet and fill in the particulars carefully.
- The test is of 3 hours duration.
- The Test Booklet consists of 90 questions. The maximum marks are 360. 3.
- There are three parts in the question paper A, B, C consisting of Chemistry, Mathematics and 4. Physics having 30 questions in each part of equal weightage. Each question is afforted 4 (four) marks for each correct response.
- Candidates will be awarded marks as stated above in instruction No. 5 for correct response of each question. 1/4 (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
- There is only one correct response for each question. Filling up more than one response in each question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction 6 above.
- For writing particulars/marking responses on Side-1 and Side-2 of the Answer Sheet use only Blue/Black Ball Point Pen provided by the Board.
- No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. except the Admit Card inside the examination room/hall.
- 10. Rough work is to be done on the space provided for this purpose in the Test Booklet only. This space is given at the bottom of each page and in three pages (Pages 21 - 23) at the end of the booklet.
- 11. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty in the Room/Hall. However, the candidates are allowed to take away this Test Booklet with them.
- 12. The CODE for this Booklet is F. Make sure that the CODE printed on Side-2 of the Answer Sheet and also tally the serial number of the Test Booklet and Answer Sheet are the same as that on this booklet. In case of discrepancy, the candidate should immediately report the matter to the Invigilator for replacement of both the Test Booklet and the Answer Sheet.

13. Do not f	ld or make any stray mark on the Answer Sheet.
Name of the Cat	lidate (in Capital letters): ROSELYH LORSON
Roll Number	: in figures 17107938 : in words Due crove seventy one laker seven temperal nin
Examination Cer	IMINANDA MALINA
Name of Examin	tion Centre (in Capital letters): <u>CATHAYE</u> COLLEGE
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	2. Invigilator's Signature:

## PART A - CHEMISTRY

- A stream of electrons from a heated 1. filament was passed between two charged plates kept at a potential difference V esu. If e and m are charge and mass of an electron, respectively, then the value of  $h/\lambda$  (where  $\lambda$  is wavelength associated with electron wave) is given by :
  - (1) 2meV
  - (2) JmeV.



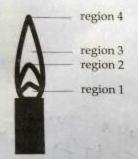
- 2-chloro-2-methylpentane on reaction 2. with sodium methoxide in methanol vields:
  - C2H5CH2C-OCH3
  - C2H5CH2C=CH2
  - C2H5CH=C-CH3
  - (a) and (c) (c) only
    - (a) and (b)
    - All of these (4)
- Which of the following compounds is 3. metallic and ferromagnetic?
  - CrO2 (1)
  - VO2 (2)
  - MnO2 (3)
  - TiO2 (4)

- Which of the following statements about low density polythene is FALSE?
  - It is a poor conductor of electricity. (1)
  - Its synthesis requires dioxygen or a (2) peroxide initiator as a catalyst.
  - It is used in the manufacture of (3) buckets, dust-bins etc.
  - Its synthesis requires high pressure. (4)

For a linear plot of  $\log (x/m)$  versus  $\log p$ 5. in a Freundlich adsorption isotherm, which of the following statements is correct? (k and n are constants)

- 1/n appears as the intercept. (1)
- Only 1/n appears as the slope. (2)
- $\log (1/n)$  appears as the intercept. (3)
- Both k and 1/n appear in the slope (4) term.
- The heats of combustion of carbon and 6. carbon monoxide are -393.5 and -283.5 kJ mol-1, respectively. The heat of formation (in kJ) of carbon monoxide per mole is:
  - (1) 676.5
  - (2)-676.5
  - -110.5
  - 110.5 (4)

7. The hottest region of Bunsen flame shown in the figure below is:



- (1) region 2
- (2) region 3 region 4

(4) region 1 CB C-C-C

- 8. Which of the oflowing is an anionic detergent?
  - (1) Sodium lauryl sulphate
  - (2) Cetyltrimethyl ammonium bromide

(3) Glycery pleate

(4) Sodium stearate C2

- 9. 18 g glucose  $H_{12}O_6$ ) is added to 178.2 g water. The vapor pressure of water (in torr) for this aqueous solution is:
  - (1) 76.0
  - (2) 752.4
  - (3) 759.0
  - (4) 7.6

- 10. The distillation technique most suited for separating glycerol from spent-lye in the soap industry is:
  - (1) Fractional distillation
  - (2) Steam distillation
  - (3) Distillation under reduced pressure
  - (4) Simple distillation

58

11. The species in which the N atom is in a state of sp hybridization is:

NO<sub>2</sub>

- N-0
- (3) NO2 X

NO<sub>3</sub>

4) NO2 X 3+

3+4 9-1

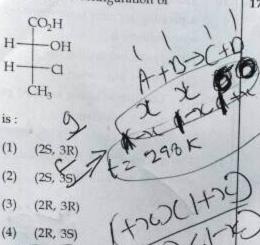
- 12. Decomposition of H<sub>2</sub>O<sub>2</sub> follows a first order reaction. In fifty minutes the concentration of H<sub>2</sub>O<sub>2</sub> decreases from 0.5 to 0.125 M in one such decomposition. When the concentration of H<sub>2</sub>O<sub>2</sub> reaches 0.05 M, the rate of formation of O<sub>2</sub> will be:
  - (1) 6.93×10<sup>-4</sup> mol min<sup>-1</sup> (2) 2.66 L min<sup>-1</sup> at STP
  - (3) 1.34×10<sup>-2</sup> mol min<sup>-1</sup>
  - (4) 6.93×10<sup>-2</sup> mol min<sup>-1</sup>
- 7/20
- 13. The pair having the same magnetic moment is:

[At. No. : Cr = 24, Mn = 25, Fe = 26, Co = 27]

- (1)  $[Cr(H_2O)_6]^{2+}$  and  $[Fe(H_2O)_6]^{2+}$
- (2)  $[Mn(H_2O)_6]^{2+}$  and  $[Cr(H_2O)_6]^{2+}$
- (3) [CoCl<sub>4</sub>]<sup>2-</sup> and [Fe(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup>
- (4) [Cr(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup> and [CoCl<sub>4</sub>]<sup>2-</sup>

pater> 17.

The absolute configuration of



15. The equilibrium constant at 298 K for a reaction A + B = C + D is 100. If the initial concentration of all the four species were 1 M each, then equilibrium concentration of D (in mol L-1) will be:

Keg =100 1.818 (2)1.182 ~ = (C) CD)

(3) (4)

Which one of the following ores is best 16. concentrated by froth floatation method?

- Siderite (1)
- Galena (2)
- Malachite (3)
- Magnetite (4)

At 300 K and 1 atm, 15 mL of a gaseous hydrocarbon requires 375 mL air containing 20% O2 by volume for complete combustion. After combustion the gases occupy 330 mL. Assuming that the water formed is in liquid form and the volumes were measured at the same temperature and pressure, the formula of the C3H/60x -10022 C4H8 = 1+2x+x hydrocarbon is:

- (2)

18.

The pair in which phosphorous atoms

- have a formal oxidation state of +3 is: Pyrophosphorous and hypophosphoric acids
- (2) Orthophosphorous and hypophosphoric acids
- (3) Pyrophosphorous and pyrophosphoric acids
- Orthophosphorous and (4) pyrophosphorous acids

Which one of the following complexes 19. shows optical isomerism?

- cis[Co(en)2Cl2]Cl (1)
- trans[Co(en),Cl2]CI

[Co(NH<sub>1</sub>),Cl<sub>2</sub>]Cl ×

[Co(NH3)3Cl3] y

(en = ethylenediaming)

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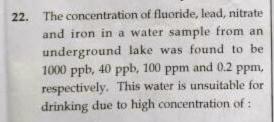
	The reaction of zinc with dilute and
	concentrated nitric acid, respectively,
	produces:

- (1) NO2 and NO
- (2) NO and  $N_2O$  103-99(3)  $NO_2$  and  $N_2O$  3=O
- (4) N<sub>2</sub>O and NO<sub>2</sub>

#### Which one of the following statements 21. about water is FALSE?

- Water can act both as an acid and as a base.
- There is extensive intramolecular (2) hydrogen bonding in the condensed phase.
- Ice formed by heavy water sinks in (3) normal water.

Water is oxidized to oxygen during photosynthesis.



- (1) Lead
- Nitrate (2)
- Iron (3)
- Fluoride (4)

- The main oxides formed on combustion of Li, Na and K in excess of air are, respectively:
  - LiO2, Na2O2 and K2O
  - (2) Li<sub>2</sub>O<sub>2</sub>, Na<sub>2</sub>O<sub>2</sub> and KO<sub>2</sub>
  - (3) Li<sub>2</sub>O, Na<sub>2</sub>O<sub>2</sub> and KO<sub>2</sub>
  - (4) Li<sub>2</sub>O, Na<sub>2</sub>O and KO<sub>2</sub>

#### Thiol group is present in: 24.

- (1) Cystine
- Cysteine (2)
- Methionine (3)
- Cytosine (4)

#### Galvanization is applying a coating of : 25.

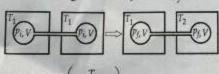
- Cr

Pb

Which of the following atoms has the 26. highest first ionization energy?

- (1) Na
- (2) K
- Sc (3)
- Rb (4)

- 27. In the Hofmann bromamide degradation reaction, the number of moles of NaOH and Br<sub>2</sub> used per mole of amine produced are:
  - Four moles of NaOH and two moles of Br<sub>2</sub>.
  - Two moles of NaOH and two moles of Br<sub>2</sub>.
  - (3) Four moles of NaOH and one mole of Br<sub>2</sub>.
  - One mole of NaOH and one mole of Br<sub>2</sub>.
- 28. Two closed bulbs of equal volume (V) containing an ideal gas initially at pressure p<sub>i</sub> and temperature T<sub>1</sub> are connected through a narrow tube of negligible volume as shown in the figure below. The temperature of one of the bulbs is then raised to T<sub>2</sub>. The final pressure p<sub>i</sub> is:



- $(1) 2p_i \left(\frac{T_1}{T_1 + T_2}\right)$
- $(2) 2p_1 \left( \frac{T_2}{T_1 + T_2} \right)$
- $(3) \quad 2p_i \left( \frac{T_1 T_2}{T_1 + T_2} \right)$
- $(4) p_i \left( \frac{T_1 T_2}{T_1 + T_2} \right)$

- 29. The reaction of propene with HOCI (Cl<sub>2</sub>+H<sub>2</sub>O) proceeds through the intermediate:
  - CH<sub>3</sub>-CH<sup>+</sup>-CH<sub>2</sub>-CI
  - (2) CH<sub>3</sub>-CH(OH)-CH<sub>2</sub><sup>+</sup>
  - (3) CH<sub>3</sub>-CHCl-CH<sub>2</sub><sup>+</sup>
  - (4) CH3-CH+-CH2-OH
- 30. The product of the reaction given below is:

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2(-5)+3

## PART B - MATHEMATICS

 Two sides of a rhombus are along the lines, x-y+1=0 and 7x-y-5=0. If its

(2)  $\left(\frac{1}{3}, -\frac{8}{3}\right)$  (2) (2) (2) (3) (3) (3) (3) (3) (4) (4) (4) (5) (5) (5) (6) (7) (7) (7) (8) (9) (9) (1) (1) (2) (2) (2) (3) (3) (3) (4) (4) (4) (5) (5) (5) (6) (7) (7) (7) (7) (8) (9) (9) (1) (1) (2) (2) (2) (2) (3) (3) (4) (4) (4) (4) (4) (5) (5) (6) (6) (7) (7) (7) (7) (7) (8) (9) (9) (9) (1) (1) (1) (2) (3) (4)

- (3)  $\left(-\frac{10}{3}, -\frac{7}{2}\right)$
- (4) (-3, -9) 21-1-9-2
- If the 2nd, 5th and 9th terms of a 32. non-constant A.P. are in G.P., then the common ratio of this G.P. is:

  - (2)

  - (4)

Let P be the point on the parabola,  $y^2 = 8x$ 33. which is at a minimum distance from the centre C of the circle,  $x^2 + (y+6)^2 = 1$ . Then the equation of the circle, passing through C and having its centre at P is:

The system of linear equations

4nc-4= 2y-4 471-28 = 0 x+y-12=0 - 4=0 2(-1)+(8)

has a non-trivial solution for:

- exactly one value of  $\lambda$ . 2(-5)-(-9)
- exactly two values of  $\lambda$ .

--6+9

- exactly three values of  $\lambda$ . (3)
- infinitely many values of A. (4)

35. If 
$$f(x) + 2f(\frac{1}{x}) = 3x$$
,  $x \neq 0$ , and

 $S = \{x \in \mathbb{R} : f(x) = f(-x)\}; \text{ then } S:$ 

- contains exactly one element.
- (2) contains exactly two elements.
- (3) contains more than two elements.
- (4) is an empty set. +25(-15)

36. Let 
$$p = \lim_{x \to 0+} (1 + \tan^2 \sqrt{x})^{\frac{1}{2x}}$$
 then  $\log p$  is equal to:

- (1) 1 (+ the se) to
- (2) ½ (+nc) tn
  - $\frac{4}{(4)} = mch^{-1}$

37. A value of 
$$\theta$$
 for which  $\frac{2+3i \sin \theta}{1-2i \sin \theta}$  is purely imaginary, is:

- $\sim^{(1)} \frac{\pi}{6} \times 2 + 8 i \text{ sim} \theta$ 

  - (3)  $\sin^{-1}\left(\frac{1}{\sqrt{3}}\right)\chi$   $\sqrt{-2i}$   $\frac{1}{2}$

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- 38. The eccentricity of the hyperbola whose length of the latus rectum is equal to 8 and the length of its conjugate axis is equal to half of the distance between its principle.

  - $\frac{2}{(2)} \frac{2}{\sqrt{3}}$
- 39. If the standard deviation of the numbers 2, 3, a and 11 is 3.5, then which of the following is true?
  - $(1) \quad 3a^2 32a + 84 = 0$ 
    - (2)  $3a^2 34a + 91 = 0$  + 9i = 0 + 9i = 0 + 9i = 0 + 9i = 0 + 9i = 0
    - (3)  $3a^2 23a + 44 = 0$
  - (4) 3a2-26a+150 20
- SPACE FOR ROUGH WORK

  2+3i5
  - -215
- 2+14

40. The integral 
$$\int \frac{2x^{12} + 5x^9}{(x^5 + (x^3)^3)^3} dx$$
 is equal

(1) 
$$\frac{x^{10}}{2(x^5+x^3+1)^2} + C$$

(2) 
$$\frac{x^5}{2(x^5+x^3+1)^2} + C$$

(3) 
$$\frac{-x^{10}}{2(x^5+x^3+1)^2} + C$$

(4) 
$$\frac{-x^5}{\left(x^5 + x^3 + 1\right)^2} + C$$

where C is an arbitrary constant.

- If the line,  $\frac{x-3}{2} = \frac{y+2}{-1} = \frac{z+4}{3}$  lies in 41. the plane, lx + my - z = 9, then  $l^2 + m^2$  is equal to:
  - 18 (1)
  - 5 (2)
  - (3)
  - 26 (4)

If  $0 \le x \le 2\pi$ , then the number of real values of x, which satisfy the equation  $\cos x + \cos 2x + \cos 3x + \cos 4x = 0$ , is:

$$\cos x + \cos 2x + \cos 3x + \cos 4x = 0, \text{ is :}$$

$$(3 + 1)^{3} \quad \text{State}$$

$$(1) \quad 5 \quad \text{State}$$

$$(2) \quad 7 \quad \text{State}$$

$$(3) \quad 9 \quad 5 \quad \text{State}$$

$$(4) \quad 3 \quad \text{State}$$

- The area (in sq. units) of the region 43.  $\{(x, y): y^2 \ge 2x \text{ and } x^2 + y^2 \le 4x, x \ge 0, y \ge 0\}$

$$(1) \quad \pi - \frac{8}{3}$$

- (2)  $\pi \frac{4\sqrt{2}}{3}$
- (3)  $\frac{\pi}{2} \frac{2\sqrt{2}}{3}$
- (4)  $\pi \frac{4}{3}$
- Let  $\overrightarrow{a}$ ,  $\overrightarrow{b}$  and  $\overrightarrow{c}$  be three unit vectors such that  $\overrightarrow{a} \times \left(\overrightarrow{b} \times \overrightarrow{c}\right) = \frac{\sqrt{3}}{2} \left(\overrightarrow{b} + \overrightarrow{c}\right)$ . If  $\overrightarrow{b}$  is not parallel to  $\overrightarrow{c}$  , then the angle between  $\overrightarrow{a}$  and  $\overrightarrow{b}$  is:
  - (1)  $\frac{\pi}{2}$
  - (2)  $\frac{2\pi}{3}$
  - (3)  $\frac{5\pi}{6}$
  - (4)

- A wire of length 2 units is cut into two parts which are bent respectively to form a square of side=x units and a circle of radius = r units. If the sum of the areas of the square and the circle so formed is minimum, then:
- (2) x=2r 2 Units (3) 2x=r x=1
- (4)  $2x = (\pi + 4)r$
- The distance of the point (1, -5, 9) from the plane x-y+z=5 measured along the line x = y = z is:
- 10√3 (1)

16.

47.

- (2)
- (3)
- $3\sqrt{10}$ (4)
- If a curve y = f(x) passes through the point (1, -1) and satisfies the differential equation, y(1+xy) dx = x dy, then  $f\left(-\frac{1}{2}\right)$
- is equal to:
  - (1)
  - (2)  $\frac{2}{5}$
  - (3)
- (4)

- If the number of terms in the expansion of 48.  $\left(1-\frac{2}{x}+\frac{4}{x^2}\right)^{3/6}$ ,  $x \neq 0$ , is 28, then the sum of the coefficients of all the terms in this
  - 2187

expansion, is:

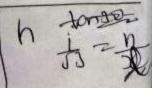
- 243 (2)
- (3) 729
- (4) 64
- Consider 49.

$$f(x) = \tan^{-1}\left(\sqrt{\frac{1+\sin x}{1-\sin x}}\right), \ x \in \left(0, \frac{\pi}{2}\right).$$

A normal to y = f(x) at  $x = \frac{\pi}{6}$  also passes through the point:

- (1)  $\left(0, \frac{2\pi}{3}\right)$
- (2)  $\left(\frac{\pi}{6}, 0\right)$
- (3)  $\left(\frac{\pi}{4}, 0\right)$
- (0, 0)(4)
- For  $x \in \mathbb{R}$ ,  $f(x) = |\log 2 \sin x|$  and 50. g(x) = f(f(x)), then:
  - $g'(0) = \cos(\log 2)$
  - $g'(0) = -\cos(\log 2)$
  - g is differentiable at x=0 and (3) $g'(0) = -\sin(\log 2)$
  - g is not differentiable at x = 0





- Let two fair six-faced dice A and B be thrown simultaneously. If  $E_1$  is the event that die A shows up four,  $E_2$  is the event that die B shows up two and  $E_3$  is the event that the sum of numbers on both dice is odd, then which of the following statements is NOT true?
  - (1)  $E_2$  and  $E_3$  are independent.
  - $E_1$  and  $E_3$  are independent. (2)
  - $E_1$ ,  $E_2$  and  $E_3$  are independent. (3)
  - $E_1$  and  $E_2$  are independent. (4)
- 52. If  $A = \begin{bmatrix} 5a & -b \\ 3 & 2 \end{bmatrix}$  and A adj  $A = A A^T$ , then 5a + b is equal to:
  - (1)
  - (2)
  - (3)
  - (4)
- The Boolean Expression (p^-q) 53. is equivalent to:
  - (1)
  - (2)

  - (4)
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The sum of all real values of x satisfying 54. the equation

 $(x^2-5x+5)^{x^2+4x-60}$   $\ge 1$  is: 41-60 LOJ 602-5-13

- (2)

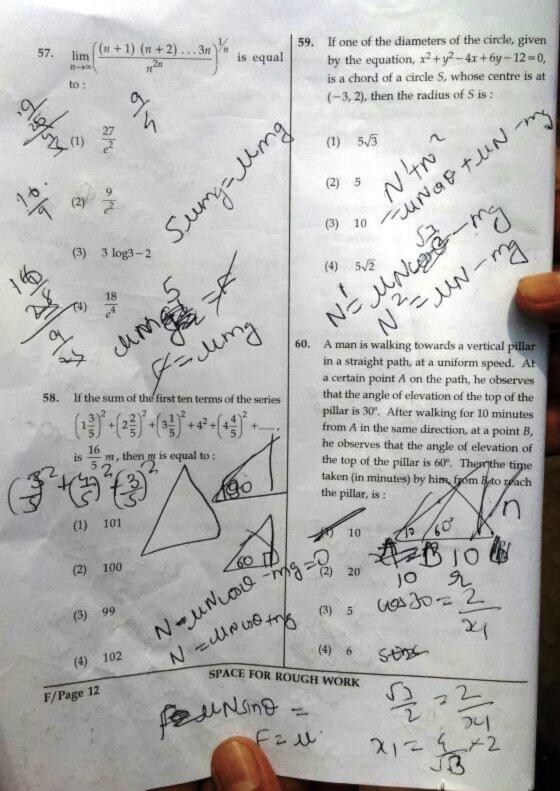
- The centres of those circles which to 55.  $x^2 + y^2 - 8x - 8y - 4 = 0$ circle, externally and also touch the x-axis, lie WA- 25-5 on:
  - an ellipse which is not a circle. (1)
  - a hyperbola. (2)
  - a parabola.
  - (4)a circle.



- 56. If all the words (with or without meaning) having five letters, formed using the letters of the word SMALL and arranged as in a dictionary; then the position of the word SMALL is:
  - 59th
  - 52nd (2)

  - 46th

SPACE FOR ROUGH WORK



## PART C - PHYSICS

# ALL THE GRAPHS GIVEN ARE SCHEMATIC AND NOT DRAWN TO SCALE.

A uniform string of length 20 m is suspended from a rigid support. A short wave pulse is introduced at its lowest end. It starts moving up the string. The time taken to reach the support is:  $(take g = 10 ms^{-2})$ 

> (3) (4)

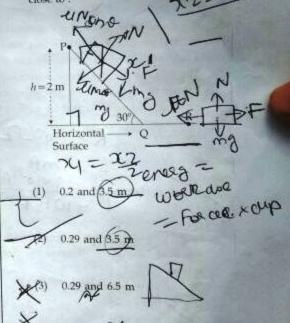
A person trying to lose weight by burning 62. fat lifts a mass of 10 kg upto a height of 1 m 1000 times. Assume that the potential energy lost each time he lowers the mass is dissipated. How much fat will he use up considering the work done only when the weight is lifted up? Fat supplies 3.8×107 J of energy per kg which is converted to mechanical energy with a 20% efficiency rate. Take  $g = 9.8 \text{ ms}^{-2}$ :

- $6.45 \times 10^{-3} \text{ kg}$ (1)
- $9.89 \times 10^{-3} \text{ kg}$ (2)
- $12.89 \times 10^{-3} \text{ kg}$ (3)
- $2.45 \times 10^{-3} \text{ kg}$ (4)

A point particle of mass m, moves along the uniformly rough track PQR as shown in the figure. The coefficient of friction, between the particle and the rough track equals  $\mu$ . The particle is released, from rest, from the point P and it comes to rest at a point R. The energies, lost by the ball, over the parts, PQ and QR, of the track, are equal to each other, and no energy is lost when particle changes direction from PQ to QR.

My My MA

The values of the coefficient of friction µ and the distance x(=QR), are, respectively close to:



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ILN = ULN SINDO

SPACE FOR ROUGH WORK

0.2 and 6.5 m

6-27×

64. Two identical wires A and B, each of length 'l', carry the same current l. Wire A is bent into a circle of radius R and wire B is bent to form a square of side 'a'. If B<sub>A</sub> and B<sub>B</sub> are the values of magnetic field at the centres of the circle and square respectively, then the ratio B<sub>A</sub> is:

- (1)  $\frac{\pi^2}{16\sqrt{2}}$ (2)  $\frac{\pi^2}{16}$ (3)  $\frac{\pi^2}{8\sqrt{2}}$ (4)  $\frac{\pi^2}{16}$
- A galvanometer having a coil resistance of 100 Ω gives a full scale deflection, when a current of 1 mA is passed through it. The value of the resistance, which can convert this galvanometer into ammeter giving a full scale deflection for a current of 10 A, is:
  (1) 2 Ω
  (2) R = 100 Ω
  - (2)  $0.1\Omega$  T = ImA(3)  $3\Omega$   $= 1 \times 10^{-5} A$ (4)  $0.01\Omega$   $T = 10A \times 3^{-1}$ .
- 66. An observer looks at a distant tree of height 10 m with a telescope of magnifying power of 20. To the observer the tree appears:
  - (1) 10 times nearer.
  - (2) 20 times taller.
  - (3) 20 times nearer.
  - (4) 10 times taller.

- The temperature dependence of resistances of Cu and undoped Si in the temperature range 300-400 K, is best described by :
  - Linear increase for Cu, exponential increase for Si.
  - Linear increase for Cu, exponential decrease for Si.
  - Linear decrease for Cu, linear decrease for Si.
  - Linear increase for Cu, linear increase for Si.

### Choose the correct statement:

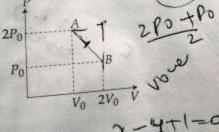
- In amplitude modulation the frequency of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal.
- (2) In frequency modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal.
- (3) In frequency modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the frequency of the audio signal.
- (4) In amplitude modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal.

SPACE FOR ROUGH WORK

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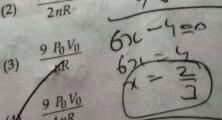
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2092 X to Half-lives of two radioactive elements 71. A and B are 20 minutes and 40 minutes, respectively. Initially, the samples have equal number of nuclei. After 80 minutes, the ratio of decayed numbers of A and B nuclei will be: 4:1 (2)1:4 5:4 (3)1:16 (4)'n' moles of an ideal gas undergoes a 70. process  $A \rightarrow B$  as shown in the figure. The maximum temperature of the gas during the process will be :





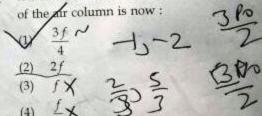
(2) 
$$\frac{9 P_0 V_0}{2nR}$$
  $2n + 5 = 0$ 



An arc lamp requires a direct current of 10 A at 80 V to function. If it is connected to a 220 V (rms), 50 Hz AC supply, the series inductor peeded for it to work is close to:

- 0.08 H (1)
- 0.044 H (2)
- 0.065 H (3)
- to her

A pipe open at both ends has a 72. fundamental frequency f in air. The pipe is dipped vertically in water so that half of it is in water. The fundamental frequency AN



The box of a pin hole camera, of length L, 73. has a hole of radius a. It is assumed that when the hole is illuminated by a parallel beam of light of wavelength  $\lambda$  the spread of the spot (obtained on the opposite wall of the camera) is the sum of its geometrical spread and the spread due to diffraction. The spot would then have its minimum size (say bmin) when:

(1) 
$$a = \sqrt{\lambda L}$$
 and  $b_{\min} = \left(\frac{2\lambda^2}{L}\right)$ 

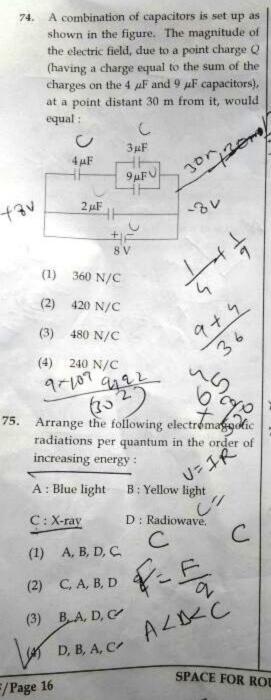
(2) 
$$a = \sqrt{\lambda L}$$
 and  $b_{\min} = \sqrt{4\lambda L}$ 

(3) 
$$a = \frac{\lambda^2}{\epsilon}$$
 and  $b_{\min} = \sqrt{4\lambda L}$ 

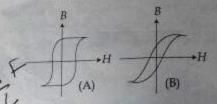
(4) 
$$a = \frac{\lambda^2}{L}$$
 and  $b_{\min} = \left(\frac{2\lambda^2}{L}\right)$ 

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76. Hysteresis loops for two magnetic materials
A and B are given below:



These materials are used to make magnets for electric generators, transformer core and electromagnet core. Then it is proper to use:

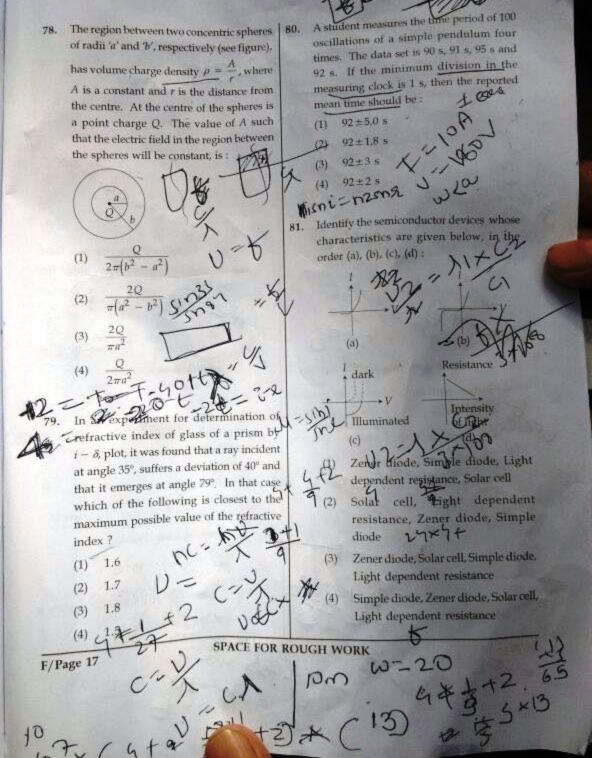
- A for electromagnets and B for electric generators.
- A for transformers and B for electric generators.
- (3) B for electromagnets and transformers,
- (4) A for electric generators and transformers.
- 77. A pendulum clock loses 12 s a day if the temperature is 40°C and gains 4 s a day if the temperature is 20°C. The temperature at which the clock will show correct time, and the co-efficient of linear expansion (a) of the metal of the pendulum shaft are respectively:
  - (1) 60°C; α=1.850010-4/°C
  - (2)  $30^{\circ}\text{C}$ ;  $\alpha = 1.85 \times 10^{-3}/{^{\circ}\text{C}}$
  - (3) 55°C;  $\alpha = 1.85 \times 10^{-2}/^{\circ}$ C

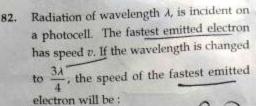
 $25^{\circ}\text{C}; \alpha = 1.85 \times 10^{-5}/^{\circ}\text{C}$ 

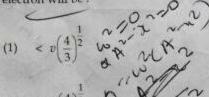
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(9 x 10 (4 + 2 + 4) + 2) x (3 + 4) x (0)







(2) 
$$= v \left(\frac{4}{3}\right)^{\frac{1}{2}}$$

$$(2) = v\left(\frac{4}{3}\right)^{2}$$

$$(3) = v\left(\frac{3}{4}\right)^{\frac{1}{2}}$$

$$(4)^{\frac{1}{2}}$$

$$(4)^{\frac{1}{2}}$$

$$(4)^{\frac{1}{2}}$$

$$(5)^{\frac{1}{2}}$$

$$(7)^{\frac{1}{2}}$$

$$(8)^{\frac{1}{2}}$$

$$(9)^{\frac{1}{2}}$$

$$(1)^{\frac{1}{2}}$$

$$(1)^{\frac{1}{2}}$$

$$(2)^{\frac{1}{2}}$$

$$(3)^{\frac{1}{2}}$$

$$(4)^{\frac{1}{2}}$$

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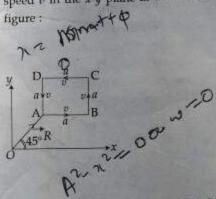
$$(1)^$$

A particle performs simple harmonic motion with amplitude A. Its speed is trebled at the instant that it is at a distance  $\frac{2A}{3}$  from equilibrium position. The new

- amplitude of the motion is:

  (1) 3A

A particle of mass m is moving along the side of a square of side 'a', with a uniform speed v in the x-y plane as shown in the figure:



Which of the following statements is false for the angular momentum L about the

(1) 
$$\vec{L} = mv \left[ \frac{R}{\sqrt{2}} - a \right] \hat{k}$$
 when the particle is moving from C to D.

(2) 
$$\vec{L} = mv \left[ \frac{R}{\sqrt{2}} + a \right] \hat{k}$$
 when the particle is moving from B to C.

(3) 
$$\overrightarrow{L} = \frac{mv}{\sqrt{2}} R \hat{k}$$
 when the particle is moving from D to A.

(4) 
$$\vec{L} = -\frac{mv}{\sqrt{2}} R \hat{k}$$
 when the particle is moving from A to B.

An ideal gas undergoes a quasi static, A roller is made by joining together two 87. reversible process in which its molar heat cones at their vertices O. It is kept on two capacity C remains constant. If during this process the relation of pressure P and volume V is given by  $PV^n = constant$ , then n is given by (Here  $C_p$  and  $C_V$  are molar specific heat at constant pressure and constant volume, respectively):

A screw gauge with a pitch of 0.5 mm and 86. a circular scale with 50 divisions is used to measure the thickness of a thin sheet of Before Aluminium. starting measurement, it is found that when the two jaws of the screw gauge are brought in contact, the 45th division coincides with the main scale line and that the zero of the main scale is barely visible. What is the thickness of the sheet if the main scale reading is 0.5 mm and the 25th division coincides with the main scale line? 0.80 mm

rails AB and CD which are placed asymmetrically (see figure), with its axis perpendicular to CD and its centre O at the centre of line joining AB and CD (see figure). It is given a light push so that it starts rolling with its centre O moving parallel to CD in the direction shown. As it moves, the roller will tend to:

- (1) turn right.
- go straight. (2)
- (3) turn left and right alternately.
- (4) turn left.
- 88. If a, b, c, d are inputs to a gate and x is its output, then, as per the following time graph, the gate is:

(2)

OR NANE

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0.75 mm

(1)

(2)

(3)

(4)

0.50 mm 2t 60=

- 89. For a common emitter configuration, if α and β have their usual meanings, the incorrect relationship between α and β is:
  - (1)  $\alpha = \frac{\beta}{1-\beta}$
  - (2)  $\alpha = \frac{\beta}{1+\beta}$
  - (3)  $a = \frac{\beta^2}{1 + \beta^2}$
  - $(4) \quad \frac{1}{\alpha} = \frac{1}{\beta} + 1$

- 90. A satellite is revolving in a circular orbit at a height h' from the earth's surface (radius of earth R; h<<R). The minimum increase in its orbital velocity required, so that the satellite could escape from the earth's gravitational field, is close to: (Neglect the effect of atmosphere.)
  - (1)  $\sqrt{gR}$
  - (2)  $\sqrt{gR/2}$
  - (3)  $\sqrt{gR} \left(\sqrt{2} 1\right)$   $\sqrt{2gR}$

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$$t2 = a + d = 4$$
  
 $t5 = a + 4d$   
 $t9 = a + 8d$