



# JEE MAIN 2015

## ONLINE EXAMINATION

**DATE : 11-04-2015**



### **Resonance Eduventures Pvt. Ltd.**

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# VIVEK

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**Target: JEE (Advanced) 2015**



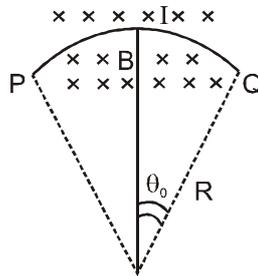
Target	Duration	Commencement Date/(Day)	End Date/(Day)
JEE (Advanced) 2015	05 Weeks*	13.04.2015 (Monday)	17.05.2015 (Sunday)

\*Approximate Duration

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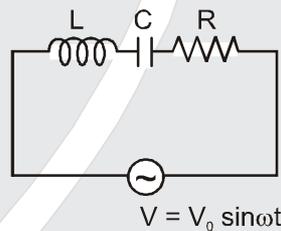


6. A wire carrying current  $I$  is tied between points  $P$  and  $Q$  and is in the shape of a circular arch of radius  $R$  due to a uniform magnetic field  $B$  (perpendicular to the plane of the paper, shown by  $\times \times \times$ ) in the vicinity of the wire. If the wire subtends an angle  $2\theta_0$  at the centre of the circle (of which it forms an arch) then the tension in the wire is :



- (1)  $\frac{IBR}{2\sin\theta_0}$       (2)  $\frac{IBR\theta_0}{\sin\theta_0}$       (3)  $IBR$       (4)  $\frac{IBR}{\sin\theta_0}$

7. For the LCR circuit, shown here, the current is observed to lead the applied voltage. An additional capacitor  $C'$ , when joined with the capacitor  $C$  present in the circuit, makes the power factor of the circuit unity. The capacitor  $C'$ , must have been connected in :



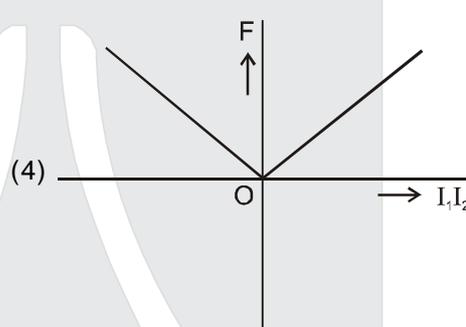
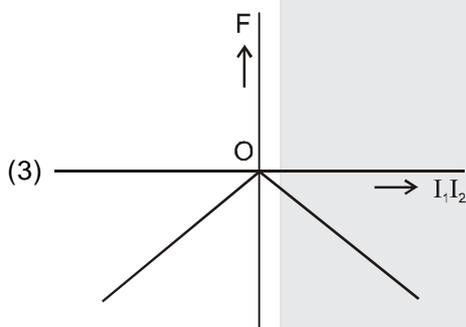
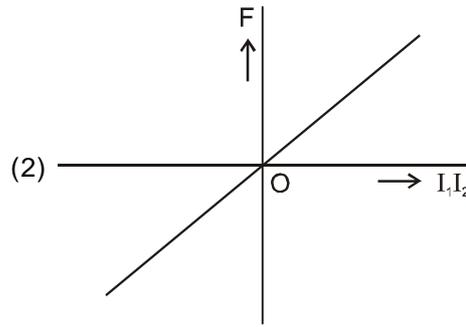
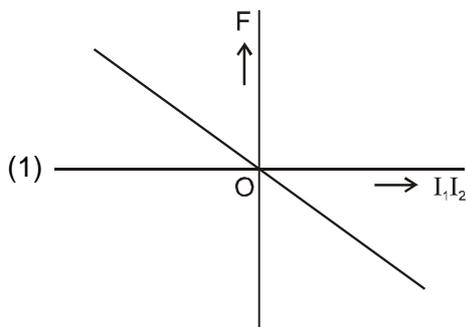
- (1) series with  $C$  and has a magnitude  $\frac{C}{(\omega^2 LC - 1)}$
- (2) series with  $C$  and has a magnitude  $\frac{1 - \omega^2 LC}{\omega^2 L}$
- (3) parallel with  $C$  and has a magnitude  $\frac{1 - \omega^2 LC}{\omega^2 L}$
- (4) parallel with  $C$  and has a magnitude  $\frac{C}{(\omega^2 LC - 1)}$



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8. Two long straight parallel wires, carrying (adjustable) current  $I_1$  and  $I_2$ , are kept at a distance  $d$  apart. If the force 'F' between the two wires is taken as 'positive' when the wires repel each other and 'negative' when the wires attract each other, the graph showing the dependence of 'F', on the product  $I_1 I_2$ , would be :



9. A pendulum with time period of 1s is losing energy due to damping. At certain time its energy is 45 J. If after completing 15 oscillations, its energy has become 15 J, its damping constant (in  $s^{-1}$ ) is :

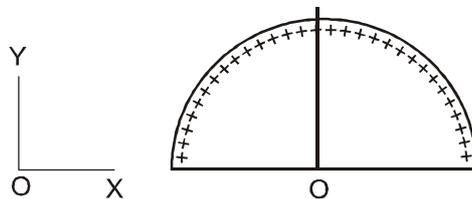
(1)  $\frac{1}{2}$

(2)  $\frac{1}{30} \ln 3$

(3) 2

(4)  $\frac{1}{15} \ln 3$

10. A wire, of length  $L$  ( $=20$  cm), is bent into a semicircular arc. If the two equal halves, of the arc, were each to be uniformly charged with charges  $\pm Q$ , [ $|Q| = 10^3 \epsilon_0$ . Coulomb where  $\epsilon_0$  is the permittivity (in SI units) of free space] the net electric field at the centre  $O$  of the semicircular arc would be :



(1)  $(50 \times 10^3 \text{ N/C}) \hat{j}$

(2)  $(50 \times 10^3 \text{ N/C}) \hat{i}$

(3)  $(25 \times 10^3 \text{ N/C}) \hat{j}$

(4)  $(25 \times 10^3 \text{ N/C}) \hat{i}$

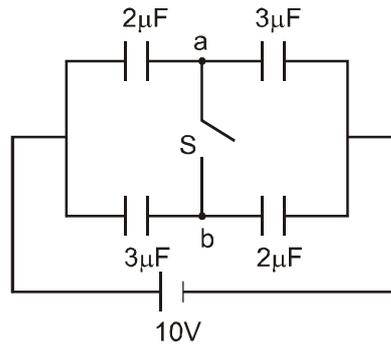


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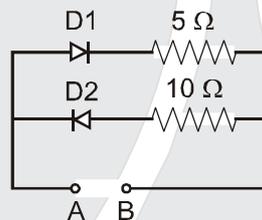
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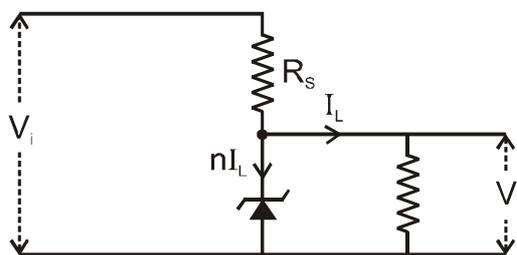
11. In figure is shown a system of four capacitors connected across a 10 V battery. Charge that will flow from switch S when it is closed is :



- (1)  $5 \mu\text{C}$  from b to a    (2)  $20 \mu\text{C}$  from a to b    (3) zero    (4)  $5 \mu\text{C}$  from a to b
12. A 2V battery is connected across AB as shown in the figure. The value of the current supplied by the battery when in one case battery's positive terminal is connected to A and in other case when positive terminal of battery is connected to B will respectively be :



- (1) 0.4 A and 0.2 A    (2) 0.2 A and 0.4 A    (3) 0.1 A and 0.2 A    (4) 0.2 A and 0.1 A
13. A cylindrical block of wood (density =  $650 \text{ kg m}^{-3}$ ), of base area  $30\text{cm}^2$  and height 54 cm, floats in a liquid of density  $900 \text{ kg m}^{-3}$ . The block is depressed slightly and then released. The time period of the resulting oscillations of the block would be equal to that of a simple pendulum of length (nearly) :
- (1) 52 cm    (2) 65 cm    (3) 39 cm    (4) 26 cm
14. The value of the resistor,  $R_s$ , needed in the dc voltage regulator circuit shown here, equals :



- (1)  $\frac{(V_i - V_L)}{(n+1)I_L}$     (2)  $\frac{(V_i + V_L)}{(n+1)I_L}$     (3)  $\frac{(V_i - V_L)}{nI_L}$     (4)  $\frac{(V_i + V_L)}{nI_L}$



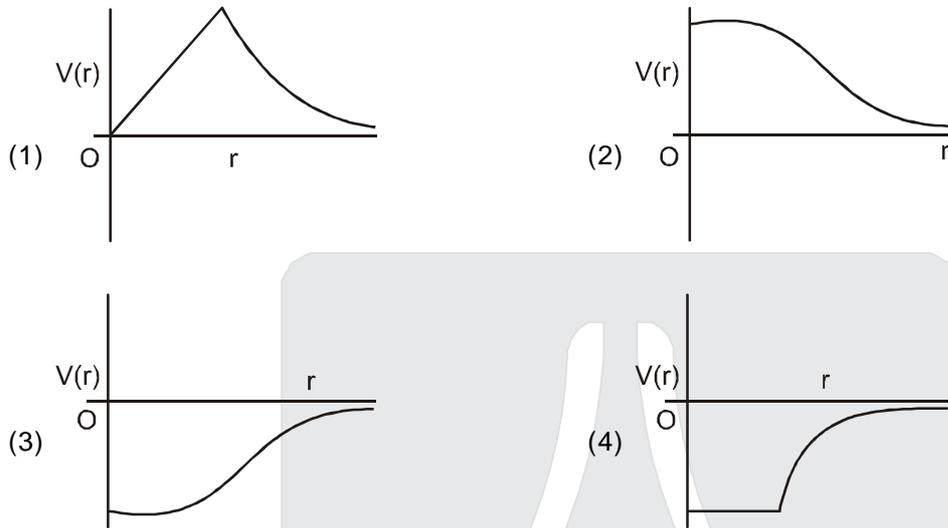
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15. If electronic charge  $e$ , electron mass  $m$ , speed of light in vacuum  $c$  and Planck's constant  $h$  are taken as fundamental quantities, the permeability, of vacuum  $\mu_0$  can be expressed in units of :

(1)  $\left(\frac{h}{me^2}\right)$       (2)  $\left(\frac{hc}{me^2}\right)$       (3)  $\left(\frac{h}{ce^2}\right)$       (4)  $\left(\frac{mc^2}{he^2}\right)$

16. Which of the following most closely depicts the correct variation of the gravitation potential  $V(r)$  due to a large planet of radius  $R$  and uniform mass density ? (figures are not drawn to scale)



17. In a Young's double slit experiment with light of wavelength  $\lambda$  the separation of slits is  $d$  and distance of screen is  $D$  such that  $D \gg d \gg \lambda$ . If the fringe width is  $\beta$ , the distance from point of maximum intensity to the point where intensity falls to half of maximum intensity on either side is:

(1)  $\frac{\beta}{6}$       (2)  $\frac{\beta}{3}$       (3)  $\frac{\beta}{4}$       (4)  $\frac{\beta}{2}$

18. Let  $N_\beta$  be the number of  $\beta$  particles emitted by 1 gram of  $N_a^{24}$  radioactive nuclei (half life = 15 hrs) in 7.5 hours,  $N_\beta$  is close to (Avogadro number =  $6.023 \times 10^{23}$ /g. mole) :

(1)  $6.2 \times 10^{21}$       (2)  $7.5 \times 10^{21}$       (3)  $1.25 \times 10^{22}$       (4)  $1.75 \times 10^{22}$

19. A short bar magnet is placed in the magnetic meridian of the earth with north pole pointing north. Neutral points are found at a distance of 30 cm from the magnet on the East - West line, drawn through the middle point of the magnet. The magnetic moment of the magnet in  $\text{Am}^2$  is close to :

(Given  $\frac{\mu_0}{4\pi} = 10^{-7}$  in SI units and  $B_H$  = Horizontal component of earth's magnetic field =  $3.6 \times 10^{-5}$  Tesla)

(1) 14.6      (2) 19.4      (3) 9.7      (4) 4.9

20. An experiment takes 10 minutes to raise the temperature of water in a container from  $0^\circ\text{C}$  to  $100^\circ\text{C}$  and another 55 minutes to convert it totally into steam by a heater supplying heat at a uniform rate. Neglecting the specific heat of the container and taking specific heat of water to be  $1 \text{ cal / g } ^\circ\text{C}$ , the heat of vapourization according to this experiment will come out to be :

(1) 560 cal/ g      (2) 550 cal / g      (3) 540 cal/ g      (4) 530 cal/ g

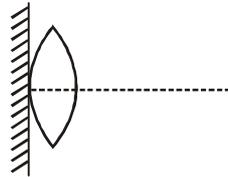


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21. A thin convex lens of focal length 'f' is put on a plane mirror as shown in the figure. When an object is kept at a distance 'a' from the lens - mirror combination, its image is formed at a distance  $\frac{a}{3}$  in front of the combination. The value of 'a' is :



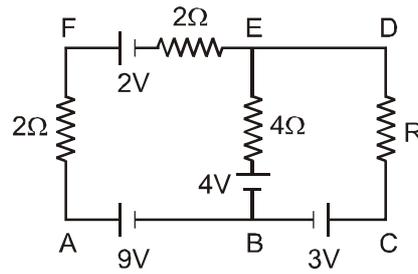
- (1)  $3f$                       (2)  $\frac{3}{2}f$                       (3)  $f$                       (4)  $2f$
22. A beaker contains a fluid of density  $\rho$  kg / m<sup>3</sup>, specific heat  $S$  J / kg°C and viscosity  $\eta$ . The beaker is filled upto height  $h$ . To estimate the rate of heat transfer per unit area ( $Q / A$ ) by convection when beaker is put on a hot plate, a student proposes that it should depend on  $\eta$ ,  $\left(\frac{S\Delta\theta}{h}\right)$  and  $\left(\frac{1}{\rho g}\right)$  when  $\Delta\theta$  (in °C) is the difference in the temperature between the bottom and top of the fluid. In that situation the correct option for ( $Q / A$ ) is :
- (1)  $\eta \left(\frac{S\Delta\theta}{h}\right) \left(\frac{1}{\rho g}\right)$       (2)  $\left(\frac{S\Delta\theta}{\eta h}\right) \left(\frac{1}{\rho g}\right)$       (3)  $\frac{S\Delta\theta}{\eta h}$                       (4)  $\eta \frac{S\Delta\theta}{h}$
23. The AC voltage across a resistance can be measured using a :
- (1) hot wire voltmeter                      (2) moving coil galvanometer  
(3) potential coil galvanometer                      (4) moving magnet galvanometer
24. Unpolarized light of intensity  $I_0$  is incident on surface of a block of glass at Brewster's angle. In that case, which one of the following statements is true ?
- (1) reflected light is completely polarized with intensity less than  $\frac{I_0}{2}$
- (2) transmitted light is completely polarized with intensity less than  $\frac{I_0}{2}$
- (3) transmitted light is partially polarized with intensity  $\frac{I_0}{2}$
- (4) reflected light is partially polarized with intensity  $\frac{I_0}{2}$
25. An electric field  $\vec{E} = (25\hat{i} + 30\hat{j})\text{NC}^{-1}$  exists in a region of space. If the potential at the origin is taken to be zero then the potential at  $x = 2$  m,  $y = 2$  m is :
- (1)  $-110$  J                      (2)  $-140$  J                      (3)  $-120$  J                      (4)  $-130$  J



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26. In the electric network shown, when no current flows through the  $4\Omega$  resistor in the arm EB, the potential difference between the points A and D will be :



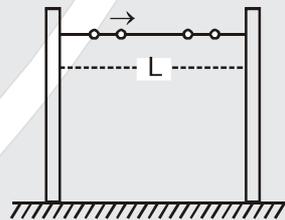
- (1) 6 V                      (2) 3 V                      (3) 5 V                      (4) 4 V
27. Using equipartition of energy, the specific heat (in  $\text{J kg}^{-1} \text{K}^{-1}$ ) of aluminium at room temperature can be estimated to be (atomic weight of aluminium = 27)
- (1) 410                      (2) 25                      (3) 1850                      (4) 925

28. A uniform thin rod AB of length L has linear mass density  $\mu(x) = a + \frac{bx}{L}$ , where x is measured from

A. If the CM of the rod lies at a distance of  $\left(\frac{7}{12}\right)L$  from A, then a and b are related as :

- (1)  $a = 2b$                       (2)  $2a = b$                       (3)  $a = b$                       (4)  $3a = 2b$

29. A large number (n) of identical beads, each of mass m and radius r are strung on a thin smooth rigid horizontal rod of length L ( $L \gg r$ ) and are at rest at random positions. The rod is mounted between two rigid supports (see figure). If one of the beads is now given a speed v, the average force experienced by each support after a long time is (assume all collisions are elastic) :



- (1)  $\frac{mv^2}{2(L-nr)}$                       (2)  $\frac{mv^2}{L-2nr}$                       (3)  $\frac{mv^2}{L-nr}$                       (4) zero

30. The de-Broglie wavelength associated with the electron in the  $n = 4$  level is :

- (1)  $\frac{1}{4}$  th of the de-Broglie wavelength of the electron in the ground state.  
 (2) four times the de-Broglie wavelength of the electron in the ground state  
 (3) two times the de-Broglie wavelength of the electron in the ground state  
 (4) half of the de-Broglie wavelength of the electron in the ground state

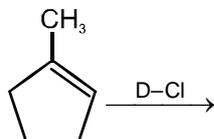


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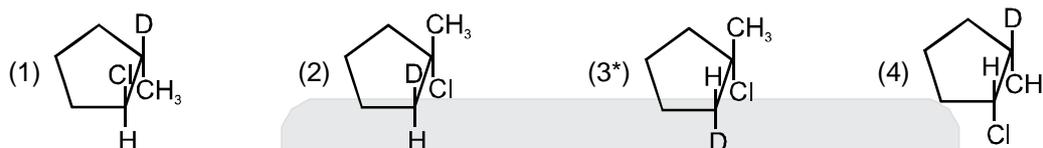
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**PART - B : CHEMISTRY**

1. What is the major product expected from the following reaction ?



Where D is an isotope of Hydrogen.



2. Which physical property of dihydrogen is wrong ?

(1) Odourless gas      (2) Tasteless gas      (3) Colourless gas      (4) Non-inflammable gas

3.  $A + 2B \rightarrow C$ , the rate equation for this reaction is given as

$$\text{Rate} = K[A][B]$$

If the concentration of A is kept the same but that of B is doubled what will happen to the rate itself ?

(1) halved      (2) the same      (3) doubled      (4) quadrupled

4. The number of structural isomers for  $C_6H_{14}$  is :

(1) 4      (2) 3      (3) 6      (4) 5

5. When concentrated HCl is added to an aqueous solution of  $CoCl_2$ , its colour changes from reddish pink to deep blue. Which complex ion gives blue colour in this reaction ?

(1)  $[CoCl_4]^{2-}$       (2)  $[CoCl_6]^{3-}$       (3)  $[CoCl_6]^{4-}$       (4)  $[Co(H_2O)_6]^{2+}$

6. Under ambient conditions, which among the following surfactants will form micelles in aqueous solution at lowest molar concentration ?

(1)  $CH_3-(CH_2)_8-COO^- Na^+$       (2)  $CH_3(CH_2)_{11}N^+(CH_3)_3Br^-$

(3)  $CH_3-(CH_2)_{13}-OSO_3^- Na^+$       (4)  $CH_3(CH_2)_{15}N^+(CH_3)_3Br^-$

7. Match the organic compounds in column-I with the Lassaigne's test results in column-II appropriately :

Column-I

(A) Aniline  
(B) Benzene sulfonic acid  
(C) Thiourea

Column-II

(i) Red colour with  $FeCl_3$   
(ii) Violet colour with sodium nitroprusside  
(iii) Blue colour with hot and acidic solution of  $FeSO_4$

(1) A - (ii); (B) - (iii); (C) - (i)

(2) A - (iii); (B) - (i); (C) - (ii)

(3) A - (iii); (B) - (ii); (C) - (i)

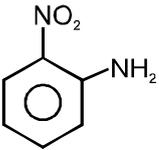
(4) A - (ii); (B) - (i); (C) - (iii)

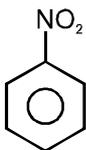


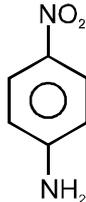
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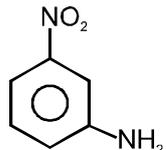
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8. The increase of pressure on ice  $\rightleftharpoons$  water system at constant temperature will lead to :  
 (1) a decrease in the entropy of the system      (2) an increase in the Gibbs energy of the system  
 (3) no effect on the equilibrium      (4) a shift of the equilibrium in the forward direction
9. Accumulation of which of the following molecules in the muscles occurs as a result of vigorous exercise ?  
 (1) Glycogen      (2) Glucose      (3) Pyruvic acid      (4) L-lactic acid
10. Which of the alkaline earth metal halides given below is essentially covalent in nature ?  
 (1)  $\text{SrCl}_2$       (2)  $\text{CaCl}_2$       (3)  $\text{BaCl}_2$       (4)  $\text{MgCl}_2$
11. Which of the following complex ions has electrons that are symmetrically filled in both  $t_{2g}$  and  $e_g$  orbitals ?  
 (1)  $[\text{FeF}_3]^{3-}$       (2)  $[\text{Mn}(\text{CN})_6]^{4-}$       (3)  $[\text{CoF}_6]^{3-}$       (4)  $[\text{Co}(\text{NH}_3)_6]^{2+}$
12. At 298 K, the standard reduction potentials are 1.51 V for  $\text{MnO}_4^-|\text{Mn}^{2+}$ , 1.36 V for  $\text{Cl}_2|\text{Cl}^-$ , 1.07 V for  $\text{Br}_2|\text{Br}$ , and 0.54 V for  $\text{I}_2|\text{I}^-$ . At pH = 3, permanganate is expected to oxidize :  $\left(\frac{RT}{F} = 0.059 \text{ V}\right)$   
 (1)  $\text{Cl}^-$ ,  $\text{Br}^-$  and  $\text{I}^-$       (2)  $\text{Br}^-$  and  $\text{I}^-$       (3)  $\text{Cl}^-$  and  $\text{Br}^-$       (4)  $\text{I}^-$  only
13. Calamine is an ore of :  
 (1) Zinc      (2) Aluminium      (3) Iron      (4) Copper
14. Which one of the following structures represents the neoprene polymer ?  
 (1)  $\left(\text{CH}_2-\underset{\text{C}_6\text{H}_5}{\text{CH}}\right)_n$       (2)  $\left(\text{CH}_2-\underset{\text{CN}}{\text{CH}}\right)_n$       (3)  $\left(\text{CH}_2-\underset{\text{Cl}}{\text{C}}=\text{CH}-\text{CH}_2\right)_n$       (4)  $\left(\text{CH}_2-\underset{\text{Cl}}{\text{CH}}\right)_n$
15. When does a gas deviate the most from its ideal behaviour ?  
 (1) At low pressure and low temperature      (2) At low pressure and high temperature  
 (3) At high pressure and low temperature      (4) At high pressure and high temperature
16. Which compound exhibits maximum dipole moment among the following ?
- (1) 

(2) 

(3) 

(4) 
17. Addition of phosphate fertilisers to water bodies causes :  
 (1) increase in amount of dissolved oxygen in water  
 (2) deposition of calcium phosphate  
 (3) increase in fish population  
 (4) enhanced growth of algae



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18. At temperature T, the average kinetic energy of any particle is  $\frac{3}{2}KT$ . The de Broglie wavelength follows the order :
- (1) Visible photon > Thermal neutron > Thermal electron
  - (2) Thermal proton > Thermal electron > Visible photon
  - (3) Thermal proton > Visible photon > Thermal electron
  - (4) Visible photon > Thermal electron > Thermal neutron
19. Which artificial sweetener contains chlorine ?
- (1) Sucralose
  - (2) Alitame
  - (3) Aspartame
  - (4) Saccharin
20. For the equilibrium,  $A(g) \rightleftharpoons B(g)$ ,  $\Delta H$  is  $-40$  kJ/mol. If the ratio of the activation energies of the forward ( $E_f$ ) and reverse ( $E_b$ ) reactions is  $\frac{2}{3}$  then :
- (1)  $E_f = 80$  kJ/mol;  $E_b = 120$  kJ/mol
  - (2)  $E_f = 60$  kJ/mol;  $E_b = 100$  kJ/mol
  - (3)  $E_f = 30$  kJ/mol;  $E_b = 70$  kJ/mol
  - (4)  $E_f = 70$  kJ/mol;  $E_b = 30$  kJ/mol
21. Chlorine water on standing loses its colour and forms :
- (1) HCl only
  - (2) HCl and  $HClO_2$
  - (3) HCl and HOCl
  - (4) HOCl and  $HClO_2$
22. Determination of the molar mass of acetic acid in benzene using freezing point depression is affected by :
- (1) partial ionization
  - (2) dissociation
  - (3) complex formation
  - (4) association
23.  $A + 3B + 3C \rightleftharpoons AB_2C_3$   
Reaction of 6.0 g of A,  $6.0 \times 10^{23}$  atoms of B, and 0.036 mol of C yields 4.8 g of compound  $AB_2C_3$ . If the atomic mass of A and C are 60 and 80 amu, respectively, the atomic mass of B is (Avogadro no. =  $6 \times 10^{23}$ ):
- (1) 50 amu
  - (2) 60 amu
  - (3) 70 amu
  - (4) 40 amu
24. Which of the following pairs of compounds are positional isomers ?
- (1)  $CH_3-CH_2-CH_2-\overset{\overset{O}{\parallel}}{C}-CH_3$  and  $CH_3-CH_2-\overset{\overset{O}{\parallel}}{C}-CH_2-CH_3$
  - (2)  $CH_3-CH_2-CH_2-CH_2-CHO$  and  $CH_3-CH_2-CH_2-\overset{\overset{O}{\parallel}}{C}-CH_3$
  - (3)  $CH_3-CH_2-CH_2-\overset{\overset{O}{\parallel}}{C}-CH_3$  and  $CH_3-\underset{\underset{CH_3}{|}}{CH}-CH_2-CHO$
  - (4)  $CH_3-CH_2-\overset{\overset{O}{\parallel}}{C}-CH_2-CH_3$  and  $\begin{matrix} CH_3 \\ \diagdown \\ CH_3-CH_2-CHO \\ \diagup \\ CH_3 \end{matrix}$
25. Which of the following compound has a P-P bond ?
- (1)  $H_4P_2O_5$
  - (2)  $(HPO_3)_3$
  - (3)  $H_4P_2O_6$
  - (4)  $H_4P_2O_7$



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26. Choose the incorrect formula out of the four compounds for an element X below :  
 (1)  $X_2O_3$                       (2)  $X_2Cl_3$                       (3)  $X_2(SO_4)_3$                       (4)  $XPO_4$
27. Molecular AB has a bond length of  $1.61 \text{ \AA}$  and a dipole moment of 0.38 D. The fractional charge on each atom (absolute magnitude) is : ( $e_0 = 4.802 \times 10^{-10} \text{ esu}$ )  
 (1) 0.5                      (2\*) 0.05                      (3) 0                      (4) 1.0
28. Which of the following statements is false ?  
 (1)  $Na_2Cr_2O_7$  is less soluble than  $K_2Cr_2O_7$                       (2)  $Na_2Cr_2O_7$  is primary standard in volumetry  
 (3)  $CrO_4^{2-}$  is tetrahedral in shape                      (4)  $CrO_7^{2-}$  has a Cr–O–Cr bond

29. In the reaction sequence



- (1)  $CH_3-CH_2-CH_2-CH_2-OH$                       (2)  $CH_3-CH=CH-CHO$   
 (3)  $CH_3-CH_2-CH_2-CH_3$                       (4)  $CH_3-\overset{\overset{O}{\parallel}}{C}-CH_3$

30. A pink coloured salt turns blue on heating. The presence of which cation is most likely ?  
 (1)  $Co^{2+}$                       (2)  $Cu^{2+}$                       (3)  $Zn^{2+}$                       (4)  $Fe^{2+}$



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**PART - C : MATHEMATICS**

1. The term independent of  $x$  in the binomial expansion of  $\left(1 - \frac{1}{x} + 3x^5\right)\left(2x^2 - \frac{1}{x}\right)^8$  is :  
 (1) 496                      (2) -496                      (3) 400                      (4) -400
2. Let  $k$  be a non-zero real number. If  $f(x) = \begin{cases} \frac{(e^x - 1)^2}{\sin\left(\frac{x}{k}\right) \log\left(1 + \frac{x}{4}\right)}, & x \neq 0 \\ 12, & x = 0 \end{cases}$  is a continuous function, then the value of  $k$  is :  
 (1) 4                      (2) 1                      (3) 3                      (4) 2
3. If the incentre of an equilateral triangle is  $(1, 1)$  and the equation of its one side is  $3x + 4y + 3 = 0$ , then the equation of the circumcircle of this triangle is :  
 (1)  $x^2 + y^2 - 2x - 2y - 14 = 0$                       (2)  $x^2 + y^2 - 2x - 2y - 2 = 0$   
 (3)  $x^2 + y^2 - 2x - 2y + 2 = 0$                       (4)  $x^2 + y^2 - 2x - 2y - 7 = 0$
4. Let  $f: \mathbb{R} \rightarrow \mathbb{R}$  be a function such that  $f(2-x) = f(2+x)$  and  $f(4-x) = f(4+x)$ , for all  $x \in \mathbb{R}$  and  $\int_0^2 f(x) dx = 5$ . Then the value of  $\int_{10}^{50} f(x) dx$  is :  
 (1) 125                      (2) 80                      (3) 100                      (4) 200
5. If  $\begin{vmatrix} x^2 + x & x + 1 & x - 2 \\ 2x^2 + 3x - 1 & 3x & 3x - 3 \\ x^2 + 2x + 3 & 2x - 1 & 2x - 1 \end{vmatrix} = ax - 12$ , then 'a' is equal to :  
 (1) 24                      (2) -12                      (3) -24                      (4) 12
6. Let  $k$  and  $K$  be the minimum and the maximum values of the function  $f(x) = \frac{(1+x)^{0.6}}{1+x^{0.6}}$  in  $[0, 1]$  respectively, then the ordered pair  $(k, K)$  is equal to :  
 (1)  $(2^{-0.4}, 1)$                       (2)  $(2^{-0.4}, 2^{0.6})$                       (3)  $(2^{-0.6}, 1)$                       (4)  $(1, 2^{0.6})$
7. If  $\cos \alpha + \cos \beta = \frac{3}{2}$  and  $\sin \alpha + \sin \beta = \frac{1}{2}$  and  $\theta$  is the arithmetic mean of  $\alpha$  and  $\beta$ , then  $\sin 2\theta + \cos 2\theta$  is equal to :  
 (1)  $\frac{3}{5}$                       (2)  $\frac{7}{5}$                       (3)  $\frac{4}{5}$                       (4)  $\frac{8}{5}$



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8. Let PQ be a double ordinate of the parabola,  $y^2 = -4x$ , where P lies in the second quadrant. If R divides PQ in the ratio 2 : 1 then the locus of R is :  
 (1)  $3y^2 = -2x$       (2)  $3y^2 = 2x$       (3)  $9y^2 = 4x$       (4)  $9y^2 = -4x$
9. In a parallelogram ABC,  $|\vec{AB}| = a$ ,  $|\vec{AD}| = b$  and  $|\vec{AC}| = c$ , then  $\vec{DA} \cdot \vec{AB}$  has the value :  
 (1)  $\frac{1}{2}(a^2 + b^2 + c^2)$       (2)  $\frac{1}{2}(a^2 - b^2 + c^2)$       (3)  $\frac{1}{4}(a^2 + b^2 - c^2)$       (4)  $\frac{1}{3}(b^2 + c^2 - a^2)$
10. If the two roots of the equation,  $(a - 1)(x^4 + x^2 + 1) + (a + 1)(x^2 + x + 1)^2 = 0$  are real and distinct, then the set of all values of 'a' is :  
 (1)  $\left(0, \frac{1}{2}\right)$       (2)  $\left(-\frac{1}{2}, 0\right) \cup \left(0, \frac{1}{2}\right)$       (3)  $\left(-\frac{1}{2}, 0\right)$       (4)  $(-\infty, -2) \cup (2, \infty)$
11. The solution of the differential equation  $ydx - (x + 2y^2)dy = 0$  is  $x = f(y)$ . If  $f(-1) = 1$ , then  $f(1)$  is equal to :  
 (1) 4      (2) 3      (3) 1      (4) 2
12. The shortest distance between the z-axis and the line  $x + y + 2z - 3 = 0 = 2x + 3y + 4z - 4$ , is :  
 (1) 1      (2) 2      (3) 4      (4) 3
13. From the top of a 64 metres high tower, a stone is thrown upwards vertically with the velocity of 48 m/s. The greatest height (in metres) attained by the stone, assuming the value of the gravitational acceleration  $g = 32 \text{ m/s}^2$ , is :  
 (1) 128      (2) 88      (3) 112      (4) 100
14. Let  $A = \{x_1, x_2, \dots, x_7\}$  and  $B = \{y_1, y_2, y_3\}$  be two sets containing seven and three distinct elements respectively. Then the total number of functions  $f : A \rightarrow B$  that are onto, if there exist exactly three elements  $x$  in  $A$  such that  $f(x) = y_2$ , is equal to :  
 (1)  $14 \cdot {}^7C_3$       (2)  $16 \cdot {}^7C_3$       (3)  $14 \cdot {}^7C_2$       (4)  $12 \cdot {}^7C_2$
15. If the lengths of the sides of a triangle are decided by the three throws of a single fair die, then the probability that the triangle is of maximum area given that it is an isosceles triangle, is :  
 (1)  $\frac{1}{21}$       (2)  $\frac{1}{27}$       (3)  $\frac{1}{15}$       (4)  $\frac{1}{26}$
16. If in a regular polygon the number of diagonals is 54, then the number of sides of this polygon is :  
 (1) 12      (2) 6      (3) 10      (4) 9
17. Let  $f : (-1, 1) \rightarrow \mathbb{R}$  be a continuous function. If  $\int_0^{\sin x} f(t) dt = \frac{\sqrt{3}}{2}x$ , then  $f\left(\frac{\sqrt{3}}{2}\right)$  is equal to :  
 (1)  $\frac{1}{2}$       (2)  $\frac{\sqrt{3}}{2}$       (3)  $\sqrt{\frac{3}{2}}$       (4)  $\sqrt{3}$



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18. If  $\int \frac{\log(1+\sqrt{1+t^2})}{\sqrt{1+t^2}} dt = \frac{1}{2}(g(t))^2 + C$ , where C is a constant, then  $g(2)$  is equal to :
- (1)  $\frac{1}{\sqrt{5}} \log(2 + \sqrt{5})$       (2)  $\frac{1}{2} \log(2 + \sqrt{5})$       (3)  $2\log(2 + \sqrt{5})$       (4)  $\log(2 + \sqrt{5})$
19. If a circle passing through the point  $(-1, 0)$  touches y-axis at  $(0, 2)$ , then the length of the chord of the circle along the x-axis is :
- (1)  $\frac{3}{2}$       (2) 3      (3)  $\frac{5}{2}$       (4) 5
20. The sum of the 3<sup>rd</sup> and the 4<sup>th</sup> terms of a G.P. is 60 and the product of its first three terms is 1000. If the first term of this G.P. is positive, then its 7<sup>th</sup> term is :
- (1) 7290      (2) 640      (3) 2430      (4) 320
21. A straight line L through the point  $(3, -2)$  is inclined at an angle of  $60^\circ$  to the line  $\sqrt{3}x + y = 1$ . If L also intersects the x-axis, then the equation of L is :
- (1)  $y + \sqrt{3}x + 2 - 3\sqrt{3} = 0$       (2)  $\sqrt{3}y + x - 3 + 2\sqrt{3} = 0$   
 (3)  $y - \sqrt{3}x + 2 + 3\sqrt{3} = 0$       (4)  $\sqrt{3}y - x + 3 + 2\sqrt{3} = 0$
22. If  $z$  is a non-real complex number, then the minimum value of  $\frac{\text{Im } z^5}{(\text{Im } z)^5}$  is :
- (1) -1      (2) -4      (3) -2      (4) -5
23. Let 10 vertical poles standing at equal distances on a straight line, subtend the same angle of elevation at a point O on this line and all the poles are on the same side of O. If the height of the longest pole is 'h' and the distance of the foot of the smallest pole from O is 'a'; then the distance between two consecutive poles, is :
- (1)  $\frac{h \cos \alpha - a \sin \alpha}{9 \sin \alpha}$       (2)  $\frac{h \sin \alpha + a \cos \alpha}{9 \sin \alpha}$   
 (3)  $\frac{h \cos \alpha - a \sin \alpha}{9 \cos \alpha}$       (4)  $\frac{h \sin \alpha - a \cos \alpha}{9 \cos \alpha}$
24. If the distance between the foci of an ellipse is half the length of its latus rectum, then the eccentricity of the ellipse is :
- (1)  $\frac{2\sqrt{2}-1}{2}$       (2)  $\sqrt{2}-1$       (3)  $\frac{1}{2}$       (4)  $\frac{\sqrt{2}-1}{2}$
25. A plane containing the point  $(3, 2, 0)$  and the line  $\frac{x-1}{1} = \frac{y-2}{5} = \frac{z-3}{4}$  also contains the point :
- (1)  $(0, 3, 1)$       (2)  $(0, 7, -10)$       (3)  $(0, -3, 1)$       (4)  $(0, 7, 10)$



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26. If  $\sum_{n=1}^5 \frac{1}{n(n+1)(n+2)(n+3)} = \frac{k}{3}$ , then k is equal to :
- (1)  $\frac{1}{6}$                       (2)  $\frac{17}{105}$                       (3)  $\frac{55}{336}$                       (4)  $\frac{19}{112}$
27. If the mean and the variance of a binomial variate X are 2 and 1 respectively, then the probability that X takes a value greater than or equal to one is :
- (1)  $\frac{9}{16}$                       (2)  $\frac{3}{4}$                       (3)  $\frac{1}{16}$                       (4)  $\frac{15}{16}$
28. If A is a  $3 \times 3$  matrix such that  $|5 \cdot \text{adj}A| = 5$ , then  $|A|$  is equal to :
- (1)  $\pm \frac{1}{5}$                       (2)  $\pm \frac{1}{25}$                       (3)  $\pm 1$                       (4)  $\pm 5$
29. The equation of a normal to the curve,  $\sin y = x \sin \left( \frac{\pi}{3} + y \right)$  at  $x = 0$ , is :
- (1)  $2x - \sqrt{3}y = 0$                       (2)  $2x + \sqrt{3}y = 0$   
 (3)  $2y - \sqrt{3}x = 0$                       (4)  $2y + \sqrt{3}x = 0$
30. Consider the following statements :  
 P : Suman is brilliant  
 Q : Suman is rich.  
 R : Suman is honest  
 the negation of the statement  
 "Suman is brilliant and dishonest if and only if suman is rich" can be equivalently expressed as :
- (1)  $\sim Q \leftrightarrow \sim P \vee R$                       (2)  $\sim Q \leftrightarrow \sim P \wedge R$   
 (3)  $\sim Q \leftrightarrow P \vee \sim R$                       (4)  $\sim Q \leftrightarrow P \wedge \sim R$



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