

## SCHOLASTIC APTITUDE TEST

(For Students of Class X)
Time: 120 Minutes
Max. Marks: 100

## INSTRUCTIONS TO CANDIDATES

Read the following instructions carefully before you open the test-booklet.

1. Answers are to be given on a separate OMR sheet.
2. There are $\mathbf{1 0 0}$ questions in this test. All are compulsory. The question numbers 1 to $t 0$ belong to Social Sciences. 41 to 60 pertain 10 Mathematics and 6t to 100 arc on Science subjects.
3. Please follow the instructions given on the OMR sheet for marking the answers.
4. Write your eight-digit roll number as allotted to you in the admission card clearly on the test-booklet and darken the appropriate circles on the OMR sheet as per instructions given.
5. Write down and darken test-booklet number in the appropriate circles on the OMR sheet as per instructions given.
6. Since the time allotted for this question paper is very limited and all questions carry equal marks, you should make the best use of it by not spending too much time on any one question.
7. Rough work can be done anywhere in the booklet but not on the OMR sheet.
8. Each correct answer will be awarded one mark.
9. THERE WILL BE NO NEGATIVE MARKING.
10. Please return only the OMR sheet to the invigilator after the test.
11. English version of the question paper will be considered as final in case of any dispute arising out of variation in translation.
Please turn over the page and start answering immediately after you are asked to do so.

Roll No.
रोल नंबर


Booklet Number
पुस्तिका संख्या 152547

## त्रोद्निक अमिक्नाता फ्रीसा

(कक्षा X के विद्यार्थियों के लिए)
समय: 120 मिनट
अध्यिकतम अंक: 100

## परीक्षार्थियों के लिए अनुवेश

प्रश्न-पुस्तिका खोलने से पहले, निम्नलिखित अनुदेशों को ध्यान से पढ़िए।

1. उत्तर एक अलग ओएमआर-पत्र पर देने है।
2. डस परोक्षा मे 100 प्रश्न हैं। सरी प्रश्न अनिबार्य है। प्रश्न 1 से 40 सामाजिक विज्ञान, 41 से 60 गणित और 61 से 100 विज्ञान के विषयों पर आधारित हैं।
3. कृपया उतर विक्षित करने के लिए ओएमआर-पत्र पर दिए गए निर्देशों का अनुपालन कीजिए।
4. दिए एए निरेशों के अनुसार आपके प्रथेश-पत्र पर दिए गए अपने आठ-अंकीय रोल नंबर को प्रश्न-पुस्तिका और ओएमआर-पत्र पर बिलकुल स्पार रूप से लिखिए और उपयुक्त गोलों को काला कीजिए।
5. दिए गए निर्वेशों के अनुसार ओएमआर-पत्र पर प्रश्न-पुस्तिका
संख्या लिखिए और उपयुक्त गोलों को काला कीजिए।
6. चूंकि इस प्रश्न-पत्र के लिए निर्धारित समय बहुत सीमित हैं, और सभी प्रश्नों के समान अंक है, इसलिए समय का अधिकतम उपयोग कीजिए और किसी प्रश्न पर बहुत अधिक समय न लगाइए।
7. रफ कार्य पुस्तिका में कहीं भी किया जा सकता है. कितु ओएमआर-पत्र / अलग कागज्त पर नहीं।
8. प्रत्येक सही उत्तर के लिए एक अंक प्रदान किया जाएगा।
9. गलत उत्तर के लिए कोई अंक नहीं काटा जाएगा।
10. कृपया परीक्षा के पश्चात केवल ओएमआर-पत्र ही निरीक्षक को वापस कीजिए।
11. अनुवादित सरंकरण में अंतर से उठे किसी भी विवाद की स्थिति में. प्रश्न-पत्र के अंग्रेज़ी सस्करण को निर्णायक माना जाएगा।

कृपया पृष्ठ पलटिए और अपना कार्य आरंभ कीजिए।

## NCERT 2019

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1. A set of guidelines called a model code of conduct is enforced during Parliamentary election in India to avert corrupt practices. This is applied to $\qquad$ .
2. Political parties only
3. Political parties and voters
4. Political parties and candidates
5. Candidates contesting elections only
6. A Country has the ultimate rights and power to make decisions on internal and external matters. Also it is not dictated by any external power on its decisions relating to its relation with other country. Which feature is reflected in these statements ?
7. Republic
8. Socialism
9. Sovereignty
10. Authoritarian
11. Anti-defection law is an important feature of Indian party system. The final authority to decide on the disqualification of a member of the parliament with respect to anti-deflection lies with the $\qquad$ .
12. President
13. Prime Minister
14. Chief Justice the Supreme Court of India
15. Speaker of the Lok Sabha or the Chairman of the Rajya Sabha
16. Which of the following statements exemplify the Independence of judiciary in India?
I. Judiciary is not under the control of executive and legislature.
II. There is less scope for interference in the working of judiciary by the political executive
III. A judge of higher judiciary can be removed only through a resolution which requires $2 / 3^{\text {rd }}$ majority of both the houses of parliament.
Choose the correct option.
17. I and II
18. I and III
19. I, II and III
20. II and III
21. Rajya Sabha is also called the upper house, elders house and permanent house. Which of the statement/s given below is/are true about it?
I. Rajya Sabha has more power related to financial matters.
II. Rajya Sabha members continue to be in the office till the next general election.
III. Resolution for removing the Vice-president and the president originates in Rajya Sabha.
IV. Number of seats allotted to a state in Rajya Sabha is directly proportionate to its population
22. I, II, and III
23. III and IV
(3) I and IV
(D) IV only
24. In the context of Indian elections, the parties which fail to gain majority in the parliament play the role of opposition. Consider the following statements and choose which statement/s is/are is not true.
I. Opposition parties in India play an important role in building public opinion.
II. Opposition parties are not constitutionally recognized.
III. Opposition immediately assumes power of government, if the majority party loses its vote of confidence in the Parliament.
IV. Opposition parties keep a close check on the activities of the government.
25. I and II only
26. II and III only
27. III only
28. IV only
29. Consider the following statements :

Statement I: All countries that are democratic have written constitution,
Statement II : All countries that have written constitution are not necessarily democratic.
Which of the above statement/s is/ are correct?

1. I only
2. II only
3. Both I and II
4. Neither I nor II
5. China exports a toy to India at ₹ 150 , Whereas the same toy is manufactured and available in India for ₹ 250 . When China continues to export this toy to India, this trade practice is known as $\qquad$ .
6. Dumping
7. Export Promotion
8. Import Substitution
9. Export Subsidisation
10. Shruti and Gautami were discussing about India's GDP and Kerala's SDP. Some of the observation made were:
I. Kerala's per capita SDP is India's GDP divided by Kerala's population in a particular year.
II. Since Kerala has best literacy rate and excellent quality of life indices, it must have the highest SDP.
III. In a federal structure if we know all the SDPs we can have a fair idea of how big India's GDP will be in that year.
IV. Kerala's per capita SDP in a particular year is the value of all final goods and services produced by the Kerala state in that year divided by Kerala's population in that year.
Which of the above statements are correct?
11. I and III
12. II and III
13. III and IV
14. I, III and IV
15. Shehnaaz joined a coaching institute for a professional course. At the time of joining the course, she paid a lump sum fee for the entire course of two years. However, she did not find the quality of teaching satisfactory and decided to quit after one year. When she asked for a refund of the fee for one year, she was refused. Which of the following right/s of Shehnaaz was/ were violated?
I. Right to choose
II. Right to represent
III. Right to be informed
IV. Right to seek redressal
16. Only I
17. I and IV
18. III and IV
19. Only IV
20. Which of the following reflects situation where a person is employed but do not contribute in adding to the total product?
I. Open unemployment
II. Disguised unemployment
III. Seasonal unemployment
IV. Frictional unemployment
21. I and II
22. Only II
23. III and IV
24. Only IV
25. There are 100 households in the village of Awangkhul, of which the loan taken by 20 household are from the State Bank of India, another 20 households from their friends and relatives, 5 households from Indian Bank, 10 households from a Regional Rural Bank, 15 households from businessmen, 10 households from village headmen and 20 households from cooperative societies. Which of the following inference (s) is/are correct.
I. Formal sources of credit are lower than the others.

II Institutional sources of credit are higher than others.
III. Non-institutional sources of credit are higher than others.
IV. Informal sources of credit are slightly higher than others.

1. Only I
2. I and II
3. Only II
4. III and IV
5. Which of the following statements are true about food security?
I. Landless people always have food insecurity.
II. Those who do not have enough nutritious food are food insecure.
III. Those who have enough food but not the requisite nutrition are food secure.
IV. Those who do not have enough purchasing power to buy sufficient food are food insecure.
6. I and III
7. I and IV
8. II and III
9. II and IV
10. Siddhik issues a cheque of $₹ 19,000$ in favour of Hanush's. What happens when the cheque is received and processed in Hamush's bank ?
I. There is no change in their bank accounts
II. Both their bank balances increase by ₹ 19,000 .
III. Siddhik's bank balance decreases by ₹ 19,000 and Hanush's bank balance increases by the same amount.
IV. There is no change in Siddhik' s bank balance although Hanush's bank balance sees an increase
Based on the above statements which option is correct?
11. Only I
12. I and III
13. Only III
14. III and IV
15. The daily wage of a person in rural area is ₹ 180 . Arrange the following households in descending order of vulnerability to poverty.

| Name of the <br> Household | Person days of <br> employement | Size of the <br> Household | Working <br> members <br> of the <br> family |
| :---: | :---: | :---: | :---: |
| Ruldoo | 14 | 4 | 2 |
| Mulkha | 15 | 7 | 2 |
| Fakira | 10 | 5 | 4 |
| Preeto | 12 | 6 | 3 |

1. Preeto $>$ Mulkha $>$ Fakira $>$ Ruldoo
2. Mulkha $>$ Preeto $>$ Ruldoo $>$ Fakira
3. Mulkha $>$ Ruldoo $>$ Preeto $>$ Fakira
4. Ruldoo $>$ Fakira $>$ Mulkha $>$ Preeto
5. The following graph shows the distribution of mean monthly temperature and average rainfall of a particular city during the year.


Which one of the following cities shows the climatic conditions presented in the above graph ?

1. Nagpur
2. Chennai
3. Jodhpur
4. Bengalurus

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17. The average mean monthly temperatures of four stations are given in the following table. The temperature is influenced by the movements of land and sea breezes.

|  | MONTHS <br> Temperature in Degree Celsius |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Stations/ | JAN | FEB | MAR | APR | MAY | JUN | JUL | AUG | SEP | OCT | NOV | DEC |
| A | 14.4 | 16.7 | 29.3 | 30.0 | 33.3 | 33.3 | 30.0 | 29.4 | 28.9 | 25.6 | 19.4 | 15.6 |
| B | 16.8 | 19.2 | 26.6 | 29.8 | 33.3 | 33.9 | 31.3 | 29.0 | 20.1 | 27.0 | 20.1 | 14.9 |
| C | 24.5 | 25.7 | 27.7 | 30.4 | 33.0 | 32.5 | 31.0 | 30.2 | 29.8 | 28.0 | 25.9 | 24.7 |
| D | 21.5 | 23.9 | 28.3 | 32.7 | 35.5 | 32.0 | 27.7 | 27.3 | 27.9 | 26.7 | 23.1 | 20.7 |

Which one of these stations experiences maximum moderating influence of the land and sea breezes?

1. A
2. B
3.C
4.D
3. Observe the data given in the following table.

| City | Female <br> Literacy <br> Rate <br> (\%) | Male <br> Literacy <br> Rate <br> (\%) | Sex-Ratio |
| :---: | :---: | :---: | :---: |
| A | 66.77 | 85.38 | 960 |
| B | 71.16 | 82.67 | 980 |
| C | 73.78 | 77.17 | 989 |
| D | 59.26 | 79.24 | 972 |

Based on the above table, identify the city which has the extent of equality between male and female better than the rest in terms of the given parameters?

1. A
2.B
$3 . C$
4.D
2. Ruhani observess sequential change in relation to altitudinal zones and natural vegetation types dominated by oak-chestnut; pine-deodar and silver fir-birch.
Identify the proper sequence of vegetation type she has observed from the following.
3. Alpine to Temperate to Subtropical 2. Subtropical to Temperate to Alpine
4. Subtropical to Alpine to Temperate
5. Temperate to Alpine to Subtropical
6. Observe the map given below


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Identify the shaded regions with their corresponding geographical features and select the correct option using the codes given below.

1. $A=$ Zone of laterite soil, $B=$ Coffee producing area, $C=$ Cotton textile industries, $D=$ Evergreen forest cover
2. $A=$ Evergreen forest cover, $B=$ Coffee producing areas, $C=$ Zone of laterite soil, $D=$ Cotton textile industries.
3. $\mathrm{A}=$ Evergreen forest cover, $\mathrm{B}=$ Zone of laterite soil, $\mathrm{C}=$ Coffee producing area, $\mathrm{D}=$ Cotton textile industries
4. $A=$ Cotton textile industries, $B=$ Coffee producing area, $C=$ Zone of laterite soil, $D=$ Evergreen forest cover.
5. Which of the following geological sequence properly matches the tectonic events from old to recent time periods?
6. Formation of Aravalli - Deccan volcanism- formation of Shiwalik - upliftment of Himadri.
7. Deccan volcanism - formation of Aravalli - Upliftement of Himadri- formation of Shiwalik.
8. Deccan volcanism - formation of Shiwalik - Upliflement of Himadri- formation of Aravalli.
9. Formation of Aravalli - Deccan volcanism - upliflment of Himadri- formation of Shiwalik.
10. The given map shows location of different mountain peaks in india.


A mountaineer wants to scale the mountain peaks in Peninsular India starting from North to South. Identify the correct sequence of peaks the mountaineer will follows ?

1. $A=$ Mahendragiri : $B=$ Anaimudi ; C = Dodabetta ; $D=$ Mahabaleshwar
2. $A=$ Dodabetta ; $B=$ Mahabaleshwar ; $C=$ Mahendragiri ; $D=A n a i m u d i$
3. $\mathrm{A}=$ Anaimudi ; $\mathrm{B}=$ Mahendragiri ; $\mathrm{C}=$ Dodabetta ; $\mathrm{D}=$ Mahabaleshwar
4. $A=$ Mahendragiri ; $B=$ Mahabaleshwar ; $C=$ Dodabetta ; D = Anaimudi

23 While teaching a topic on agriculture, geography teacher had made the following statement about a particular crop in her class. "Mean Monthly Temperature of about $27^{\circ} \mathrm{C}$., high relative humidity, rainfall of 150 cm in summer months and khaddar soils are the ideal physical requirements during the period of its vegetative growth."
Which one of the following crops was stated by the teacher ?

1. Tea
2. Jute
3. Rubber
4. Sugarcane
5. Bibhuti was travelling to study the traditional agricultural practices among various communities in Meghalaya, Jharkhand, Odisha and Western Ghats. Identify the correct sequence of forms of cultivation practised in these regions.
6. Jhumming -Kumari - Pama Dabi - Kuruwa
7. Kuruwa -Pama Dabi - Jhumming -Kumari
8. Jhumming -Kuruwa - Pama Dabi - Kumari
9. Pama Dabi - Kumari- Jhumming -Kuruwa
10. River Indus flows through Leh and Kargil districts in the state of Jammu and Kashmir. It has four major tributaries in India. Which one of the following is the correct sequence of the tributaries arranged from East to West in terms of their confluence with river Indus?
11. Zaskar - Dras -Hunza - Shyok
12. Zaskar - Hunza - Dras - Shyok
13. Hunza - Dras - Zaskar - Shyok
14. Zaskar - Dras - Shyok - Hunza
15. A tourist was travelling Indian States and came across a famous Buddhist Monastery, farming of three rice crops within the same agricultural year, a cement factory and floating gardens on a lake. Identify the proper sequence of the States the tourist travelled.
16. Sikkim - West Bengal - Assam - Meghalaya
17. Sikkim - Arunachal Pradesh - Assam - Manipur
18. Arunachal Pradesh - Assam - Meghalaya - Manipur
19. Arunachal Pradesh - West Bengal - Manipur - Meghalaya
20. Observe the following diagrams carefully.


Which one of the above population pyramids is an ideal representation of India's population?

1. I
2. II
3. III
4. IV
5. Which of the following statements regarding printing in Medieval Europe are correct?
I. Wood block printing reached Europe in the $13^{\text {th }}$ Century.
II. The aristocrats and monks criticized printed books as cheap vulgarities in the beginning.
III. Printing did not entirely displace the art of producing books by hand.
IV. Martin Luther had reservations against printing of books.
6. I, II and III
7. I, III and IV
8. I, II and IV
9. II, III and IV
10. Which of the following statements related to Mahatma Gandhi's view on Satyagraha are correct?
I. The movement in South Africa was not passive resistance.
II. It is the weapon of the people, who are not weak.
III. India could not militarily face Britain.
IV. Truth is the supreme dharma.
11. I, II and III
12. I, II and IV
13. II, III and IV
14. I, III and IV
15. Which of the following statements relating to the 'Scorched Earth Policy' in Java are correct?
I. The Dutch destroyed the saw mills.
II. Teak logs were burnt by the Dutch.
III. Trees were cut freely to meet war needs.
IV. The villagers were encouraged to expand cultivation in the forest areas.
16. I and II
17. I, II and III
18. I and IV
19. II, III and IV
20. Which of the following statements about opium cultivation in India during the British period are correct?
I. The peasants could sell off the produce freely.
II. Local traders offered higher prices for opium.
III. Opium production was increasing in territories that were not under the British
IV. Peasants were getting money advances from the village headman to produce opium.
21. I, II and III
22. I, II and IV
23. I, III and IV
24. II, III and IV
25. Observe the given picture taken from New Orleans, Illustrated London news, 1851.


What does the picture represent?

1. Mourning
2. Slave auction
3. Market place
4. Roadside gathering
5. Why were Nghe An and Ha Tinh provinces called 'electrical fuses' of Vietnam?
1.They were near to the capital city and were centers of power.
6. They were among the poorest provinces and had an old radical tradition.
7. They were very rich and had strong trade links with the outer world.
8. They were at the borders and were in conflicts with the neighbouring countries.
9. Which of the following would be the part of the surroundings in a Chawl in Bombay during the colonial period?
I. Large number of people living in shared rooms.
II. A large population of people belonging to depressed and lower classes.
III. Streets and neighbourhood being used for a variety of activities such as cooking, washing and sleeping.
IV. Liquor shops and Akharas in any open spot.
1.I, II and III
10. I, III and IV
11. II and III
12. II, III and IV
13. Which of the following statements are true in the context of Cricket in Victorian England?
I. The rules of Cricket were made to favour those who were described as "Players".
II. The wages of professionals was paid by patronage or subscription or gate money.
III. Cricket was viewed as a way of teaching English boys discipline, importance of hierarchy and leadership qualities.
IV. The rich who played were called amateurs.
14. I,II and III
15. I, II and IV
16. I, III and IV
17. II, III and IV
18. Which of the following statements are true for eighteenth century France?
I. There was much criticism of slavery.
II. The national Assembly feared opposition from businessmen who were dependant on slave trade.
III. Plantation owners understood their freedom as including the right to enslave Africans.
IV. The Convention of 1791 legislated to free all slaves in the French overseas possessions.
19. I and II
20. I, II and IV
21. II and III
22. II, III and IV
23. Which of the following statements are true in the context of Liberals in Modern Europe?
I. They opposed the uncontrolled power of dynastic rulers.
II. They wanted to safeguard the rights of individuals against governments.
III. They argued for Independent judiciary.
IV. They believed in universal adult franchise for all men and women with property.
24. I, II and III
25. I, II and IV
26. I, III and IV
27. II, III and IV

Directions : (Questions 38-40)
Read the statements and select the correct answer from the options given below.

1. Statement I is true, Statement II is false.
2. Statement I is false, Statement II is true.
3. Both Statements are true and Statement II provides explanation to Statement I.
4. Both Statements are true but Statement II does not provides explanation to Statement I.
5. Statement I: The Bretton woods System came up during the Post-World War Period.

Statement II : The industrial nations had massive growth of trade and incomes.
39. Statement I: Potatoes had been discovered by the Europeans in the Americas.

Statement II : Poor people in Ireland were dependent on potatoes to escape starvation in the $19^{\text {th }}$ century.
40. Statement I: The President of India cannot claim the kind of direct mandate that the prime Minister of India can.
Statement II : A candidate contesting for the post of President has to gain a majority of votes to be elected as the President of India.
41. If $m=n^{2}-n$, where $n$ is an integer, then $m^{2}-2 m$ is divisible by

1. 20
2. 24
3. 30
4. 16

Sol. (2)
$m^{2}-2 m \Rightarrow\left(n^{2}-n\right)^{2}-2\left(n^{2}-n\right)$
$\Rightarrow n^{4}-2 n^{3}-n^{2}+2 n$
$\Rightarrow n\left(n^{3}-2 n^{2}-n+2\right)$
$\Rightarrow(n-2)(n-1) n(n+1)$
$\therefore \mathrm{m}^{2}-2 \mathrm{~m}$ is divisible by 24 .
42. The value of $\sqrt{97 \times 98 \times 99 \times 100+1}$ is equal to

1. 9901
2. 9891
3. 9801
4. 9701

Sol. (4)
$\sqrt{97 \times 98 \times 99 \times 100+1}$
$\sqrt{(x-1) x(x+1)(x+2)+1}$
$\sqrt{\left(x^{2}+x-2\right)\left(x^{2}+x\right)+1}$ let $x^{2}+x=y$
$\sqrt{(y-2)(y)+1}$
$\sqrt{y^{2}-2 y+1}$
$\sqrt{(y-1)^{2}}$
$\Rightarrow \mathrm{y}-1$
$\Rightarrow x^{2}+x-1$
$\Rightarrow(98)^{2}+98-1$
$\Rightarrow 9701$
43. Let $P(x)$ be a polynomial of degree 3 and $P(n)=\frac{1}{n}$ for $n=1,2,3,4$. Then the value of $P(5)$ is

1. 0
2. $\frac{1}{5}$
3. $-\frac{2}{5}$
4. $\frac{3}{5}$

Sol. (1)

$$
P(n)=\frac{1}{n}, P(1)=1, P(2)=\frac{1}{2}, P(3)=\frac{1}{3}, P(4)=\frac{1}{4}
$$

$P(n) \cdot n=k(n-1)(n-2)(n-3)(n-4)+1$
$\mathrm{n}=0$
$0=k(-1)(-2)(-3)(-4)+1$
$\Rightarrow 24 \mathrm{k}+1=0$
$k=-\frac{1}{24}$
$P(n) \cdot n=-\frac{1}{24}(n-1)(n-2)(n-3)(n-4)+1$
$P(5)$.(5) $=-\frac{1}{24} \times 4 \times 3 \times 2 \times 1+1$
$P(5)=0$
44. If $\alpha$ and $\beta$ are the roots of the equation $3 x^{2}-5 x+3=0$, then the quadratic equation whose roots are $\alpha^{2} \beta$ and $\alpha \beta^{2}$ is

1. $3 x^{2}-5 x+3=0$
2. $3 x^{2}-8 x+5=0$
3. $3 x^{2}-8 x+3=0$
4. $3 x^{2}-5 x-3=0$

Sol. (1)
$3 x^{2}-5 x+3=0$
$\alpha \& \beta$ are zero
$\alpha+\beta=\frac{5}{3} \quad \alpha \beta=\frac{3}{3}=1$
Now the required quadratic equation
$x^{2}-\left(\alpha^{2} \beta+\alpha \beta^{2}\right) x+\alpha^{2} \beta \times \alpha \beta^{2}=0$
$x^{2}-\alpha \beta(\alpha+\beta) x+(\alpha \beta)^{3}=0$
$x^{2}-(1) \frac{5}{3} x+(1)^{3}=0$
$x^{2}-\frac{5 x}{3}+1=0$
$3 x^{2}-5 x+3=0$
45. In village Madhubani 8 women and 12 girls can paint a large mural in 10 hours. 6 women and 8 girls can paint it in 14 hours. The number of hours taken by 7 women and 14 girls to paint the mural is

1. 10
2. 15
3. 20
4. 35

Sol. (1)
$8 W+12 \mathrm{~g}=\frac{1}{10}$
$6 W+8 g=\frac{1}{14}$
$80 \mathrm{~W}+120 \mathrm{~g}=1$.....(1)
$84 \mathrm{~W}+112 \mathrm{~g}=1$......(2)
from eq. (1) \& (2)
$80 \mathrm{~W}+120 \mathrm{~g}=84 \mathrm{~W}+112 \mathrm{~g}$
$8 \mathrm{~g}=4 \mathrm{~W}$
$W=2 g$

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Now $8(2 \mathrm{~g})+12 \mathrm{~g}=\frac{1}{10}$
$28 \mathrm{~g}=\frac{1}{10}$
$\mathrm{g}=\frac{1}{280}$
$W=\frac{1}{140}$
So, $7 \mathrm{~W}+14 \mathrm{~g}=10$
46. If $x=\frac{3+\sqrt{5}}{2}$ and $y=x^{3}$, then $y$ satisfies the quadratic equation

1. $y^{2}-18 y+1=0$
2. $y^{2}+18 y+1=0$
3. $y^{2}-18 y-1=0$
4. $y^{2}+18 y-1=0$

Sol. (1)
$x=\frac{3+\sqrt{5}}{2}$
$x^{3}=\left(\frac{3+\sqrt{5}}{2}\right)^{3}=\frac{72+32 \sqrt{5}}{8}=9+4 \sqrt{5}$
$y=9+4 \sqrt{5}$
Since one root is $9+4 \sqrt{5}$
$\therefore \quad$ other root is $9-4 \sqrt{5}$
Sum of roots $=9+4 \sqrt{5}+9-4 \sqrt{5}=18$
Product $=(9+4 \sqrt{5})(9-4 \sqrt{5})=1$
$\therefore \quad$ polynomial is $\mathrm{y}^{2}-18 \mathrm{y}+1=0$.
47. If $\tan ^{2} \theta=1-e^{2}$, then the value of $\sec \theta+\tan ^{3} \theta \operatorname{cosec} \theta$ is equal to

1. $\left(1-\mathrm{e}^{2}\right)^{1 / 2}$
2. $\left(2-e^{2}\right)^{1 / 2}$
3. $\left(2-\mathrm{e}^{2}\right)^{3 / 2}$
4. $\left(1-e^{2}\right)^{3 / 2}$

Sol. (3)
$\tan ^{2} \theta=1-\mathrm{e}^{2}$
$\sec \theta+\tan ^{3} \theta \operatorname{cosec} \theta$

$$
\begin{aligned}
& =\frac{1}{\cos \theta}+\frac{\sin ^{3} \theta}{\cos ^{3} \theta} \times \frac{1}{\sin \theta}=\frac{1}{\cos \theta}+\frac{\sin ^{2} \theta}{\cos ^{3} \theta}=\frac{1}{\cos \theta}+\frac{1}{\cos \theta} \times \tan ^{2} \theta \\
& =\frac{1}{\cos \theta}\left(1+\tan ^{2} \theta\right)=\frac{1}{\frac{1}{\sqrt{2-e^{2}}}}\left(1+1-e^{2}\right)=\sqrt{2-e^{2}}\left(2-e^{2}\right)=\left(2-e^{2}\right)^{3 / 2}
\end{aligned}
$$

48. Let the volume of a solid sphere be $288 \pi \mathrm{~cm}^{3}$. A horizontal plane cuts the sphere at a distance of 3 cm from the centre so that the ratio of the curved surface areas of the two parts of the sphere is $3: 1$. The total surface area of the bigger part of the sphere (in $\mathrm{cm}^{2}$ ) is
49. $36 \pi$
50. $108 \pi$
51. $135 \pi$
52. $144 \pi$

Sol. (3)

$\frac{4}{3} \pi r^{3} \Rightarrow 288 \pi$
$r^{3}=\frac{3}{4} \times 288$
$r^{3}=216$
$r=6$
Curved surface Area ADB $=x$
Curved surface Area $A C B=4 \pi(6)^{2}-x=144 \pi-x$
Now,
$\frac{144 \pi-x}{x} \Rightarrow \frac{3}{1}$
$144 \pi-x=3 x$
$144 \pi=4 x$
$x=36 \pi$
$k=\sqrt{6^{2}-3^{2}} \Rightarrow \sqrt{36-9} \Rightarrow \sqrt{27}$
Area of flat surface of ACB
$\pi(\sqrt{27})^{2} \Rightarrow 27 \pi$
Total surface area of bigger part $\Rightarrow 27 \pi+$ Curved surface Area of bigger part
$\Rightarrow 27 \pi+144 \pi-36 \pi$
$\Rightarrow 135 \pi$
49. A solid metallic cylinder of height 10 cm and diameter 14 cm is melted to make two cones in the proportion of their volumes as $3: 4$, keeping the height 10 cm , what would be the percentage increase in the flat surface area?

1. 9
2. 16
3. 50
4. 200

Sol. (3)
Volume of cylinder $=\pi(7)^{2} \times 10 \Rightarrow 490 \pi$
Volume of first Cone $=\frac{1}{3} \pi\left(r_{1}\right)^{2} \times 10=\frac{490 \pi}{7} \times 3$
$r_{1}^{2}=7 \times 9 \Rightarrow r_{1}=3 \sqrt{7}$
Volume of second Cone $=\frac{1}{3} \pi\left(r_{2}\right)^{2} \times 10=\frac{490 \pi}{7} \times 4$
$r_{2}=2 \sqrt{21}$
Flat surface area of cylinder $=2 \pi(7)^{2}=98 \pi$
Flat surface area of cone $=\pi(3 \sqrt{7})^{2}+\pi(2 \sqrt{21})^{2}=147 \pi$
$\%$ increase $=\frac{49 \pi}{98 \pi} \times 100=50 \%$

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50. Each vertical face of square based vertical pillar of height 3 m has 7 equal, semi-cylindrical surfaces in such a way that its horizontal cross-section is as shown in the figure.


If the radius of each semi circle is 10 cm , the volume (in $\mathrm{m}^{3}$ ) of the pillar so designed $\left(\right.$ taking $\left.\pi=\frac{22}{7}\right)$ is

1. 5.88
2. 6.14
3. 6.42
4. 7.2

Sol. (4)


For cylinder
$r=10 \mathrm{~cm}=\frac{1}{100} \mathrm{~m}$
$h=3 \mathrm{~m}$
For cuboid
$I=140 \mathrm{~cm}=\frac{140}{100} \mathrm{~m}$
$\mathrm{b}=140 \mathrm{~cm}=\frac{140}{100} \mathrm{~m}$
$h=3 \mathrm{~m}$
Total volume $=14 \times$ volume of one cylinder + volume of cuboid
$=14 \times \pi r^{2} h+I \times b \times h$
$=14 \times \frac{22}{7} \times \frac{10}{100} \times \frac{10}{100} \times 3+\frac{140}{100} \times \frac{140}{100} \times 3$
$=1.32+5.88=7.2 \mathrm{~m}^{3}$
51. Let $A B C D$ be a square of side 20 cm . The area of the square $P Q R S$ (in cm$)^{2}$ interior to $A B C D$, shown in figure is


1. 60
2. 80
3. 100
4. 400

Sol. (2)


Area of $\triangle \mathrm{LQB}=$ Area of $\triangle \mathrm{NSD}=$ Area of $\triangle \mathrm{ORC}=$ Area of $\triangle \mathrm{MPA}=\mathrm{a}$
Area of square $=A+16 a=400$
Area of $\triangle A D N=\frac{1}{2} \times 20 \times 10=5 a$
$5 \mathrm{a}=100 \Rightarrow \mathrm{a}=20$
$16 a+A=400$
$A=400-16 \times 20$
$A=400-320=80$
52. A circle is inscribed in a right angled triangle of perimeter $7 \pi$. Then the ratio of numerical values of circumference of the circle to the area of the right angled triangle is

1. $4: 7$
2. $3: 7$
3. $2: 7$
4. $1: 7$

Sol. (1)
Given perimeter $=7 \pi$
Semiperimeter $=\frac{7 \pi}{2}$
Inradius $=\frac{\Delta}{S}$
$r=\frac{\frac{1}{2} \times b h}{\frac{7 \pi}{2}}$
$\frac{\pi r}{b h}=\frac{1}{7}$
$\Rightarrow \frac{\text { circumfere nce }}{\text { Area of right triangle }}=\frac{2 \pi r}{\frac{1}{2} \times b \times h} \quad \Rightarrow \frac{4 \pi r}{b h} \Rightarrow \frac{4}{7}$

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53. It is known that area of a cyclic quadrilateral is $\sqrt{(s-a)(s-b)(s-c)(s-d)}$ where $a, b, c, d$ are the sides and $s=\frac{a+b+c+d}{2}$.
If a circle can also be inscribed in the cyclic quadrilateral then the area of this quadrilateral is

1. $\sqrt{(\mathrm{ab})^{2}+(\mathrm{cd})^{2}}$
2. $\sqrt{a b c d}$
3. $\sqrt{(\mathrm{ac})^{2}+(\mathrm{ac})^{2}}$
4. $\sqrt{(\mathrm{ad})^{2}+(\mathrm{bc})^{2}}$

Sol. For cyclic quadrilateral $a+c=b+d$
$S=a+c=b+d$

$\therefore \quad$ Area $=\sqrt{(s-a)(s-b)(s-c)(s-d)}$
$=\sqrt{(a+c-a)(b+d-b)(a+c-c)(b+d-d)}$
area $=\sqrt{c \times d \times a \times b}$
area $=\sqrt{\mathrm{abcd}}$.
54. Two circles, both of radii a touch each other and each of them touches internally a circle of radius 2 a . Then the radius of the circle which touches all the three circles is

1. $\frac{1}{2} \mathrm{a}$
2. $\frac{2}{3} a$
3. $\frac{3}{4} \mathrm{a}$
4. a

Sol.
Since $\mathrm{PO}_{2}=2 \mathrm{a}-\mathrm{r}$
In $\triangle \mathrm{OPO}_{2}$ by Pythagoras theorem

$$
\begin{aligned}
& a^{2}+(2 a-r)^{2}=(a+r)^{2} \\
& a^{2}+4 a^{2}+r^{2}-4 a r=a^{2}+r^{2}+2 a r \\
& 4 a^{2} \Rightarrow 6 a r \\
& 2 a=3 r \\
& \frac{2}{3} a=r
\end{aligned}
$$

55. Let $D$ be a point on the side $B C$ of a triangle $A B C$ such that $\angle A D C=\angle B A C$. If $A C=21 \mathrm{~cm}$, then the side of an equilateral triangle whose area is equal to the area of the rectangle with sides $B C$ and $D C$ is
56. $14 \times 3^{1 / 2}$
57. $42 \times 3^{-1 / 2}$
58. $14 \times 3^{3 / 4}$
59. $42 \times 3^{1 / 2}$

Sol. (3)

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Given : AC = 21
$\triangle \mathrm{ACB} \sim \triangle \mathrm{DCA}$
$\underline{A C}=\frac{B C}{C A}=\frac{A B}{A D}$
$\overline{D C}=\frac{\overline{C A}}{=}=\frac{A D}{A D}$
$B C \times D C=A C^{2}=21^{2}=441$
$\frac{\sqrt{3}}{4} a^{2}=441$
$a^{2}=\frac{441 \times 4}{\sqrt{3}}$
$a=42 \times 3^{-114}$
$a=14 \times 3 \times 3^{-114}$
$=14 \times 3^{314}$
56. Let $A B C$ be a triangle with sides $a, b, c$. Then lengths of medians of the triangle formed by the medians of the triangle $A B C$ are

1. $\frac{1}{2} a, \frac{1}{2} b, \frac{1}{2} c$
2. $\frac{2}{3} \mathrm{a}, \frac{2}{3} \mathrm{~b}, \frac{2}{3} \mathrm{c}$
3. $\frac{3}{4} a, \frac{3}{4} b, \frac{3}{4} c$
4. $\frac{5}{6} a, \frac{5}{6} b, \frac{5}{6} c$

## Sol. (3)



By Apollonius theorem
$2\left[d_{1}^{2}+\left(\frac{a}{2}\right)^{2}\right]=b^{2}+c^{2}$
$\mathrm{d}_{1}^{2} \Rightarrow \frac{\mathrm{~b}^{2}+\mathrm{c}^{2}}{2}-\frac{\mathrm{a}^{2}}{4}$
$\mathrm{d}_{2}^{2} \Rightarrow \frac{\mathrm{a}^{2}+\mathrm{c}^{2}}{2}-\frac{\mathrm{b}^{2}}{4}$.
$d_{3}^{2} \Rightarrow \frac{a^{2}+b^{2}}{2}-\frac{c^{2}}{4}$
By Apollonius theorem


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$2\left(M_{1}^{2}+\frac{d_{3}^{2}}{4}\right) \Rightarrow d_{1}^{2}+d_{2}^{2}$
$M_{1}^{2}+\frac{d_{3}^{2}}{4} \Rightarrow \frac{d_{1}^{2}+d_{3}^{2}}{2}$
$M_{1}^{2} \Rightarrow \frac{d_{1}^{2}+d_{2}^{2}}{2}-\frac{d_{3}^{2}}{4}$
$M_{1}^{2} \Rightarrow \frac{\frac{b^{2}+c^{2}}{2}-\frac{a^{2}}{4}+\frac{a^{2}+c^{2}}{2}-\frac{b^{2}}{4}}{2}-\frac{\frac{a^{2}+b^{2}}{2}-\frac{c^{2}}{4}}{4}$
$M_{1}^{2} \Rightarrow \frac{\left(\frac{2 b^{2}+2 c^{2}-a^{2}+2 a^{2}+2 c^{2}-b^{2}}{4}\right)}{2}-\frac{\left(\frac{2 a^{2}+2 b^{2}-c^{2}}{4}\right)}{4}$
$M_{1}^{2} \Rightarrow \frac{2 b^{2}+2 c^{2}-a^{2}+2 a^{2}+2 c^{2}-b^{2}}{8}-\frac{2 a^{2}+2 b^{2}-c^{2}}{16}$
$M_{1}^{2} \Rightarrow \frac{4 b^{2}+4 c^{2}-2 a^{2}+4 a^{2}+4 c^{2}-2 b^{2}-2 a^{2}-2 b^{2}+c^{2}}{16}$
$M_{1}^{2} \Rightarrow \frac{9 c^{2}}{16} \Rightarrow M_{1}=\frac{3}{4} c$
Similarly $M_{2}=\frac{3}{4} b$
$M_{3}=\frac{3}{4} a$
57. $(x+1)^{4}$ is divided by $(x-1)^{3}$. Then the value of the remainder at $x=1$ is

1. -16
2. 0
3. 16
4. 32

Sol. (3)
Dividend $=(x+1)^{4}$
Divisor $=(x-1)^{3}$
Quotient $=\mathrm{q}(\mathrm{x})$
$R(x)=$ Remainder $=a x^{2}+b x+c$
By division algorithm
Dividend $=$ Divisor $\times$ Quotient + Remainder
$(x+1)^{4}=(x-1)^{3} q(x)+a x^{2}+b x+c$
Put $x=1$
$(1+1)^{4}=(1-1)^{3} q(1)+a(1)^{2}+b(1)+c$
$16=0+a+b+c$
$R(1)=a(1)^{2}+b(1)+c=a+b+c=16$
58. A circle passes through the vertices of a triangle ABC . If the vertices are $\mathrm{A}(-2,5), \mathrm{B}(-2,-3)$, $C(2,-3)$, then the centre of the circle is

1. $(0,0)$
2. $(0,1)$
3. $(-2,1)$
4. $(0,-3)$

Sol. (2)


```
\(\mathrm{OA}=\mathrm{OB}=\mathrm{OC}=\) Radius
\(O B^{2}=O A^{2}\)
\((x+2)^{2}+(y+3)^{2}=(x+2)^{2}+(y-5)^{2}\)
\(16 y=16\)
\(\mathrm{y}=1\)
\(\mathrm{OB}^{2}=\mathrm{OC}^{2}\)
\((x+2)^{2}+(y+3)^{2}=(x-2)^{2}+(y+3)^{2}\)
\(8 \mathrm{x}=0\)
\(x=0\)
\(\mathrm{O}(\mathrm{x}, \mathrm{y})=\mathrm{O}(0,1)\)
```

59. If two dice are thrown together, the probability that the difference of the numbers appearing on them is a prime number
60. $\frac{2}{9}$
61. $\frac{4}{9}$
62. $\frac{5}{12}$
63. $\frac{17}{36}$

Sol. (2)
Minimum difference 0
Maximum difference 5
Prime number between 0 and 5 are 2, 3, 5
Prime Dice 1 Dice 2
2
$3 \quad 1$
$1 \quad 3$
$4 \quad 2$
$2 \quad 4$
$5 \quad 3$
$3 \quad 5$
$6 \quad 4$
$4 \quad 6$
3
$\begin{array}{ll}4 & 1 \\ 1 & 4\end{array}$
$5 \quad 2$
$2 \quad 5$
$6 \quad 3$
$3 \quad 6$
5
$6 \quad 1$
So total cases 16
Total possible outcomes $=6 \times 6=36$
Required probability $=\frac{16}{36}=\frac{4}{9}$.
60. Observe the following data.

| Class | $0-20$ | $20-40$ | $40-60$ | $60-80$ | $80-100$ | Total |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Frequency | 17 | $\mathrm{f}_{1}$ | 32 | $\mathrm{f}_{2}$ | 19 | 120 |

If the above data has mean 50 , then missing frequencies $f_{1}$ and $f_{2}$ are respectively.

1. 28 and 24
2. 24 and 28
3. 28 and 30
4. 30 and 28

Sol. (1)

| C.I. | Frequency(f) | ClassMarks(x) | fx |
| :--- | :---: | :---: | :---: |
| $0-20$ | 17 | 10 | 170 |
| $20-40$ | $\mathrm{f}_{1}$ | 30 | $30 f_{1}$ |
| $40-60$ | 32 | 50 | 1600 |
| $60-80$ | $\mathrm{f}_{2}$ | 70 | $70 \mathrm{f}_{2}$ |
| $80-100$ | 19 | 90 | 1710 |
| $\Sigma \mathrm{ff}=120$ |  |  | $\Sigma \mathrm{fx}=3480+30 \mathrm{f}_{1}+70 \mathrm{f}_{2}$ |

$$
120=68+f_{1}+f_{2}
$$

$$
\mathrm{f}_{1}+\mathrm{f}_{2}=120-68
$$

$$
\begin{equation*}
\mathrm{f}_{1}+\mathrm{f}_{2}=52 \tag{i}
\end{equation*}
$$

$\overline{\mathrm{x}}=\frac{\Sigma \mathrm{fx}}{\Sigma \mathrm{f}}$
$50=\frac{3480+30 f_{1}+70 f_{2}}{120}$
$6000-3480=30 f_{1}+70 f_{2}$
$252=3 f_{1}+7 f_{2}$
On solving (i) and (ii)
$\mathrm{f}_{2}=24$
$\mathrm{f}_{1}=28$.
Direction : (Question 61-62)
Suppose that the acceleration versus time graph of a particle that starts from rest at $t=0$ is as shown in the figure.

61. At what instant does the particle come to rest for the first time ?

1. 5 s
2. 10 s
3. 15 s
4. The particle never comes to rest

Sol. (3)
$\mathrm{A}_{1}+\mathrm{A}_{2}=0$
$5(10)-(t-10)(10)=0$
$(\mathrm{t}-10)=5$
$\mathrm{t}=15 \mathrm{sec}$

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62. What is the total distance travelled by the particle during 30 s ?

1. 0 m
2. 500 m
3. 750 m
4. 1000 m

Sol. (3)
Total distance $=\frac{1}{2}(15)(50)+\left|\frac{1}{2}(15)(-50)\right|=750 \mathrm{~m}$

63. An object of mass 2 kg moving under the action of a force which varies with time as shown in the figure.


Which one of the following statements is correct for the interval from 0 to 20 s ?

1. The momentum of the object decreases by $75 \mathrm{~kg} \mathrm{~m} / \mathrm{s}$.
2. The momentum of the object increases by $75 \mathrm{~kg} \mathrm{~m} / \mathrm{s}$.
3. The momentum of the object increases by $125 \mathrm{~kg} \mathrm{~m} / \mathrm{s}$.
4. The change in momentum cannot be found as initial speed is unknown.

Sol. (2)
From graph $\Delta \mathrm{P}=\frac{1}{2}(15+5) 10-\frac{1}{2}(10) 5=75 \mathrm{kgm} / \mathrm{s}$
Momentum of object increases by $75 \mathrm{kgm} / \mathrm{s}$
64. Two cars ' $A$ ' and ' $B$ ' of same mass start from the same location at the same time but on different straight roads. Car ' $A$ ' travels on a road that has greater angle of inclination with horizontal compared to the road on which ' $B$ ' travels.
At any instant both cars ' $A$ ' and ' $B$ ' have the same height above the starting point. If $E_{A}$ and $E_{B}$ are total energies of cars ' $A$ ' and ' $B$ ' respectively, them

1. $E_{A}<E_{B}$
2. $E_{A}=E_{B}$
3. $E_{A}>E_{B}$
4. Relation between $E_{A}$ and $E_{B}$ cannot be decided based on given information.

Sol. (1)


According to the figure $\theta_{A}>\theta_{B}$
At any instant height is same so potential energy will be same but B covers more distance in same wtime so speed of $B$ will be more hence kinetic energy of $B$ will be more. Mechanical energy of $A$ will be less than that of $B$
65. The gravitational potential energy difference per unit mass between the surface of a planet and a point 100 m above it is $1000 \mathrm{~J} / \mathrm{kg}$. How much work is required to be done in moving a 5 kg object 100 m on a slope at $30^{\circ}$ to the horizontal on this planet?

1. 1250 J
2. 2500 J
3. 4350 J
4. 5000 J

Sol. (2)
Since h is only 100 m . so difference in gravitational potential energy $=\mathrm{mgh}$. or difference gravitational potential energy per unit mass,

$$
=\mathrm{gh}=1000 \text { (given) }
$$


where $h=100 \mathrm{~m}$
$\therefore \mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}$
Now work done $=\mathrm{mgh}^{\prime}=5 \times 10 \times 50=2500 \mathrm{~J}$

## Direction : (Question 66-67)

Two identical objects $A$ and $B$ each of mass $m$ start moving along the same vertical line in opposite directions at the same instant. Object A is dropped from rest from a height H above the ground and object B is projected vertically upward from the ground with speed $\mathrm{u}=\sqrt{2 g H}$
66. At what height above the ground do they collide?

1. $(1 / 4) \mathrm{H}$
2. $(1 / 2) \mathrm{H}$
3. $(2 / 3) \mathrm{H}$
4. $(3 / 4) \mathrm{H}$

Sol.
(4)

$\mathrm{S}_{1}=\mathrm{H}-\mathrm{h}=\frac{1}{2} \mathrm{~g} \mathrm{t}^{2}$
$\mathrm{S}_{2}=\mathrm{h}=\mathrm{ut}-\frac{1}{2} \mathrm{gt}^{2}$
(1) \& (2) $H=u t \Rightarrow t=\frac{H}{u}$
$h=H-\frac{1}{2} g \frac{H^{2}}{u^{2}} \quad u=\sqrt{2 g H}$
$=H-\frac{H}{4}=\frac{3 H}{4}$
67. After they collide, they stick to each other. What is the loss in their total energy?

1. 0
2. $(1 / 2) \mathrm{mgH}$
3. $(3 / 2) \mathrm{mgH}$
4. 2 mgH

Sol. (2)
$v_{1}^{2}=2 g H-2 g\left(\frac{3 H}{4}\right)=\frac{g H}{2} \Rightarrow v_{1}=\sqrt{\frac{g H}{2}}$
$v_{2} \sqrt{2 g h} \sqrt{2 g \frac{H}{4}} \sqrt{\frac{g \mathrm{H}}{2}}$
$\mathrm{U}_{1}=0$
$\mathrm{U}_{2}=\mathrm{mgH}$
$\mathrm{K}_{1}=\mathrm{mgH}$
$\mathrm{K}_{2}=0$
$\mathrm{TE}_{\mathrm{i}}=2 \mathrm{mgH}$
From momentum conservation at time of collision
$m v_{1}-m v_{2}=2 \mathrm{mv}^{\prime}$
$v^{\prime}=$ Common velocity after collision $=0$
$T E_{f}=2 m g \frac{3 H}{4}=\frac{3 m g H}{2}$
$\Delta \mathrm{TE}=\frac{3 \mathrm{mgH}}{2}-2 \mathrm{mgH}=\frac{-\mathrm{mgH}}{2}$
68. Given below are two different graphs of variation of density (or pressure) of the medium with position (fig. 1) and with time (fig. 2) as a wave passes through the medium.



Fig. 2
What will be the speed of the wave in the given medium?

1. $25 \mathrm{~m} / \mathrm{s}$
2. $50 \mathrm{~m} / \mathrm{s}$
3. $250 \mathrm{~m} / \mathrm{s}$
4. $500 \mathrm{~m} /$

Sol. (4)


Fig. 1


Fig. 2

From graph :
Time period $=2 \mathrm{~ms}$
$\lambda=100 \mathrm{~cm}$

$$
\begin{aligned}
& \text { So } c=\frac{\lambda}{T}=\frac{100}{2 \times 10^{-3}}=50 \times 10^{3} \mathrm{~cm} / \mathrm{s} \\
& \mathrm{c}=500 \mathrm{~m} / \mathrm{s}
\end{aligned}
$$

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69. A convex lens and a concave lens, each of focal length 10 cm , are kept separated by a distance of 2 cm as shown in the figure. If the light is incident from left, the combinations of lenses will be $\qquad$ .


1. Converging
2. Diverging
3. Behaving like a glass slab
4. Converging or diverging depending on whether the lenses are arranged as shown in the figure or in the reverse order.
Sol. (1)


Equivalent focal length of combination
$\frac{1}{F_{\text {eq }}}=\frac{1}{F_{1}}+\frac{1}{F_{2}}-\frac{d}{F_{1} F_{2}}$
$=\frac{1}{10}-\frac{1}{10}-\frac{2}{10 \times(-10)}=\frac{2}{100}=0.02 \mathrm{~cm}$
So, this combination behave as conversing lens.
70. In the circuit given, the ratio of work done by the battery to maintain the current between point A and $B$ to the work done for the whole circuit is


1. $\frac{1}{117}$
2. $\frac{1}{13}$
3. $\frac{1}{12}$
4. 1

Sol. (2)


Total current of circuit

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$$
I=\frac{V}{R_{\text {equ }}}=\frac{1.3}{2+\frac{1}{3}+2}=\frac{1.3 \times 3}{6+1+6}=\frac{1.3 \times 3}{130}=0.3 \mathrm{~A}
$$

Now P.d. between $A$ and $B V_{1}=I R_{A B}$

$$
=0.3 \times \frac{1}{3}=0.1 \mathrm{Volt}
$$

So Ratio of work done

$$
\frac{W_{1}}{W_{2}}=\frac{V_{1} I}{V I}=\frac{0.1}{1.3}=\frac{1}{13}
$$

71. Magnetic field at the centre of a circular coil of radius R carrying current i is $B \propto \frac{i}{R}$ and its direction is given by right-hand thumb rule. Magnetic field at the centre of a circular arc subtending an angle $\theta$ (in degree) is $B \propto \frac{i}{R}\left(\frac{\theta}{360^{\circ}}\right)$ and its direction can be found using right hand rule.


Consider two circular coils made of uniform conductors as shown in figure 3 and 4. In figure 3 points $C$ and $D$ are diametrically opposite to each other, and in figure $4 \angle P O_{2} Q=120^{\circ}$. Then magnetic fields $\qquad$ .


1. At both $\mathrm{O}_{1}$ and $\mathrm{O}_{2}$ are zero
2. is zero at $\mathrm{O}_{1}$ but non-zero at $\mathrm{O}_{2}$

3. At both $\mathrm{O}_{1}$ and $\mathrm{O}_{2}$ are non-zero
4. is non-zero at $\mathrm{O}_{1}$ but zero at $\mathrm{O}_{2}$.

Sol. (1)

$B_{1}=k \frac{\mathrm{i}_{1}}{R} \frac{270}{360}$
$B_{2}=k \frac{i_{2}}{R} \frac{90}{360}$

So net MF at $\mathrm{O}_{1}$,
$\mathrm{B}=\mathrm{B}_{2}-\mathrm{B}_{1}$
$=\frac{K}{R}\left(\frac{1}{4} i_{2}-\frac{3}{4} i_{1}\right)$
$=\frac{K}{4 R}\left[i_{2}-3 i_{1}\right]$
Now we know for parallel combination
$\mathrm{I}_{1} \mathrm{R}_{1}=\mathrm{I}_{2} \mathrm{R}_{2}$
$I_{2}\left(\frac{\mathrm{R}}{360} \times 90^{\circ}\right)=I_{2}\left(\frac{\mathrm{R}}{360} \times 270\right)$
$\frac{\mathrm{I}_{2}}{4}=\frac{3 \mathrm{I}_{1}}{4}$
$\mathrm{I}_{2}=3 \mathrm{I}_{1}$
So
$B=\frac{k}{4 R}\left[\left.3\right|_{1}-31_{1}\right]$
So, $B$ at Oss ${ }_{1}$
$B=0$
Similarly


B at $\mathrm{O}_{2}$
$B=\frac{k}{R}\left[\frac{I_{4}}{3}-\frac{2}{3} I_{3}\right]$
$B=\frac{\mathrm{k}}{3 \mathrm{R}}\left[\mathrm{l}_{4}-2 \mathrm{I}_{3}\right]$
Now we know for parallswel combination
$\mathrm{I}_{3}\left[\frac{\mathrm{R}}{360} \times 240\right]=\mathrm{I}_{4}\left(\frac{\mathrm{R}}{360} \times 120\right)$
$\frac{2 I_{3}}{3}=\frac{I_{4}}{3}$
$I_{4}=2 I_{3}$
So B at $\mathrm{O}_{2}$
$B=\frac{K}{3 R}\left[2 I_{3}-2 I_{3}\right]=0$
So, w
Net MF at $\mathrm{O}_{1}$ and $\mathrm{O}_{2}$ are zero
72. A pin $A B$ of length 2 cm is kept on the axis of a convex lens between 18 cm and 20 cm as shown in figure. Focal length of convex lens is 10 cm . Find magnification produced for the image of the pin.


1. 0.83
2. 1.00
3. 1.25
4. 6.78

Sol. (3)
Image of $A$ is formed at $20 \mathrm{~cm}\left(\mathrm{~V}_{1}\right)$ and for point B
$u_{2}=-18 \mathrm{~cm}, \quad f=10 \mathrm{~cm}$
$\frac{1}{F}=\frac{1}{v_{2}}-\frac{1}{u_{2}}$
$v_{2}=\frac{f u_{2}}{f+u_{2}}=\frac{10 \times-18}{10-18}=\frac{-10 \times 18}{-8}$
$v_{2}=22.5 \mathrm{~cm}$
Image length $=22.5-20=2.5 \mathrm{~cm}$
and linear magnification
$\mathrm{m}=\frac{\mathrm{h}_{2}}{\mathrm{~h}_{1}}=\frac{2.5}{2}=1.25$
73. What is the current supplied by the battery in the circuit shown below? Each resistance used in circuit is of $1 \mathrm{k} \Omega$ and potential difference $\mathrm{V}_{\mathrm{AB}}=8 \mathrm{~V}$.


1. 64 mA
2. 15 mA
3. $\quad 9.87 \mathrm{~mA}$
4. 1 mA

## Sol. (2)

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$R_{A B}=\frac{1}{5}+\frac{1}{3}=\frac{3+5}{15}=\frac{8}{15} \mathrm{k} \Omega$
$I=\frac{V_{A B}}{R_{A B}}=\frac{8}{\frac{8}{15} \times 10^{3}}=15 \times 10^{-3}=15 \mathrm{~mA}$
74. Read the following statements.

Statement -I : Sodium metal reacts violently with water to produce heat and fire.
Statement -II : Potassium metal reacts violently with water to form potassium hydroxide and hydrogen gas.
Select the correct answer from the options given below

1. Statement-I is true, Statement II is false
2. Statement-I is false, Statement II is true
3. Both statements are true, and Statement II provides correct explanation to Statement-I
4. Both statements are true, and Statement II does not provides correct explanation to Statement-I

Sol. Na and K are high reactive metal and reacts violently with water.
$\mathrm{Na}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{NaOH}+\mathrm{H}_{2}(\mathrm{~g})+$ Energy
$\mathrm{K}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{KOH}+\mathrm{H}_{2}(\mathrm{~g})+$ Energy
Ans 4
75. You are provided with 18 g each of $\mathrm{O}_{2}, \mathrm{~N}_{2}, \mathrm{CH}_{4}$ and $\mathrm{H}_{2} \mathrm{O}$. Which of the following is the correct decreasing order of number of atoms present in these samples ?

1. $\mathrm{CH}_{4}>\mathrm{H}_{2} \mathrm{O}>\mathrm{N}_{2}>\mathrm{O}_{2}$
2. $\mathrm{O}_{2}>\mathrm{N}_{2}>\mathrm{H}_{2} \mathrm{O}>\mathrm{CH}_{4}$
3. $\mathrm{CH}_{4}>\mathrm{N}_{2}>\mathrm{O}_{2}>\mathrm{H}_{2} \mathrm{O}$
4. $\mathrm{N}_{2}>\mathrm{H}_{2} \mathrm{O}>\mathrm{O}_{2}>\mathrm{CH}_{4}$

Sol.

|  | $\mathrm{O}_{2}$ | $\mathrm{~N}_{2}$ | $\mathrm{CH}_{4}$ | $\mathrm{H}_{2} \mathrm{O}$ |
| :--- | :--- | :--- | :--- | :--- |
| Given mass | 18 g | 18 g | 18 g | 18 g |
| Number of Mole | $\frac{18}{32}$ | $\frac{18}{28}$ | $\frac{18}{16}$ | $\frac{18}{18}$ |
| Number of atoms | $\frac{18}{32} \times 2 \times \mathrm{N}_{\mathrm{A}}$ <br> $=1.125 \mathrm{~N}_{\mathrm{A}}$ | $\frac{18}{28} \times 2 \times \mathrm{N}_{\mathrm{A}}$ <br> $=1.285 \mathrm{~N}_{\mathrm{A}}$ | $\frac{18}{16} \times 5 \times \mathrm{N}_{\mathrm{A}}$ <br> $=5.625 \mathrm{~N}_{\mathrm{A}}$ | $\frac{18}{18} \times 3 \times \mathrm{N}_{\mathrm{A}}$ <br> $=3 \mathrm{~N}_{\mathrm{A}}$ |

$N_{A}=6.023 \times 10^{23}$
$\therefore$ decreasing order of number of atoms is $\mathrm{CH}_{4}>\mathrm{H}_{2} \mathrm{O}>\mathrm{N}_{2}>\mathrm{O}_{2}$
Ans 1
76. Manya, Kartik, Gurnoor and Sheena has arranged the ions $\mathrm{F}^{-}, \mathrm{Na}^{+}, \mathrm{O}^{2-}$, and $\mathrm{Mg}^{+}$in decreasing orders of their ionic radii.
Manya $-\mathrm{O}^{2-}>\mathrm{Mg}^{2+}>\mathrm{F}^{-}>\mathrm{Na}^{+}$
Kartik $-\mathrm{Mg}^{2+}>\mathrm{Na}^{+}>\mathrm{O}^{2-}>\mathrm{F}^{-}$
Gurnoor $-\mathrm{O}^{2-}>\mathrm{F}^{-}>\mathrm{Na}^{+}>\mathrm{Mg}^{2+}$
Sheena $-\mathrm{F}^{-}>\mathrm{Na}^{+}>\mathrm{O}^{2-}>\mathrm{Mg}^{2+}$
Who had provided the correct order of their decreasing ionic radii ?

1. Manya
2. Kartik
3. Gurnoor
4. Sheena

Sol.

$$
\xrightarrow[\text { Same number of electrons (isoelectronic) }]{\mathrm{O}^{2-}>\mathrm{F}^{-}>\mathrm{Na}^{+}>\mathrm{Mg}^{2+}}
$$

$$
\xrightarrow[\text { Number of protons increases }]{ }
$$

$\qquad$
$\therefore$ Size decreases
Ans 3
77. An organic compound $A$ on heating with concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}$ gave product $B$ and on warming with alkaline $\mathrm{KMnO}_{4}$ gave compound C . Compound A on heating with compound C in presence of concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}$ formed compound D , which has fruity smell.
Identify the compounds $\mathrm{A}, \mathrm{B}, \mathrm{C}$ and D

1. $A=$ Alcohol, $B=$ Carboxylic acid, $C=$ Alkene,$D=$ Ester
2. $A=$ Carboxylic acid, $B=$ Ester, $C=$ Alkene, $D=$ Alcohol
3. $\mathrm{A}=$ Alcohol, $\mathrm{B}=$ Alkene, $\mathrm{C}=$ Carboxylic acid, $\mathrm{D}=$ Ester
4. $A=$ Alkene, $B=$ Alcohol, $C=$ Ester, $D=$ Carboxylic acid

Sol. Alcohol Alkene


## ' C '

Acetic acid


Ans 3
78. Match List-I (Mixture) and List-II (Type) with the list III (Example) and select the correct answer from the combination given below :

| List-I (Mixture) | List-II (Type) | List-III (Example) |
| :--- | :--- | :--- |
| A. Liquid in gas | 1. Emulsion | I. Mist |
| B. Liquid in liquid | 2. Aerosol | II. Sponge |
| C. Gas in solid | 3. Foam | III. Face cream |
|  | 4. Gel | IV. Butter |
|  | 2. A-2-I, B-1-III, C-3-II |  |
| 1. A-3-II, B-2-III, C-4-IV 4. A-1-II, B-4-I, C-2-III |  |  |

Sol. Liquid in liquid is emulsion e.g. milk, face cream etc.
Liquid in gas is aerosol. E.g moisture in air or mist.
Gas in solid is foam (solid foam) e.g sponge
Ans. 2
79. Which of the following set of reactions will NOT occur ?
I. $\mathrm{MgSO}_{4}(\mathrm{aq})+\mathrm{Fe}(\mathrm{s}) \rightarrow \mathrm{FeSO}_{4}(\mathrm{aq})+\mathrm{Mg}(\mathrm{s})$
II. $\mathrm{CuSO}_{4}(\mathrm{aq})+\mathrm{Fe}(\mathrm{s}) \rightarrow \mathrm{FeSO}_{4}(\mathrm{aq})+\mathrm{Cu}(\mathrm{s})$
III. $\mathrm{MgSO}_{4}(\mathrm{aq})+\mathrm{Cu}(\mathrm{s}) \rightarrow \mathrm{CuSO}_{4}(\mathrm{aq})+\mathrm{Mg}(\mathrm{s})$
IV. $\mathrm{CuSO}_{4}(\mathrm{aq})+\mathrm{Zn}(\mathrm{s}) \rightarrow \mathrm{ZnSO}_{4}(\mathrm{aq})+\mathrm{Cu}(\mathrm{s})$

1. I and III
2. II and IV
3. I, II and III
4. II, III and IV

Sol. More reactive metal can displace less reactive metal from its salt solution. According to activity series:
Reactivity order of following metal will be
$\mathrm{Mg}>\mathrm{Zn}>\mathrm{Fe}>\mathrm{Cu}$
Ans. 1
80. Two organic compounds ' $A$ ' and ' $B$ ' react with sodium metal and both produce the same gas ' $X$ ', but with sodium hydrogen carbonate only compound $B$ reacts to give a gas ' $Y$ '. Identify ' $A$ ', ' $B$ ', ' $X$ ' and ' $Y$ ':

1. $A=$ Ethylene, $B=$ Ethyl Alcohol, $X=$ Carbon dioxide, $Y=$ Hydrogen
2. $A=$ Ethyl Alcohol, $B=$ Acetic acid, $X=$ Hydrogen, $Y=$ Carbon dioxide
3. $A=$ Methyl alcohol, $B=$ Ethyl alcohol, $X=$ Hydrogen, $Y=$ Carbon dioxide
4. $A=$ Acetic acid, $B=$ Formic acid, $X=$ Carbon dioxide, $Y=$ Hydrogen

Sol. $\quad \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}+\mathrm{Na} \rightarrow \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{ONa}+\mathrm{H}_{2} \uparrow$
$\mathrm{CH}_{3} \mathrm{COOH}+\mathrm{Na} \rightarrow \mathrm{CH}_{3} \mathrm{COONa}+\mathrm{H}_{2} \uparrow$
$\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}+\mathrm{NaHCO}_{3} \rightarrow \mathrm{X}$ (No reaction)
$\mathrm{CH}_{3} \mathrm{COOH}+\mathrm{NaHCO}_{3} \rightarrow \mathrm{CH}_{3} \mathrm{COONa}+\mathrm{H}_{2} \mathrm{O}+\mathrm{CO}_{2}$
Ans. 2
81. Consider the elements A, B, C and D with atomic numbers $11,12,16$ and 17 , respectively. Which among the following statements regarding these elements are correct?
I. The element $C$ will gain electron more easily than element $D$
II. The element B tends to lose electron more readily than C .
III. The oxide of $A$ will be least basic while that of $D$ will be most basic
IV. The energy required to remove an electron from outermost shell from A will be minimum while that from $D$ will be maximum.

1. I and III only
2. I and IV only
3. II and III only
4. II and IV only

Sol.
Atomic number Element
A 11
B $\quad 12$
C $\quad 16$
Na
Mg
S
D 17 Cl
$A$ \& $B$ will show metallic properties while $C$ \& $D$ will show non metallic properties
Ans 4
82. The following observations are given for four metals :
I. Metal H does not react with dilute HCl
II. Metal K reacts with warm water
III. Metal L does not react with water but displaces metal H from its aqueous salt solution
IV. Metal $M$ reacts with cold water

Choose the correct decreasing order of reactivity of these metals amongst the following:

1. $M>L>H>K$
2. $K>M>H>L$
3. $\mathrm{M}>\mathrm{K}>\mathrm{L}>\mathrm{H}$
4. $\mathrm{L}>\mathrm{H}>\mathrm{K}>\mathrm{M}$

Sol. $\quad M \rightarrow$ React with cold water [Most reactive]
$\mathrm{K} \rightarrow$ React with Hard water [Most reactive]
$L \rightarrow$ Do not react with water but do react with salt solution
$\mathrm{H} \rightarrow$ Do not react with dil. acid
$\therefore$ decreasing order of reactivity of these metals $M>K>L>H$
Ans. 3
83. Match chemical reactions given in the list-I with the type of chemical reactions given in List-II and select the correct answer using the options given below

| List-I (Chemical Reactions) | List-II (Type of Chemical Reactions) |
| :--- | :--- |
| A. Formation of $\mathrm{NH}_{3}$ from $\mathrm{N}_{2}$ and $\mathrm{H}_{2}$ | I. Decomposition |
| B. Calcination of zinc carbonate II. Double displacement <br> C. Reaction of aqueous $\mathrm{BaCl}_{2}$ solution with III. Combination <br> dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$ IV. Redox <br> D. Rancidity of oils V. Displacement <br> 1. A-I, B-V, C-III, D-IV 2. A-III, B-IV, C-V, D-I <br> 3. A-IV, B-III, C-V, D-I 4. A-III, B-I, C-II, D-IV A. |  |

Sol. A. Formation of $\mathrm{NH}_{3}$ from $\mathrm{N}_{2} \& \mathrm{H}_{2}$ - Combination
B. Calcination of zinc carbonate - decomposition
C. Reaction of aqueous solution of $\mathrm{BaCl}_{2}$ with $\mathrm{H}_{2} \mathrm{SO}_{4}$ - Double displacement
D. Rancidity of oil - Redox

Ans. 4
84. You are provided with aqueous solutions of three salts - A, B and C, 2-3 drops of blue litmus solution, red litmus solution and phenolphthalein were added to each of these solution in separate experiments. The change in colours of different indicators were recorded in the following table

| Sample | With blue Litmus solution | With Red Litmus <br> solution | With phenolphthalein <br> solution |
| :--- | :--- | :--- | :--- |
| A | No change | No change | No change |
| B | Turns red | No change | No change |
| C | No change | Turns blue | Turns pink |

On the basis of above observations, identify $\mathrm{A}, \mathrm{B}$ and C from the following options :

1. $\mathrm{A}=\mathrm{NH}_{4} \mathrm{Cl}, \mathrm{B}=\mathrm{NaCl}, \mathrm{C}=\mathrm{CH}_{3} \mathrm{COONa}$
2. $A=\mathrm{NH}_{4} \mathrm{Cl}, B=\mathrm{CH}_{3} \mathrm{COONa}, \mathrm{C}=\mathrm{NaCl}$
3. $\mathrm{A}=\mathrm{NaCl}, \mathrm{B}=\mathrm{NH}_{4} \mathrm{Cl}, \mathrm{C}=\mathrm{CH}_{3} \mathrm{COONa}$
4. $\mathrm{A}=\mathrm{CH}_{3} \mathrm{COONa}, \mathrm{B}=\mathrm{NH}_{4} \mathrm{Cl}, \mathrm{C}=\mathrm{NaCl}$

Sol. $\quad A=\mathrm{NaCl}-$ Neutral salt (No change)
$\mathrm{B}=\mathrm{NH}_{4} \mathrm{Cl}$ acidic in nature (Turn blue litmus to red)
$\mathrm{C}=\mathrm{CH}_{3} \mathrm{COONa}-$ Basic in nature (Turns Red litmus to blue)
Ans 3
85. Match List-I (Mixture to be separated) with the List-II (method used) and select the correct answer using the options given below :

| List-I (Mixture to be separated) | List-II (Methods Used) |
| :--- | :--- |
| A. Liquid $\mathrm{N}_{2}$ and liquid $\mathrm{O}_{2}$ | I. Chromatography |
| B. Red and Blue inks | II. Sublimation |
| C. Solution of NaCl in water | III. Fractional Distillation |
| D. Naphthalene and NaCl | IV. Evaporation <br> V. Crystallisation |

Sol. Liquid $\mathrm{N}_{2}$ and liquid $\mathrm{O}_{2}$ - Fractional Distillation
Red and Blue inks - Chromatography
Solution of NaCl in water - Evaporation
Naphthalene and NaCl - Sublimation
Ans 3
86. Select the correct set of statements regarding change in properties, as we move down the second group in periodic table.
I. Atomic size increases
II. Electronegativity increases.
III. Tendency to loose electrons increases
IV. Valency remains same

1. I, II and III
2. II, III and IV
3. I, II and IV
4. I, III and IV

Sol. As we know that as we move down in a group in periodic table
Atomic size increases,
Tendency to lose electron increases,
Valency remains same.
Ans 4
87. Which of the following options containing formula bonding and nature of aqueous solution respectively is correct for the compound formed by two elements $A$ and $B$ having atomic numbers 1 and 17 respectively?

1. $A B$, Ionic, Acidic
2. $A B_{2}$, Ionic, Basic
3. AB, Covalent, Acidic
4. $A B_{2}$ Covalent, Neutral

Sol.
Atomic number
Electronic Configuration Valency
Both $A$ and $B$ requires one electron to complete their outermost shell

$\therefore$ The formula of the compound is $A B$, Its nature is covalent and its aqueous solution is acidic in nature.
Ans 3
88. Choose one of the following alternative statements given below which correctly explain the process of osmosis.

1. Movement of water from regions of concentrated to dilute solution.
2. The passage of solute from weak solution to strong solution through a selectively-permeable membrane.
3.A passive transport of a solvent through a selectively - permeable membrane from a region of low solute concentration to a region of high solute concentration.
3. An energy - dependent transport of a solvent through a selectively - permeable membrane from region of low solute concentration to a region of high solute concentration.
4. In meiosis, each of the four daughter cells has one set of chromosomes. Due to randomness of process of chromosome separation in meiosis, large number of chromosome combinations can form gametes. How many such chromosome combinations in the gametes are possible in case of humans, assuming there is no crossing - over taking place ?
5. $2^{22}$
6. $2^{23}$
7. $2^{46}$
8. $2^{34}$
9. Sclerenchyma in plants is an example of simple permanent tissue comprising of two types of cells, sclereids and fibres. Why these cells are functionally important to the plants even after they die? Choose the correct alternative from the options given below.
10. Both are thin walled cells lacking intercellular spaces.
11. Walls in both the types of cells are thick and cutinized.
12. Walls in both the cell types are thick and usually lignified.
13. Both the cells are used for conducting solutes and providing strength to the plant.
14. Which one of the following organisms has a cellular respiratory pigment dissolved in plasma and is also a predaceous carnivore and shows matriphagy ?
1.Scorpion 2.Cockroach 3.Earthworm 4.Sea cucumber
15. Lichens are sensitive to certain air pollutants and are often replaced by other plants. From the given options choose the best combination of sensitivity and replacement of lichens.
16. Sulphur dioxide and moss
17. Sulphur dioxide and algae
18. Carbon dioxide and ferns
4.Sulphur dioxide and grass
19. A student was performing an experiment to understand the enzyme-substrate reaction. The student measured the formation of coloured product using a colorimeter. The student plotted the graph below which shows the reaction rate versus the substrate concentration.


Following interpretations were drawn by the student :
A. The higher concentration of substrate acts as an enzyme inhibitor.
B. It is a sigmoidal curve with sharp transition from low to high reaction rates over the increasing substrate concentration.
C. The curve reaches a plateau and does not further increase with increasing substrate concentrations due to saturation of enzyme with the substrate.
Choose which of the interpretations of the graph are correct.

1. A and B
2. A and C
3. B only
4. B and C
5. Glucose is the prime source of energy in our body. However, it is stored in the form of glycogen in the muscle and liver of animals and in the form of starch in plants. As a result, every time a cell requires glucose, it must hydrolyze glycogen which is an energy consuming process. Why does the cell store glycogen instead of glucose in free form?
6. Glycogen is more compact and more hydrophilic.
7. Storage of glucose in free form will consume more ATP.
8. Glucose in the free form creates more osmotic pressure.
9. Glucose is highly reactive molecule hence storing in the free form can result in unwanted reactions in the cells.
10. The figure given below is designed to show yeast respiration. In one of the tubes, there is yeast suspension in glucose solution. This solution was boiled before yeast was added to it. Which one of the following is the possible reason for boiling of sugar solution?

1.To ensure aerobic fermentation.
2.To provide the initial warmth for the yeast to become active.
11. To remove the dissolved oxygen and carbon dioxide form the solution.
12. To remove dissolved carbon dioxide and trap the oxygen from the atmosphere.
13. A squirrel was eating a fruit on the ground. Suddenly it was attacked by a dog. The squirrel rushed to the tree immediately and saved itself from the dangerous attack. What immediate changes are most likely to have taken place in the body of the squirrel ?
A. Blood flows to the stomach for rapid digestion.
B. Adrenaline was secreted in the blood by the adrenal glands.
C. Heart beat becomes faster and pumps more blood so that muscles get more oxygen
D. Adrenocorticotropic hormone is secreted in the blood and blood flows more towards the vital organs.
Select the correct combination of options given below.
14. $A$ and $B$
2.A and C
15. B and C
4.C and D 0
16. Stimulus from the environment is detected by the nerve cells. The stimulus acquired is transmitted in the form of electrical impulse. From the options given below choose the correct scheme showing the direction in which the nerve impulse travels.(Arrows shows the direction of impulse flow).

17. 


98. "Double fertilization" is a complex mechanism of flowering plants that is also unique to angiosperms. Choose the most appropriate statement from the options listed below that explains this phenomenon.

1. Fertilization in two flowers of the same plant forming endosperms.
2. Two male gametes fertilize two eggs inside the ovule as a result the ovary gives rise to bigger fruits.
3. Two fertilizations occur in a flower-one fertilization results in the formation of a diploid zygote and the second fertilization results in the formation of a triploid endosperm.
4. Two pollen grains sending two pollen tubes inside the ovary, resulting in the formation of two seeds inside the fruit.
5. It is generally observed that malaria is rampant in areas where construction work and/or stagnant water are usually seen. Plasmodium species are known to cause malaria. The parasite when injected by the mosquito into the human blood stream goes through specific life cycle stages. Select from below the correct sequence of stages.
1.Mosquito (sporozoites) $\rightarrow$ human liver (merozoites) $\rightarrow$ human RBC (Gametes) $\rightarrow$ mosquito (zygote-oocyst-sporozoites)
2.Mosquito (merozoites) $\rightarrow$ human RBC (Gametes) $\rightarrow$ human liver (sporozoites) $\rightarrow$ mosquito (oocyst - zygote -sporozoites)
3.Mosquito (merozoites) $\rightarrow$ human liver (sporozoites) $\rightarrow$ husman RBC (Gametes) $\rightarrow$ mosquito (oocyst - zygote -sporozoites)
4.Mosquito (sporozoites) $\rightarrow$ human liver (sporozoites) $\rightarrow$ human RBC (merozoites) $\rightarrow$ mosquito (zygote-oocyst-sporozoites)
6. A plant with red coloured flowers is crossed with a plant having white flowers. sThe red and white colour of the flower is controlled by a single gene. Red is dominant over white. The F1 progeny is self-pollinated and the flower colour in F2 is observed.
Given the above information. What is the expected phenotypic ratio of plants with different flower colours ?
7. All plants with red flowers.
8. Red : white in the ratio of $3: 1$.
9. Pink : white in the ratio of $3: 1$
4.Red : pink : white in a ratio of $1: 2: 1$.


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